

CHAPTER-3

Molybdenumhexacarbonyl mediated imino-carbonylative acylation of *NH*-sulfoximines with aryl iodides under catalyst-free conditions

3.1 Introduction

Sulfoximines are unique structural motifs received increasing interest in chemistry and biology over the past few decades (Figure 3.1) [1-13]. Structural diversity, high metabolic stability, interesting physicochemical properties, hydrogen-bonding capability, etc. makes sulfoximine as an important pharmacophore in drug discovery [1-4]. On the other hand, sulfoximine is also important functional group in organic chemistry and have been explored as building blocks, chiral ligands, organocatalysts, directing groups, etc., [5-11]. Among the different reactions of sulfoximines, *N*-functionalization reactions received considerable interest in chemistry and biology. For instance, *N*-arylation [14, 15], alkylation [16, 17], vinylation [18], alkynylation [19], cyanation [20], thiocyanation [21], trifluoromethylation [22], thiotrifluoromethylation [23] etc., [8] have been successfully demonstrated.

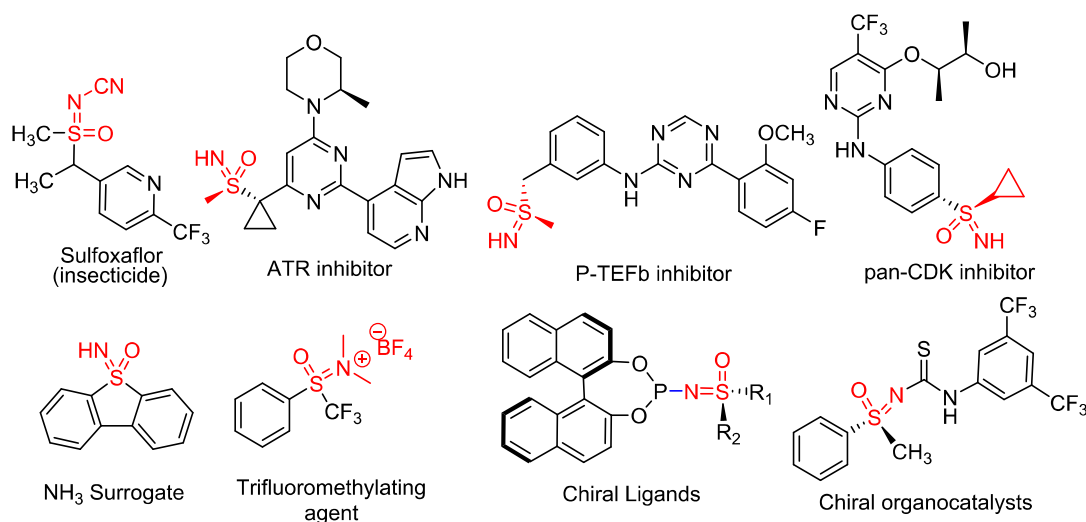
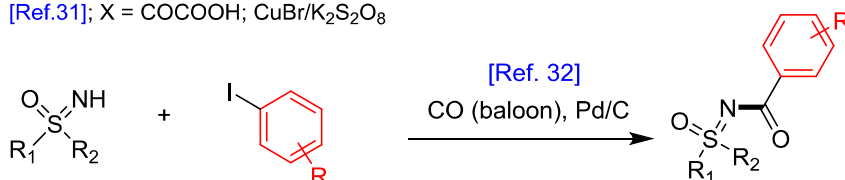
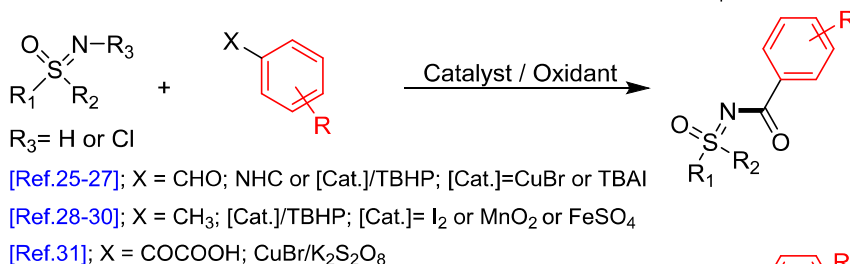
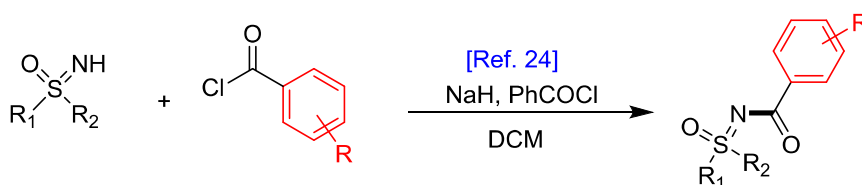


Figure 3.1: Structures of some biologically and chemically relevant sulfoximines.

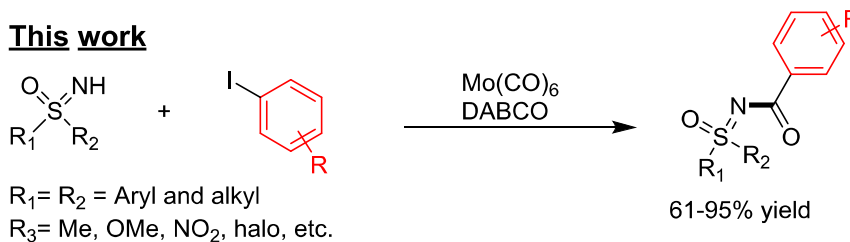
In this context, *N*-acylation of sulfoximines has received considerable interest in recent years [24-31]. The *N*-acylation of sulfoximines is typically achieved using acyl chlorides in the presence of a base [24]. On the other hand, as shown in scheme 3.1 oxidative coupling of sulfoximine with arylaldehyde [25-27], methyl arenes [28-30] or aryl glyoxylic acid [31]

also provides *N*-acylated sulfoximines (Scheme 3.1). Alternatively, palladium catalyzed gaseous carbon monoxide insertion between sulfoximines and aryl halides was also demonstrated [32]. Although few of these methods are efficient, there are some drawbacks including use of hazardous materials, inconvenient reaction conditions, longer reaction time, low yields, etc. enforce the development of alternative routes for the preparation of *N*-acyl sulfoximines.

Previous works



This work



Scheme 3.1: *N*-Acylation of sulfoximines using $Mo(CO)_6$.

Metal carbonyl compounds have been widely used as a source of solid carbon monoxide in organic synthesis [33, 34], Unlike gasses carbon monoxide, metal carbonyl compounds are stable, which can be conveniently handled and stored. In this context,

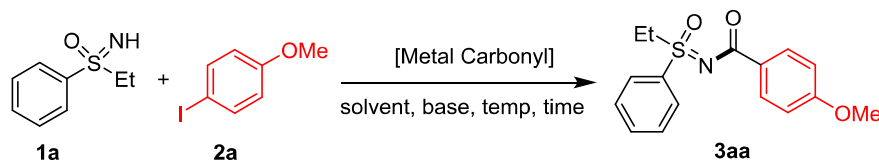
molybdenumhexacarbonyl mediated *N*-acylation of different amines was previously reported with aryl halides in the presence [35-40] as well as absence [41, 42] of palladium catalysts. To the best of our knowledge, *N*-acylation of sulfoximines or any other imines in the presence of metal carbonyl compounds was not explored. Recently, our group have demonstrated copper promoted *N*-alkylation [43] and *N*-arylation [44] of sulfoximines using aryl and alkyl boronic acids respectively under mild conditions. In continuation of previous chapter work on *N*-arylation of sulfoximines with aryl diazonium salts [45], here we report imino-carbonylative acylation of sulfoximines with aryl iodides in the presence of molybdenumhexacarbonyl under catalyst-free condition (Scheme 3.1).

3.2 Results and Discussion

At the outset, optimization of the reaction condition was performed using *S,S*-phenylethyl sulfoximine (**1a**) and 4-iodoanisole (**2a**) as model substrates in the presence of molybdenumhexacarbonyl (Table 3.1). The reaction was performed in different solvents including 1,4 dioxane, *m*-xylene, toluene, diglyme and tetrahydrofuran (THF) at 100 °C for 12 hours in the absence of any additives (Table 3.1, entries 1-5). Among them, 1,4-dioxane was found to be the best solvent providing the desired product in 53% yield while other solvents gave low yields. Further, the effect of temperature was investigated and observed that the reaction provides the desired product in 70% yield at 150 °C in 3 hours (Table 3.1, entries 6, 7). Moreover, there was no much difference in the yield with an extended time of reaction (Table 3.1, entry 8). The low yield of the product might be attributed to the less nucleophilicity of nitrogen in sulfoximines. Hence, the optimization was performed with different organic and inorganic bases (1.0 eq.) including triethylamine, *N,N*-diisopropylethylamine (*i*-Pr₂NEt), 1,8-diazabicyclo[5.4.0]undec-7-ene (DBU), 1,4-

diazabicyclo[2.2.2]octane (DABCO), caesium carbonate and potassium carbonate (Table 3.1, entries 9-14). Among them, DABCO was found to be a suitable combination with Mo(CO)₆, providing the desired product in 92% yield (Table 3.1, entry 12).

Table 3.1: Optimization of the reaction conditions for imino-carbonylation.^{a,b}

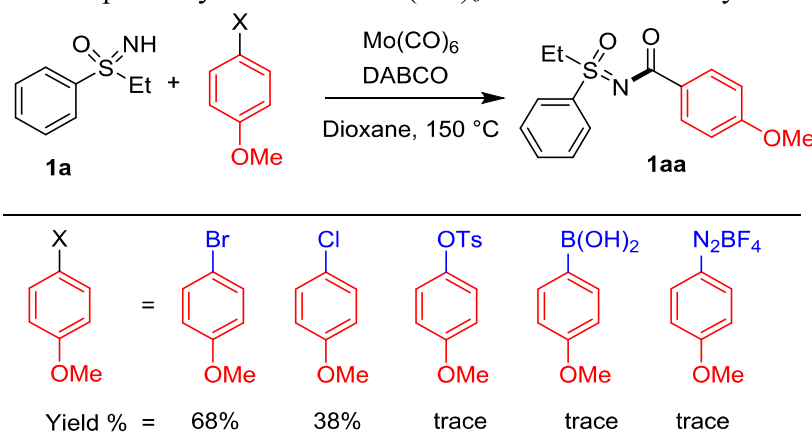


Entry	Metal carbonyl	Solvent	Base (1.0 eq.)	Catalyst (10 mol%)	Temp. (°C)	Time (h)	Yield ^b (%)
1	Mo(CO) ₆	1,4-Dioxane			100	12	53
2	Mo(CO) ₆	<i>m</i> -Xylene			100	12	32
3	Mo(CO) ₆	Toluene			100	12	35
4	Mo(CO) ₆	Diglyme			100	12	40
5	Mo(CO) ₆	THF			100	12	38
6	Mo(CO) ₆	1,4-Dioxane			120	12	65
7	Mo(CO) ₆	1,4-Dioxane			150	3	70
8	Mo(CO) ₆	1,4-Dioxane			150	6	73
9	Mo(CO) ₆	1,4-Dioxane	Et ₃ N		150	3	77
10	Mo(CO) ₆	1,4-Dioxane	<i>i</i> -Pr ₂ NEt		150	3	65
11	Mo(CO) ₆	1,4-Dioxane	DBU		150	3	75
12	Mo(CO)₆	1,4-Dioxane	DABC		150	3	92
13	Mo(CO) ₆	1,4-Dioxane	Cs ₂ CO ₃		150	3	54
14	Mo(CO) ₆	1,4-Dioxane	K ₂ CO ₃		150	3	58
15	Mo(CO) ₆	1,4-Dioxane	DABCO	Pd(OAc) ₂	100	3	65
16	Mo(CO) ₆	1,4-Dioxane	DABCO	Pd(OAc) ₂	120	3	80
17	Mo(CO) ₆	1,4-Dioxane	DABCO	Pd(OAc) ₂	150	3	93
18	Mo(CO) ₆	1,4-Dioxane		Pd(OAc) ₂	150	3	77
19	Co ₂ (CO) ₈	1,4-Dioxane	DABCO		150	3	30
20	Cr(CO) ₆	1,4-Dioxane	DABCO		150	3	10

^aS,S-phenylethyl sulfoximine (50 mg, 0.3 mmol), 4-iodoanisole (88 mg, 0.38 mmol), molybdenumhexacarbonyl (80 mg, 0.3 mmol) and solvent (2 mL). ^bIsolated yield.

Further, the reaction was performed in the presence of a catalytic amount of palladium acetate (10 mol%) at 100 °C, 120 °C and 150 °C. The intention of using catalyst i.e. Pd(OAc)₂ is that the reaction may provide higher yield of the desired product in lower temperature. However, there was only slight increment in the yield observed when compared with catalyst free condition (Table 3.1, entries 15-18 vs 6-8). Moreover, it is also clear that the base (i.e. DABCO) has some role in the reaction. For instance, the absence of DABCO, the reaction provided relatively low yields, i.e. with or without catalyst (Table 3.1, entries 8 and 18). Further, the carbonylation reaction was attempted in the presence of cobalt and chromium carbonyl compounds. These metal carbonyls gave the desired product in lower yields when compared with molybdenumhexacarbonyl (Table 3.1, entries 19 and 20).

Table 3.2: Scope of aryl source for Mo(CO)₆-mediated carbamoylation reaction.^{a,b}



^a*S,S*-phenylethyl sulfoximine (50 mg, 0.3 mmol), aryl source (0.38 mmol, 1.25eq) DABCO (34 mg, 0.3 mmol), Mo(CO)₆ (80 mg, 0.3 mmol) and dioxane (2 mL). ^bIsolated yield.

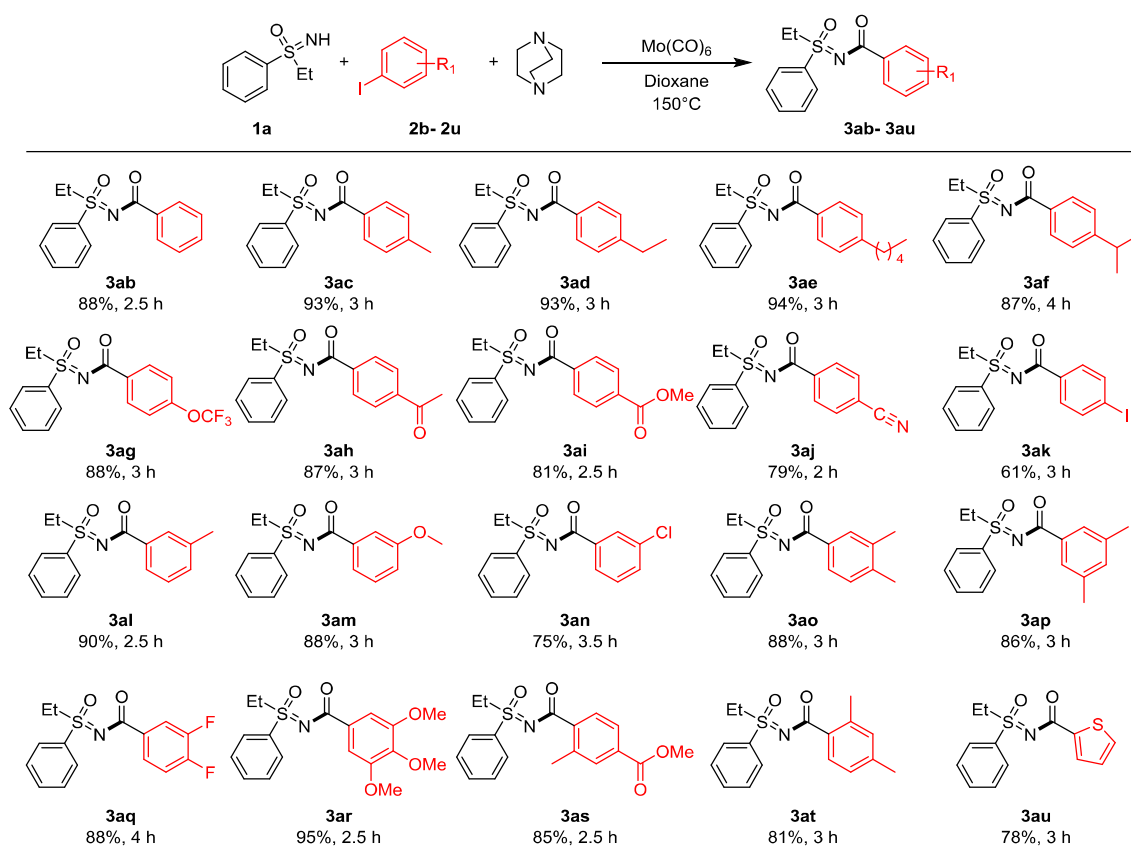
Encouraged by our initial findings, a different readily accessible aryl sources were examined for the coupling reactions instead of aryl iodide (Table 3.2). 4-Bromo and 4-chloroanisole gave the desired product in good to moderate yields while 4-methoxyphenyl tosylate, boronic acid and diazonium salt gave the desired product only in a trace amount (Scheme 3.2). Overall, the optimization study suggests that *N*-acyl sulfoximines can be

obtained efficiently from *NH*-sulfoximines in the presence of aryl iodides and molybdenum hexacarbonyl at 150 °C under catalyst-free conditions.

3.3 Substrates scope

Having established the optimized conditions, the scope of different aryl iodides in the carbonylation reaction was investigated with *S,S*-phenylethyl sulfoximine (**1a**). Aryl iodides bearing electron donating (e.g. Me, Et) and withdrawing groups (e.g. OCF₃, CN, COCH₃) at the *para*-position underwent carbonylative coupling with *S,S*-phenylethyl sulfoximine in 61–94% yields (Table 3.3, **3ab–3ak**).

Table 3.3: *N*-Acylation of sulfoximine with various aryl iodides.^{a,b}



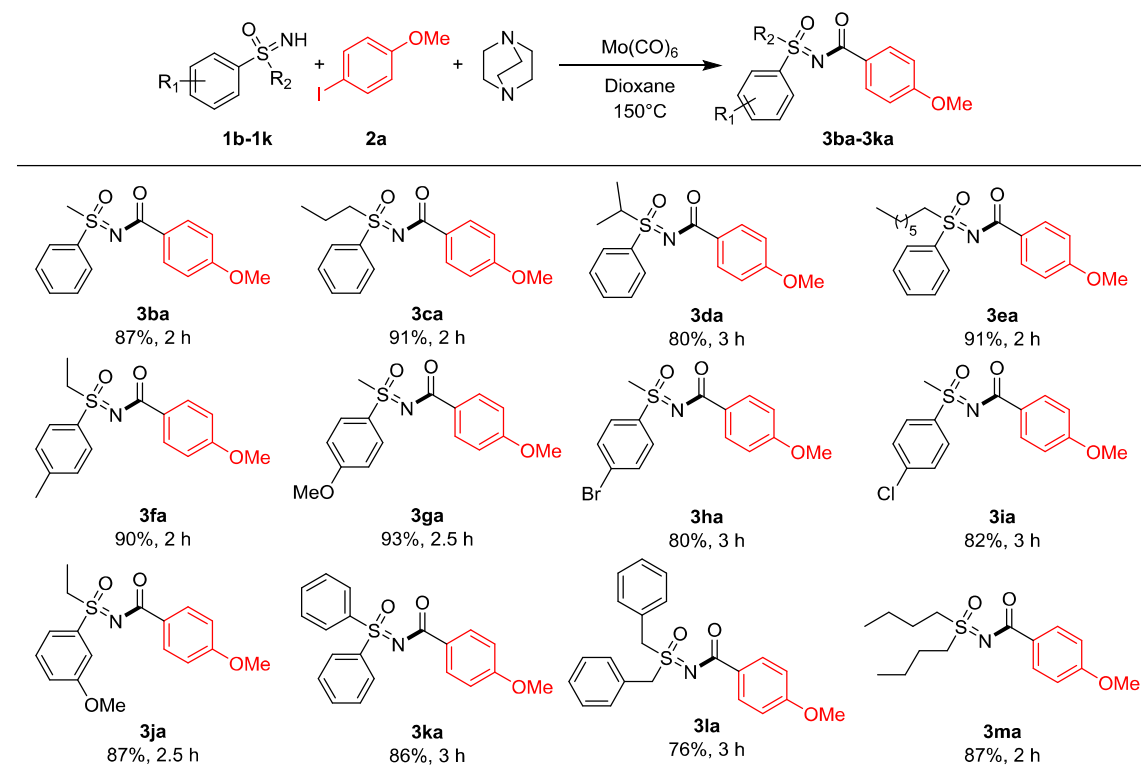
^a*S,S*-phenylethyl sulfoximine (170 mg, 1 mmol, 1eq.), aryl iodide (1.25 mmol), molybdenum hexacarbonyl (264 mg, 1 mmol), DABCO (112 mg, 1 mmol) and 1,4-dioxane (4 mL).
^bIsolated yield.

However, the electron withdrawing groups bearing aryl iodides required a slightly longer time for the coupling process while the desired products were obtained in good yields. Further, a different *meta*-functionalized aryl iodides were subjected to the coupling reactions with *S,S*-phenylethylsulfoximine (**1a**) under optimized conditions. To our delight, *m*-methyl, methoxy and fluoro-substituted iodobenzenes gave the desired *N*-acyl sulfoximines **3al-3ar** in good to excellent yields. It is also interesting to note that *meta*- as well as sterically hindered *ortho*-substituted aryl iodides were also participated in the coupling reaction very efficiently and provided the desired products **3as-3at** in excellent yields. Moreover, heterocyclic halide, *i.e.* 2-iodothiophene underwent carbonylative coupling reaction with sulfoximine **1a** smoothly and provided the desired product **3au** in 78% yield.

Encouraged, we further examined the scope of sulfoximines (**1b-1k**) in the carbonylative coupling reactions in the presence of 4-iodoanisole and molybdenumhexacarbonyl under optimized conditions (Table 3.4). Phenylsulfoximines bearing linear and branched alkyl chains underwent *N*-acylation in excellent yields (Table 3.4, **3ba-3ea**). It is worth noting that sterically hindered *S,S*-*iso*-propylphenylsulfoximine also provided the desired products in 80% yield (Table 3.3, **3da**). Further, *N*-acylation of sulfoximines bearing different substitutions on the aryl ring (*i.e.* methyl, methoxy, bromo and chloro groups) was investigated. These sulfoximines underwent coupling reaction with 4-iodoanisole and provided the desired *N*-acyl sulfoximines in 80-93% yields (Table 3.3, **3fa-3ja**). Moreover, diphenyl sulfoximine, dibenzyl sulfoximine and dibutyl sulfoximine also participated in the coupling reaction very efficiently and gave the corresponding *N*-acyl sulfoximines in 76-87% yields (Table 3.3, **3ka-3ma**). Overall, it was observed that the substituents on the sulfur as well as in the aryl ring played only a minor role in terms of yield.

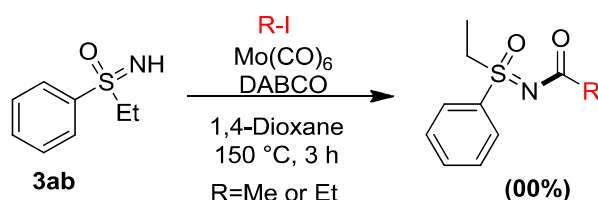
Moreover, functional groups such as ester, ether, ketone, nitrile as well as halides were well tolerated under the reaction conditions.

Table 3.4: *N*-Acylation of various sulfoximine with 4-iodoanisole.^{a,b}



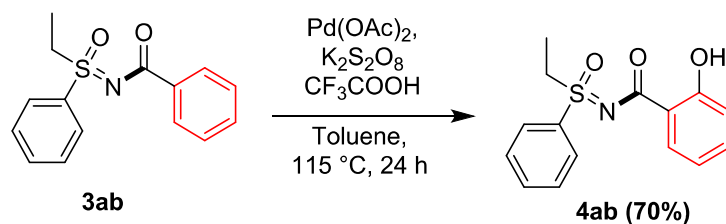
^aSulfoximine (1 mmol, 1eq), 4-iodoanisole (292 mg, 1.25 mmol), molybdenumhexacarbonyl (264 mg, 1 mmol), DABCO (112 mg, 1 mmol) and 1,4-dioxane (3mL). ^bIsolated Yield.

Having studied the scope of different aryl iodides, we have investigated suitability of alkyl iodides (e.g. MeI and EtI) for the coupling reaction under optimized condition. However, unfortunately the reaction failed to yield the desired product while starting material **3ab** was recovered quantitatively (Scheme 3.2).



Scheme 3.2: *N*-Acylation of various sulfoximine with alkyl iodides.

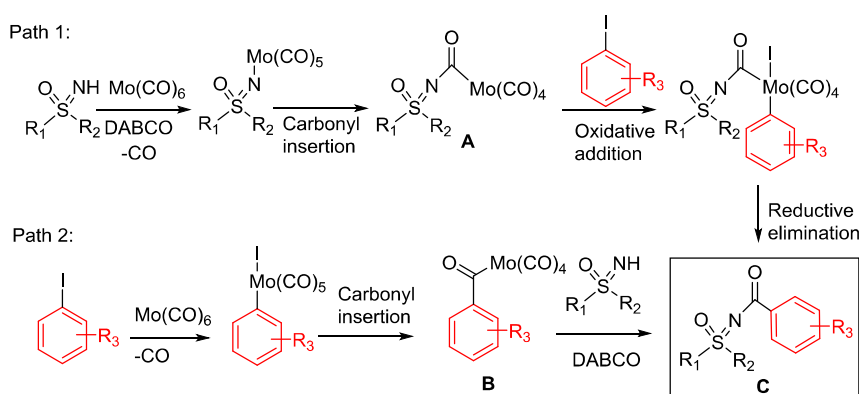
N-Acyl sulfoximines have found wide synthetic applications in organic chemistry [7, 24, 46-51]. Here we have demonstrated one of the established transformations, i.e. *ortho*-hydroxylation of *N*-acyl ring was successfully achieved in the presence of palladium acetate and potassium persulfate in 70% yield (Scheme 3.3) [25].



Scheme 3.3: *Ortho*-hydroxylation of *N*-benzoyl sulfoximine.

3.4 Plausible Reaction Mechanism

Based on the literature reports [41, 42], a plausible mechanism for the imino-carbonylation is shown in Scheme 3.4. Imine or aryl iodide undergoes substitution followed by carbonyl insertion to provide the intermediates **A** or **B**, respectively (Path 1 and 2). The intermediate **A** undergoes oxidative addition with aryl iodide followed by reductive elimination to provide the desired product **C**. Alternatively the intermediate **B** undergoes nucleophilic displacement with sulfoximine in the presence of DABCO to provide the desired product **C**.



Scheme 3.4: Plausible mechanism for carbonylative acylation of sulfoximine.

3.5 Conclusion

We have demonstrated a convenient and efficient method for the direct *N*-acylation of *NH*-sulfoximines with aryl iodides using Mo(CO)₆ under catalyst-free conditions. This protocol affords different *N*-acyl sulfoximines in good to excellent yields while several functional groups were well tolerated. This straight forward approach might serve as a practical alternative to the existing methods for the preparation of a wide range of *N*-acyl sulfoximines.

3.6 Experimental Section

3.6.1 Experimental procedure for the optimization table

S,S-Methylphenylsulfoximine (50 mg, 0.32 mmol), iodoanisole (88 mg, 0.38 mmol), molybdenumhexacarbonyl [Mo(CO)₆] (80 mg, 0.3 mmol) and solvent (2 mL) were taken in a teflon screw capped pressure tube [with or without base and catalyst] and stirred at appropriate temperature. The progress of the reaction was monitored by thin-layer chromatography. After completion of the reaction (or appropriate time), the reaction mixture was diluted with ethyl acetate and washed with water and brine solution. The organic layer was dried over anhydrous sodium sulfate and evaporated to dryness. The crude product was purified on silica-gel (100-200 mesh) column chromatography using a mixture ethyl acetate/hexane as eluent to obtain **3aa**. R_f (40 % EA/hexane) 0.54.

3.6.2 Optimized experimental procedure for the acylation of sulfoximines

Sulfoximine (1 mmol), Mo(CO)₆ (264 mg, 1 mmol), aryl iodide (1.25 mmol), DABCO (112 mg, 1 mmol) and 1,4-dioxane (4 mL) were added into 15 mL teflon screw capped pressure tube and refluxed at 150 °C on oil bath with magnetic stirrer for appropriate time. The progress of the reaction was monitored by thin-layer chromatography using 40% ethyl

acetate (EA) and hexane (hex) as eluent. After completion, the reaction mixture was diluted with ethyl acetate and washed with water and brine solution. The organic layer was dried over anhydrous sodium sulfate and evaporated to dryness. The crude product was further purified on silica-gel (100–200 mesh) column chromatography using a mixture of ethyl acetate/hexane as eluent.

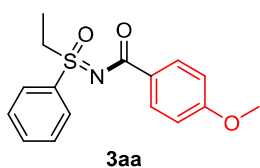
3.6.3 Experimental procedure for the *ortho*-hydroxylation of acylated sulfoximine [25]

N-Benzoyl-*S,S*-methylphenylsulfoximine **3aa** (0.55 mmol, 150 mg), K₂S₂O₈ (1.1 mmol, 270 mg), Pd(OAc)₂ (0.05 mmol, 11.2 mg), toluene (2 mL) and CF₃COOH (1 mmol, 0.076 mL) were taken into 25 ml round bottom flask and refluxed at 115 °C for 24h. The reaction mixture was diluted with ethyl acetate and washed with water and brine solution. The organic layer was dried over anhydrous sodium sulfate and evaporated to dryness. The crude product was further purified on silica-gel (100–200 mesh) column chromatography using a mixture of ethyl acetate/hexane as eluent.

3.7 Analytical data for *N*-acyl sulfoximines

3.7.1

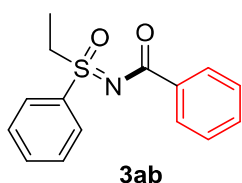
N-(4-methoxybenzoyl)-*S*-ethyl-*S*-phenyl sulfoximine (**3aa**):



The reaction was carried out using 50 mg of sulfoximine according to general procedure. The title compound was obtained as white solid, 83 mg, 92% yield, m.p. 135–136 °C. The residue was isolated by silica-gel column chromatography using 20% ethyl acetate: hexane as an eluent, R_f 0.54 (40% EA: hex). ¹H NMR (500 MHz, CDCl₃) δ 8.16–8.11 (m, 2H), 7.99 (d, *J* = 8.1 Hz, 2H), 7.69–7.64 (m, 1H), 7.59 (t, *J* = 7.7 Hz, 2H), 6.91–6.88 (m, 2H), 3.85 (s, 3H), 3.59

(q, $J = 7.4$ Hz, 2H), 1.29 (t, $J = 7.4$ Hz, 3H); ^{13}C NMR (125 MHz, CDCl_3) δ 173.8, 162.9, 136.8, 133.7, 131.5, 129.6, 128.5, 128.1, 113.3, 55.5, 50.7, 7.4. HRMS (ESI-TOF) calcd. for $\text{C}_{16}\text{H}_{18}\text{NO}_3\text{S}$ $[\text{M}+\text{H}]^+$ 304.1007; found 304.0998.

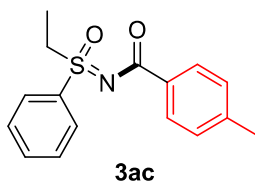
3.7.2



N-benzoyl-*S*-ethyl-*S*-phenyl sulfoximine (**3ab**):

The reaction was carried out using 169 mg of sulfoximine according to general procedure. The title compound was obtained as white solid, 238 mg, 88% yield, m.p. 132–135 °C. The residue was isolated by silica-gel column chromatography using 20% ethyl acetate:hexane as an eluent. R_f 0.56 (40% EA: hex). ^1H NMR (500 MHz, CDCl_3) δ 8.20–8.17 (m, 2H), 8.0 (dt, $J = 3.3, 1.9$ Hz, 2H), 7.67 (ddd, $J = 6.7, 3.9, 1.2$ Hz, 1H), 7.62–7.59 (m, 2H), 7.52–7.49 (m, 1H), 7.43–7.40 (m, 2H), 3.61 (q, $J = 7.4$ Hz, 2H), 1.31 (t, $J = 7.4$ Hz, 3H); ^{13}C NMR (125 MHz, CDCl_3) δ 174.2, 136.7, 135.8, 133.8, 132.2, 129.6 (d, $J = 16.9$ Hz), 128.1, 50.7, 7.3. HRMS (ESI-TOF) calcd. for $\text{C}_{15}\text{H}_{16}\text{NO}_2\text{S}$ $[\text{M}+\text{H}]^+$ 274.0902; found 274.0887.

3.7.3



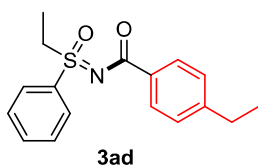
N-(4-Methylbenzoyl)-*S*-ethyl-*S*-phenyl sulfoximine (**3ac**):

The reaction was carried out using 169 mg of sulfoximine according to general procedure. The title compound was obtained as white solid, 266 mg, 93% yield, m.p. 123–126 °C. The residue was isolated by silica-gel column chromatography using 20% ethyl acetate:hexane as an eluent, R_f 0.6 (40% EA: hex). ^1H NMR (500 MHz, CDCl_3) δ 8.07 (d, $J = 8.2$ Hz, 2H), 8.00 (dd, $J = 8.3, 1.0$ Hz,

2H), 7.66 (d, $J = 7.5$ Hz, 1H), 7.60 (t, $J = 7.6$ Hz, 2H), 7.21 (d, $J = 8.0$ Hz, 2H), 3.61 (q, $J = 7.4$ Hz, 2H), 2.40 (s, 3H), 1.30 (t, $J = 7.4$ Hz, 3H); ^{13}C NMR (125 MHz, CDCl_3) δ 174.3, 142.7, 136.8, 133.8, 133.2, 129.68 (d, $J = 2.0$ Hz), 128.8, 128.1, 50.7, 21.7, 7.4. HRMS (ESI-TOF) calcd. for $\text{C}_{16}\text{H}_{18}\text{NO}_2\text{S}$ $[\text{M}+\text{H}]^+$ 288.1058; found 288.1049.

3.7.4

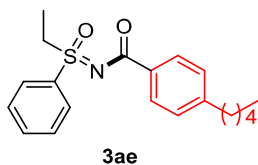
N-(4-ethylbenzoyl)-*S*-ethyl-*S*-phenyl sulfoximine (3ad):



The reaction was carried out using 169 mg of sulfoximine according to general procedure. The title compound was obtained as white solid, 280 mg, 93% yield, m.p. 85–86 °C. The residue was isolated by silica-gel column chromatography using 18% ethyl acetate:hexane as an eluent, R_f 0.6 (40% EA: hex). ^1H NMR (500 MHz, CDCl_3) δ 8.11–8.09 (m, 2H), 8.0–7.98 (m, 2H), 7.68–7.65 (m, 1H), 7.58 (dd, $J = 10.5, 4.7$ Hz, 2H), 7.24 (d, $J = 8.3$ Hz, 2H), 3.60 (q, $J = 7.4$ Hz, 2H), 2.69 (q, $J = 7.6$ Hz, 2H), 1.29 (t, $J = 7.4$ Hz, 3H), 1.25 (t, $J = 7.6$ Hz, 3H); ^{13}C NMR (125 MHz, CDCl_3) δ 174.2, 148.9, 136.7, 133.7, 133.3, 129.68 (d, $J = 7.1$ Hz), 128.1, 127.6, 50.7, 29.0, 15.4, 7.3. HRMS (ESI-TOF) calcd. for $\text{C}_{17}\text{H}_{20}\text{NO}_2\text{S}$ $[\text{M}+\text{H}]^+$ 302.1215; found 302.1204.

3.7.5

N-(4-pentylbenzoyl)-*S*-ethyl-*S*-phenyl sulfoximine (3ae):

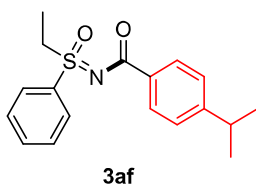


The reaction was carried out using 169 mg of sulfoximine according to general procedure. The title compound was obtained as white solid, 320 mg, 94% yield, m.p. 102–103 °C. The residue was

isolated by silica-gel column chromatography using 16% ethyl acetate:hexane as an eluent, R_f 0.62 (40% EA: hex). ^1H NMR (500 MHz, CDCl_3) δ 8.10–8.08 (m, 2H), 8.0–7.98 (m, 2H), 7.67–7.63 (m, 1H), 7.59–7.56 (m, 2H), 7.21 (d, $J = 8.2$ Hz, 2H), 3.59 (q, $J = 7.4$ Hz, 2H), 2.66–2.63 (m, 2H), 1.65–1.59 (m, 2H), 1.31 (ddd, $J = 14.8, 9.3, 5.0$ Hz, 7H), 0.88 (t, $J = 6.9$ Hz, 3H); ^{13}C NMR (125 MHz, CDCl_3) δ 174.2, 147.6, 136.6, 133.7, 133.2, 129.58 (d, $J = 5.3$ Hz), 128.09 (d, $J = 13.7$ Hz), 50.6, 35.9, 31.4, 30.9, 22.5, 14.0, 7.3. HRMS (ESI-TOF) calcd. for $\text{C}_{20}\text{H}_{26}\text{NO}_2\text{S}$ $[\text{M}+\text{H}]^+$ 344.1684; found 344.1671.

3.7.6

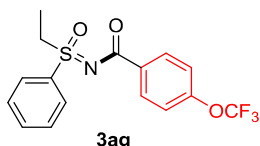
N-(4-isopropylbenzoyl)-*S*-ethyl-*S*-phenyl sulfoximine (3af):



The reaction was carried out using 169 mg of sulfoximine according to general procedure. The title compound was obtained as white solid, 274 mg, 87% yield, m.p. 64–65 °C. The residue was isolated by silica-gel column chromatography using 20% ethyl acetate:hexane as an eluent, R_f 0.54 (40% EA: hex). ^1H NMR (500 MHz, CDCl_3) δ 8.12–8.10 (m, 2H), 8.0–7.99 (m, 2H), 7.68–7.65 (m, 1H), 7.59 (dd, $J = 8.1, 7.2$ Hz, 2H), 7.28 (s, 1H), 7.26 (d, $J = 1.5$ Hz, 1H), 3.60 (q, $J = 7.4$ Hz, 2H), 2.96 (dt, $J = 13.8, 6.9$ Hz, 1H), 1.29 (d, $J = 7.4$ Hz, 3H), 1.27 (d, $J = 1.6$ Hz, 3H), 1.26 (d, $J = 1.5$ Hz, 3H); ^{13}C NMR (125 MHz, CDCl_3) δ 174.2, 153.5, 136.7, 133.7, 133.5, 129.69 (d, $J = 8.7$ Hz), 128.1, 126.2, 50.7, 34.3, 23.9, 7.3. HRMS (ESI-TOF) calcd. for $\text{C}_{18}\text{H}_{22}\text{NO}_2\text{S}$ $[\text{M}+\text{H}]^+$ 316.1371; found

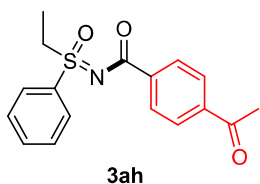
316.1362.

3.7.7 *N*-(4-trifluoromethoxybenzoyl)-*S*-ethyl-*S*-phenyl sulfoximine (**3ag**):



The reaction was carried out using 169 mg of sulfoximine according to general procedure. The title compound was obtained as white solid, 314 mg, 88% yield, m.p. 74–75 °C. The residue was isolated by silica-gel column chromatography using 20% ethyl acetate:hexane as an eluent, R_f 0.5 (40% EA: hex). ^1H NMR (500 MHz, CDCl_3) δ 8.23–8.20 (m, 2H), 7.98 (dd, $J = 5.3, 4.1$ Hz, 2H), 7.71–7.68 (m, 1H), 7.63–7.60 (m, 2H), 7.24 (dd, $J = 9.4, 8.6$ Hz, 2H), 3.64–3.56 (m, 2H), 1.32 (t, $J = 7.4$ Hz, 3H); ^{13}C NMR (125 MHz, CDCl_3) δ 172.9, 152.1, 136.4, 134.2, 134.0, 131.4, 129.8, 128.0, 120.0, 50.8, 7.3; ^{19}F NMR (500 MHz, CDCl_3) δ -57.5. HRMS (ESI-TOF) calcd. for $\text{C}_{16}\text{H}_{15}\text{F}_3\text{NO}_3\text{S}$ $[\text{M}+\text{H}]^+$ 358.0725; found 358.0711.

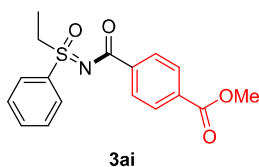
3.7.8 *N*-(4-acetylbenzoyl)-*S*-ethyl-*S*-phenyl sulfoximine (**3ah**):



The reaction was carried out using 169 mg of sulfoximine according to general procedure. The title compound was obtained as white solid, 274 mg, 87% yield, m.p. 132–133 °C. The residue was isolated by silica-gel column chromatography using 18% ethyl acetate:hexane as an eluent, R_f 0.56 (40% EA: hex). ^1H NMR (500 MHz, CDCl_3) δ 8.24–8.23 (m, 2H), 8.00–7.95 (m, 4H), 7.70–7.67 (m, 1H), 7.61 (t, $J = 7.7$ Hz, 2H), 3.61 (qq, $J = 14.4, 7.4$ Hz, 2H),

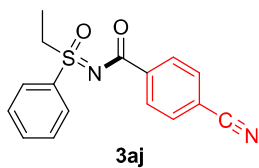
2.62 (s, 3H), 1.31 (t, $J = 7.4$ Hz, 3H); ^{13}C NMR (125 MHz, CDCl_3) δ 198.0, 173.2, 139.68 (d, $J = 8.4$ Hz), 136.3, 134.0, 133.1, 129.75 (d, $J = 8.8$ Hz), 129.2, 128.6, 128.07 (d, $J = 4.6$ Hz), 50.7, 27.0, 7.2. HRMS (ESI-TOF) calcd. for $\text{C}_{17}\text{H}_{18}\text{NO}_3\text{S}$ $[\text{M}+\text{H}]^+$ 316.1007; found 316.0995.

3.7.9 *N*-(methyl 4-benzoatebenzoyl)-*S*-ethyl-*S*-phenyl sulfoximine (3ai):



The reaction was carried out using 169 mg of sulfoximine according to general procedure. The title compound was obtained as white solid, 268 mg, 81% yield, m.p. 134–135 °C. The residue was isolated by silica-gel column chromatography using 18% ethyl acetate:hexane as an eluent, R_f 0.56 (40% EA: hex). ^1H NMR (500 MHz, CDCl_3) δ 8.22 (d, $J = 8.5$ Hz, 2H), 8.07 (d, $J = 8.4$ Hz, 2H), 8.00–7.98 (m, 2H), 7.68 (d, $J = 7.4$ Hz, 1H), 7.61 (t, $J = 7.7$ Hz, 2H), 3.93 (s, 3H), 3.62 (dt, $J = 14.7, 7.1$ Hz, 2H), 1.32 (t, $J = 7.4$ Hz, 3H); ^{13}C NMR (125 MHz, CDCl_3) δ 173.3, 166.8, 139.6, 136.3, 134.0, 133.1, 129.8, 129.44 (d, $J = 8.8$ Hz), 128.0, 52.4, 50.7, 7.3. HRMS (ESI-TOF) calcd. for $\text{C}_{17}\text{H}_{18}\text{NO}_4\text{S}$ $[\text{M}+\text{H}]^+$ 332.0957; found 332.0946.

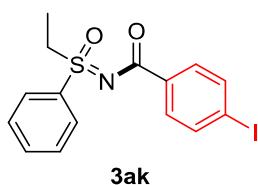
3.7.10 *N*-(4-cyanobenzoyl)-*S*-ethyl-*S*-phenyl sulfoximine (3aj):



The reaction was carried out using 169 mg of sulfoximine according to general procedure. The title compound was obtained as white solid, 235 mg, 79% yield, m.p. 86–88 °C. The residue was isolated

by silica-gel column chromatography using 20% ethyl acetate: hexane as an eluent, R_f 0.53 (40% EA: hex). ^1H NMR (500 MHz, CDCl_3) δ 8.25 (d, $J = 8.2$ Hz, 2H), 7.99–7.97 (m, 2H), 7.72 (dd, $J = 10.7, 4.5$ Hz, 3H), 7.63 (t, $J = 7.7$ Hz, 2H), 3.68–3.55 (m, 2H), 1.33 (t, $J = 7.4$ Hz, 3H); ^{13}C NMR (125 MHz, CDCl_3) δ 172.4, 139.6, 136.0, 134.2, 132.0, 129.95 (d, $J = 15.1$ Hz), 128.0, 118.6, 115.4, 50.8, 7.2. HRMS (ESI-TOF) calcd. for $\text{C}_{16}\text{H}_{15}\text{N}_2\text{O}_2\text{S}$ $[\text{M}+\text{H}]^+$ 299.0854; found 299.0845.

3.7.11

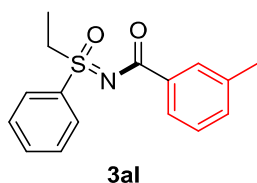


N-(4-iodobenzoyl)-*S*-ethyl-*S*-phenyl sulfoximine (3ak):

The reaction was carried out using 169 mg of sulfoximine according to general procedure. The title compound was obtained as white solid, 244 mg, 61% yield, m.p. 142–143 °C. The residue was isolated by silica-gel column chromatography using 20% ethyl acetate:hexane as an eluent, R_f 0.52 (40% EA: hex). ^1H NMR (500 MHz, CDCl_3) δ 7.98 (ddd, $J = 8.6, 3.3, 1.7$ Hz, 2H), 7.89–7.86 (m, 2H), 7.77–7.74 (m, 2H), 7.69–7.66 (m, 1H), 7.62–7.58 (m, 2H), 3.62–3.58 (m, 2H), 1.31–1.28 (m, 3H); ^{13}C NMR (125 MHz, CDCl_3) δ 173.5, 137.3, 136.4, 135.3, 133.9, 131.1, 129.7, 128.0, 99.8, 50.7, 7.30. HRMS (ESI-TOF) calcd. for $\text{C}_{15}\text{H}_{15}\text{INO}_2\text{S}$ $[\text{M}+\text{H}]^+$ 399.9868; found 399.9855.

3.7.12

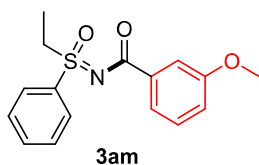
N-(3-methylbenzoyl)-*S*-ethyl-*S*-phenyl sulfoximine (3al):



The reaction was carried out using 169 mg of sulfoximine according to general procedure. The title compound was obtained as white solid, 258 mg, 90% yield, m.p. 88-90 °C. The residue was isolated by silica-gel column chromatography using 20% ethyl acetate:hexane as an eluent, R_f 0.54 (40% EA: hex). ^1H NMR (500 MHz, CDCl_3) δ 8.01 (s, 1H), 8.00–7.98 (m, 3H), 7.69–7.66 (m, 1H), 7.60 (dd, $J = 10.5, 4.8$ Hz, 2H), 7.33–7.29 (m, 2H), 3.61 (q, $J = 7.4$ Hz, 2H), 2.40 (s, 3H), 1.31 (t, $J = 7.4$ Hz, 3H); ^{13}C NMR (125 MHz, CDCl_3) δ 174.5, 137.8, 136.7, 135.7, 133.8, 133.0, 130.1, 129.7, 128.10 (d, $J = 10.0$ Hz), 126.7, 50.7, 21.4, 7.3. HRMS (ESI-TOF) calcd. for $\text{C}_{16}\text{H}_{18}\text{NO}_2\text{S}$ $[\text{M}+\text{H}]^+$ 288.1058; found 288.1052.

3.7.13

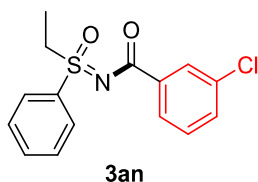
N-(3-methoxybenzoyl)-*S*-ethyl-*S*-phenyl sulfoximine (**3am**):



The reaction was carried out using 169 mg of sulfoximine according to general procedure. The title compound was obtained as white solid, 266 mg, 88% yield, m.p. 102–103 °C. The residue was isolated by silica-gel column chromatography using 20% ethyl acetate:hexane as an eluent, R_f 0.55 (40% EA: hex). ^1H NMR (500 MHz, CDCl_3) δ 8.00 (d, $J = 7.4$ Hz, 2H), 7.80 (d, $J = 7.6$ Hz, 1H), 7.70–7.66 (m, 2H), 7.60 (t, $J = 7.8$ Hz, 2H), 7.32 (t, $J = 7.9$ Hz, 1H), 7.06 (dd, $J = 8.1, 2.6$ Hz, 1H), 3.84 (s, 3H), 3.60 (q, $J = 7.4$ Hz, 2H), 1.31 (t, $J = 7.4$ Hz, 3H); ^{13}C NMR (125 MHz, CDCl_3) δ 174.1, 159.5, 137.3, 136.7, 133.8, 129.7, 129.1, 128.1, 122.1, 118.7, 113.9, 55.5, 50.7, 7.3. HRMS (ESI-TOF) calcd. for $\text{C}_{16}\text{H}_{18}\text{NO}_3\text{S}$ $[\text{M}+\text{H}]^+$

304.1007; found 304.1002.

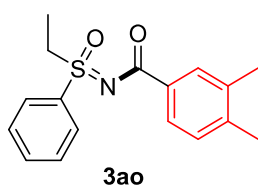
3.7.14



N-(3-chlorobenzoyl)-*S*-ethyl-*S*-phenyl sulfoximine (**3an**):

The reaction was carried out using 169 mg of sulfoximine according to general procedure. The title compound was obtained as white solid, 230 mg, 75% yield, m.p. 103–105 °C. The residue was isolated by silica-gel column chromatography using 20% ethyl acetate:hexane as an eluent. R_f 0.53 (40% EA: hex). ^1H NMR (500 MHz, CDCl_3) δ 8.15 (t, $J = 1.7$ Hz, 1H), 8.03 (dd, $J = 7.7, 1.1$ Hz, 1H), 8.00–7.96 (m, 2H), 7.70 (dd, $J = 10.7, 4.1$ Hz, 1H), 7.62 (t, $J = 7.6$ Hz, 2H), 7.47 (ddd, $J = 7.9, 2.0, 1.0$ Hz, 1H), 7.35 (t, $J = 7.8$ Hz, 1H), 3.66–3.52 (m, 2H), 1.32 (t, $J = 7.4$ Hz, 3H); ^{13}C NMR (125 MHz, CDCl_3) δ 172.9, 137.6, 136.3, 134.2, 134.0, 132.1, 129.75 (d, $J = 14.2$ Hz), 129.4, 128.0, 127.6, 50.8, 7.3. HRMS (ESI-TOF) calcd. for $\text{C}_{15}\text{H}_{15}\text{ClNO}_2\text{S}$ $[\text{M}+\text{H}]^+$ 308.0512; found 308.0504.

3.7.15

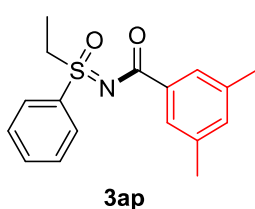


N-(3,4-dimethylbenzoyl)-*S*-ethyl-*S*-phenyl sulfoximine (**3ao**):

The reaction was carried out using 169 mg of sulfoximine according to general procedure. The title compound was obtained as white solid, 264 mg, 88% yield, m.p. 117–118 °C. The residue was isolated by silica-gel column chromatography using 18% ethyl acetate:hexane as an eluent, R_f 0.56 (40% EA: hex). ^1H NMR (500 MHz, CDCl_3) δ 7.99 (d, $J = 7.9$ Hz, 2H), 7.94–7.91 (m, 2H), 7.66 (t, $J = 7.4$ Hz, 1H), 7.59 (t, $J = 7.7$ Hz, 2H), 7.17 (d, $J = 7.8$ Hz, 1H), 3.60 (q, $J = 7.4$ Hz, 2H), 2.31 (t, $J = 5.3$ Hz, 6H), 1.30 (t, $J = 7.4$ Hz,

3H); ^{13}C NMR (125 MHz, CDCl_3) δ 174.5, 141.4, 136.7, 136.3, 133.7, 133.4, 130.6, 129.6, 129.4, 128.1, 127.2, 50.7, 20.0, 19.8, 7.3. HRMS (ESI-TOF) calcd. for $\text{C}_{17}\text{H}_{20}\text{NO}_2\text{S}$ $[\text{M}+\text{H}]^+$ 302.1215; found 302.1203.

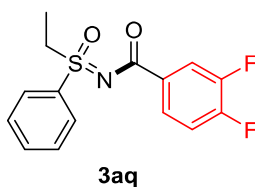
3.7.16



N-(3,5-dimethylbenzoyl)-*S*-ethyl-*S*-phenyl sulfoximine (3ap):

The reaction was carried out using 169 mg of sulfoximine according to general procedure. The title compound was obtained as white solid, 256 mg, 86% yield, m.p. 142–143 °C. The residue was isolated by silica-gel column chromatography using 18% ethyl acetate:hexane as an eluent, R_f 0.56 (40% EA: hex). ^1H NMR (500 MHz, CDCl_3) δ 8.02–7.96 (m, 2H), 7.79 (s, 2H), 7.69–7.63 (m, 1H), 7.62–7.56 (m, 2H), 7.14 (s, 1H), 3.60 (q, $J = 7.4$ Hz, 2H), 2.35 (s, 6H), 1.30 (t, $J = 7.4$ Hz, 3H); ^{13}C NMR (125 MHz, CDCl_3) δ 174.6, 137.6, 136.6, 135.6, 133.81 (d, $J = 13.4$ Hz), 129.6, 128.0, 127.2, 50.6, 21.3, 7.2. HRMS (ESI-TOF) calcd. for $\text{C}_{17}\text{H}_{20}\text{NO}_2\text{S}$ $[\text{M}+\text{H}]^+$ 302.1215; found 302.1203.

3.7.17

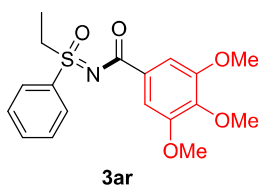


N-(3,4-Difluorobenzoyl)-*S*-ethyl-*S*-phenyl sulfoximine (3aq):

The reaction was carried out using 169 mg of sulfoximine according to general procedure. The title compound was obtained as white solid, 269 mg, 88% yield, m.p. 88–87 °C. The residue was isolated by silica-gel column chromatography using 18% ethyl acetate:hexane as an eluent, R_f 0.52 (40% EA: hex). ^1H NMR (500 MHz, CDCl_3) δ 8.03–7.99 (m, 3H), 7.97–7.94 (m, 1H), 7.74–7.70

(m, 1H), 7.64 (dd, $J = 10.6, 4.7$ Hz, 2H), 7.22–7.17 (m, 1H), 3.68–3.55 (m, 2H), 1.34 (t, $J = 7.4$ Hz, 3H); ^{13}C NMR (125 MHz, CDCl_3) δ 171.9, 154.1, 152.10 (d, $J = 13.1$ Hz), 150.85 (d, $J = 13.0$ Hz), 148.88 (d, $J = 13.2$ Hz), 136.1, 133.9, 132.8, 129.7, 127.9, 126.15 (dd, $J = 7.3, 3.5$ Hz), 118.68 (d, $J = 18.2$ Hz), 116.77 (d, $J = 17.6$ Hz), 50.7, 7.1. ^{19}F NMR (500 MHz, CDCl_3) δ -57.5, -62.8. HRMS (ESI-TOF) calcd. for $\text{C}_{15}\text{H}_{14}\text{F}_2\text{NO}_2\text{S}$ $[\text{M}+\text{H}]^+$ 310.0713; found 310.0706.

3.7.18

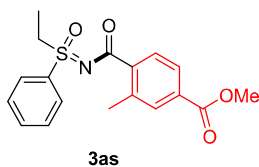


N-(3,4,5-trimethoxybenzoyl)-*S*-ethyl-*S*-phenyl sulfoximine (**3ar**):

The reaction was carried out using 169 mg of sulfoximine according to general procedure. The title compound was obtained as white solid, 348 mg, 95% yield, m.p. 157–158 °C. The residue was isolated by silica-gel column chromatography using 18% ethyl acetate:hexane as an eluent, R_f 0.56 (40% EA: hex). ^1H NMR (500 MHz, CDCl_3) δ 7.96–7.94 (m, 2H), 7.65 (t, $J = 7.4$ Hz, 1H), 7.57 (t, $J = 7.6$ Hz, 2H), 7.43 (s, 2H), 3.86 (d, $J = 1.5$ Hz, 9H), 3.63–3.51 (m, 2H), 1.28 (t, $J = 7.4$ Hz, 3H); ^{13}C NMR (125 MHz, CDCl_3) δ 173.5, 152.6, 141.5, 136.5, 133.7, 130.9, 129.6, 127.8, 106.6, 60.8, 56.1, 50.6, 7.1. HRMS (ESI-TOF) calcd. for $\text{C}_{18}\text{H}_{22}\text{NO}_5\text{S}$ $[\text{M}+\text{H}]^+$ 364.1219; found 364.1203.

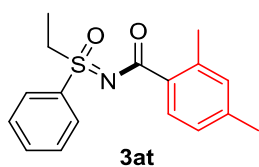
3.7.19

N-(methyl-2-methyl-4-benzoate-benzoyl)-*S*-ethyl-*S*-phenyl sulfoximine (**3as**):



The reaction was carried out using 169 mg of sulfoximine according to general procedure. The title compound was obtained as white solid, 293 mg, 85% yield, m.p. 130–131 °C. The residue was isolated by silica-gel column chromatography using 18% ethyl acetate:hexane as an eluent, R_f 0.56 (40% EA: hex). ^1H NMR (500 MHz, CDCl_3) δ 8.03 (d, $J = 8.4$ Hz, 1H), 7.96 (d, $J = 7.9$ Hz, 2H), 7.84 (d, $J = 6.7$ Hz, 2H), 7.67–7.64 (m, 1H), 7.60–7.57 (m, 2H), 3.89 (d, $J = 1.1$ Hz, 3H), 3.54 (dt, $J = 14.3, 7.0$ Hz, 2H), 2.58 (s, 3H), 1.28 (t, $J = 7.3$ Hz, 3H); ^{13}C NMR (125 MHz, CDCl_3) δ 175.9, 166.7, 139.8, 138.9, 136.1, 133.9, 132.4, 131.5, 130.2, 129.7, 127.9, 126.5, 52.2, 50.7, 21.4, 7.1. HRMS (ESI-TOF) calcd. for $\text{C}_{18}\text{H}_{20}\text{NO}_4\text{S}$ $[\text{M}+\text{H}]^+$ 346.1113; found 346.1102.

3.7.20

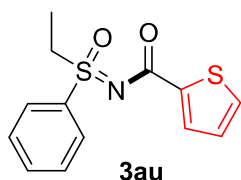


***N*-(2,5-dimethylbenzoyl)-*S*-ethyl-*S*-phenyl sulfoximine (3at):**

The reaction was carried out using 169 mg of sulfoximine according to general procedure. The title compound was obtained as white solid, 244 mg, 81% yield, m.p. 75 °C. The residue was isolated by silica-gel column chromatography using 18% ethyl acetate:hexane as an eluent, R_f 0.56 (40% EA: hex). ^1H NMR (500 MHz, CDCl_3) δ 8.03 (d, $J = 7.8$ Hz, 1H), 8.02–7.96 (m, 2H), 7.67–7.64 (m, 1H), 7.61–7.58 (m, 2H), 7.04–6.99 (m, 2H), 3.55 (q, $J = 7.4$ Hz, 2H), 2.57 (s, 3H), 2.34 (s, 3H), 1.29 (t, $J = 7.4$ Hz, 3H); ^{13}C NMR (125 MHz, CDCl_3) δ 176.4, 141.3, 139.6, 136.7, 133.7, 132.50 (d, $J = 4.9$ Hz), 131.0, 129.6, 128.0, 126.1, 50.7, 22.0, 21.4, 7.3. HRMS (ESI-TOF)

calcd. for $C_{17}H_{20}NO_2S$ $[M+H]^+$ 302.1215; found 302.1223.

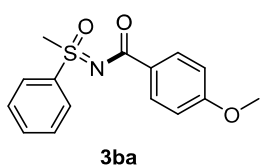
3.7.21



N-(2-thienylacetyl)-*S,S*-methylphenyl sulfoximines (3au):

The reaction was carried out using 169 mg of sulfoximine according to general procedure. The title compound was obtained as white solid, 217 mg, 78% yield m.p. 88–87 °C. The residue was isolated by silica-gel column chromatography using 20% ethyl acetate:hexane as an eluent, R_f 0.6 (40% EA: hex). 1H NMR (500 MHz, $CDCl_3$) δ 8.01–8.00 (m, 2H), 7.80 (dd, $J = 3.6, 1.1$ Hz, 1H), 7.69 (t, $J = 7.4$ Hz, 1H), 7.61 (t, $J = 7.7$ Hz, 2H), 7.48 (dd, $J = 4.9, 1.0$ Hz, 1H), 7.07 (dd, $J = 4.9, 3.8$ Hz, 1H), 3.61 (q, $J = 7.4$ Hz, 2H), 1.30 (t, $J = 7.4$ Hz, 3H); ^{13}C NMR (125 MHz, $CDCl_3$) δ 168.9, 141.3, 136.4, 133.9, 132.2, 131.6, 129.7, 128.1, 127.8, 50.8, 7.3. **HRMS** (ESI-TOF) calcd. for $C_{13}H_{14}NO_2S_2$ $[M+H]^+$ 280.0466; found 280.0458.

3.7.22

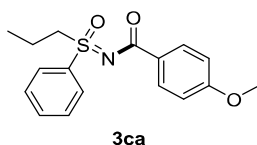


N-(4-methoxybenzoyl)-*S*-methyl-*S*-phenyl sulfoximine (3ba):

The reaction was carried out using 155 mg of sulfoximine according to general procedure. The title compound was obtained as white solid, 250 mg, 87% yield, m.p. 137–138 °C. The residue was isolated by silica-gel column chromatography using 20% ethyl acetate:hexane as an eluent, R_f 0.56 (40% EA: hex). 1H NMR (500 MHz, $CDCl_3$) δ 8.14–8.11 (m, 2H), 8.06–8.04 (m, 2H), 7.68–7.61 (m, 1H), 7.61 (t, $J = 7.8$ Hz, 2H), 6.90 (d, $J = 8.6$ Hz, 2H), 3.85 (s, 3H), 3.46 (s, 3H); ^{13}C NMR (125 MHz, $CDCl_3$) δ 173.9, 163.0,

139.3, 133.8, 131.6, 129.7, 128.4, 127.3, 113.3, 55.5, 44.5. HRMS (ESI-TOF) calcd. for $C_{15}H_{16}NO_3S$ $[M+H]^+$ 290.0851; found 290.0852.

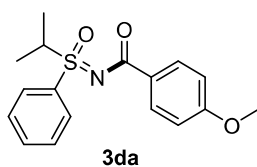
3.7.23



N-(4-methoxybenzoyl)-*S*-phenyl-*S*-propyl- sulfoximine (**3ca**):

The reaction was carried out using 183 mg of sulfoximine according to general procedure. The title compound was obtained as white solid, 288 mg, 91% yield, m.p. 92–93 °C. The residue was isolated by silica-gel column chromatography using 18% ethyl acetate:hexane as an eluent, R_f 0.58 (40% EA: hex). 1H NMR (500 MHz, $CDCl_3$) δ 8.13 (d, $J = 8.8$ Hz, 2H), 8.00 (d, $J = 7.5$ Hz, 2H), 7.66 (t, $J = 7.4$ Hz, 1H), 7.59 (t, $J = 7.6$ Hz, 2H), 6.90 (d, $J = 8.8$ Hz, 2H), 3.85 (s, 3H), 3.54 (dddd, $J = 34.7, 14.0, 10.8, 5.3$ Hz, 2H), 1.80 (tdd, $J = 11.0, 7.6, 3.6$ Hz, 1H), 1.72–1.69 (m, 1H), 1.00 (t, $J = 7.4$ Hz, 3H); ^{13}C NMR (125 MHz, $CDCl_3$) δ 173.8, 162.9, 137.6, 133.7, 131.6, 129.6, 128.6, 128.0, 113.3, 57.7, 55.5, 16.4, 12.9. HRMS (ESI-TOF) calcd. for $C_{17}H_{20}NO_3S$ $[M+H]^+$ 318.1164; found 292.0759.

3.7.24

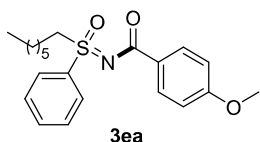


N-(4-methoxybenzoyl)-*S*-phenyl-*S*-iso-propyl- sulfoximine (**3da**):

The reaction was carried out using 183 mg of sulfoximine according to general procedure. The title compound was obtained as white solid, 253 mg, 80% yield, m.p. 120–121 °C. The residue was isolated by silica-gel column chromatography using 18% ethyl acetate:hexane as an eluent, R_f 0.56 (40% EA: hex). 1H NMR (500

MHz, CDCl₃) δ 8.15–8.12 (m, 2H), 7.93–7.91 (m, 2H), 7.65 (t, J = 7.4 Hz, 1H), 7.57 (t, J = 7.7 Hz, 2H), 6.91–6.88 (m, 2H), 3.85 (s, 3H), 3.74 (dq, J = 13.7, 6.8 Hz, 1H), 1.45 (d, J = 6.8 Hz, 3H), 1.30 (d, J = 6.8 Hz, 3H); ¹³C NMR (125 MHz, CDCl₃) δ 173.7, 162.8, 134.9, 133.6, 131.5, 129.4, 128.9, 128.7, 113.2, 56.3, 55.4, 15.9, 15.4. HRMS (ESI-TOF) calcd. for C₁₇H₂₀NO₃S [M+H]⁺ 318.1164; found 318.1149.

3.7.25

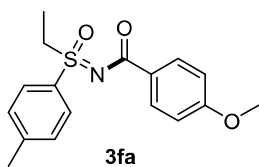


N-(4-methoxybenzoyl)-*S*-heptyl-*S*-phenyl sulfoximine (3ea):

The reaction was carried out using 239 mg of sulfoximine according to general procedure. The title compound was obtained as white solid, 339 mg, 91% yield, m.p. 87–88 °C. The residue was isolated by silica-gel column chromatography using 16% ethyl acetate:hexane as an eluent, R_f 0.59 (40% EA: hex). ¹H NMR (500 MHz, CDCl₃) δ 8.12 (t, J = 5.8 Hz, 2H), 7.99 (d, J = 7.5 Hz, 2H), 7.65 (t, J = 7.4 Hz, 1H), 7.58 (t, J = 7.6 Hz, 2H), 6.89 (d, J = 8.9 Hz, 2H), 3.83 (d, J = 6.9 Hz, 3H), 3.54 (dddd, J = 40.9, 13.9, 11.2, 5.1 Hz, 2H), 1.78–1.60 (m, 2H), 1.38–1.31 (m, 2H), 1.27–1.17 (m, 6H), 0.84 (t, J = 7.0 Hz, 3H); ¹³C NMR (125 MHz, CDCl₃) δ 173.7, 162.9, 137.5, 133.6, 131.5, 129.6, 128.5, 127.9, 113.2, 56.0, 55.4, 31.4, 28.7, 28.1, 22.52 (d, J = 6.1 Hz), 14.0. HRMS (ESI-TOF) calcd. for C₂₁H₂₈NO₃S [M + H]⁺ 374.1790; found 374.1791.

3.7.26

N-(4-methoxybenzoyl)-*S*-ethyl-*S*-(4-methylphenyl) sulfoximine (3fa):

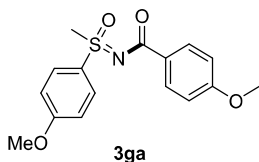


The reaction was carried out using 183 mg of sulfoximine according to general procedure. The title compound was obtained as white solid, 285 mg, 90% yield, m.p. 117–118 °C. The residue was isolated by silica-gel column chromatography using 18% ethyl acetate:hexane as an eluent, R_f 0.56 (40% EA: hex). ^1H NMR (500 MHz, CDCl_3) δ 8.14–8.11 (m, 2H), 7.85 (d, $J = 8.3$ Hz, 2H), 7.37 (d, $J = 8.2$ Hz, 2H), 6.89–6.87 (m, 2H), 3.83 (s, 3H), 3.59–3.54 (m, 2H), 2.43 (s, 3H), 1.27 (t, $J = 7.4$ Hz, 3H); ^{13}C NMR (125 MHz, CDCl_3) δ 173.7, 162.8, 144.7, 133.6, 131.5, 130.2, 128.5, 128.0, 113.2, 55.4, 50.7, 21.6, 7.3. HRMS (ESI-TOF) calcd. for $\text{C}_{17}\text{H}_{20}\text{NO}_3\text{S}$ $[\text{M}+\text{H}]^+$ 318.1164 found 318.1154.

3.7.27

N-(4-methoxybenzoyl)-*S*-methyl-*S*-(4-methoxyphenyl)

sulfoximine (3ga):

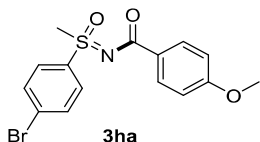


The reaction was carried out using 185 mg of sulfoximine according to general procedure. The title compound was obtained as white solid, 295 mg, 93% yield, m.p. 122–123 °C. The residue was isolated by silica-gel column chromatography using 18% ethyl acetate:hexane as an eluent, R_f 0.57 (40% EA: hex). ^1H NMR (500 MHz, CDCl_3) δ 8.10 (dd, $J = 5.7, 5.3$ Hz, 2H), 7.8 (dd, $J = 8.9, 1.2$ Hz, 2H), 7.02 (dd, $J = 8.9, 1.2$ Hz, 2H), 6.88–6.86 (m, 2H), 3.84 (d, $J = 1.5$ Hz, 3H), 3.82 (d, $J = 1.2$ Hz, 3H), 3.42 (d, $J = 0.9$ Hz, 3H); ^{13}C NMR (125 MHz, CDCl_3) δ 173.8, 163.8, 162.8, 131.4, 130.2, 129.3, 128.5, 114.9, 113.2, 55.8, 55.4, 44.7. HRMS (ESI-TOF)

calcd. for C₁₈H₁₈NO₄S [M+H]⁺ 320.0957 found 320.0945.

3.7.28

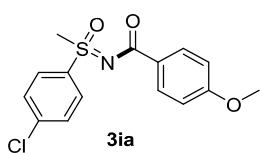
N-(4-methoxybenzoyl)-*S*-methyl-*S*-(4-bromophenyl) sulfoximine (3ha):



The reaction was carried out using 234 mg of sulfoximine according to general procedure. The title compound was obtained as white solid, 293 mg, 80% yield, m.p. 212–214 °C. The residue was isolated by silica-gel column chromatography using 18% ethyl acetate:hexane as an eluent, R_f 0.54 (40% EA: hex). ¹H NMR (500 MHz, CDCl₃) δ 8.10 (d, *J* = 8.7 Hz, 2H), 7.90 (d, *J* = 8.5 Hz, 2H), 7.74 (d, *J* = 8.6 Hz, 2H), 6.89 (d, *J* = 8.8 Hz, 2H), 3.85 (s, 3H), 3.43 (s, 3H); ¹³C NMR (125 MHz, CDCl₃) δ 173.8, 163.1, 138.4, 133.1, 131.6, 129.2, 128.8, 128.0, 113.4, 55.5, 44.5. HRMS (ESI-TOF) calcd. for C₁₅H₁₅BrNO₃S [M+H]⁺ 367.9956 found 367.9935.

3.7.29

N-(4-methoxybenzoyl)-*S*-methyl-*S*-(4-chlorophenyl) sulfoximine (3ia):

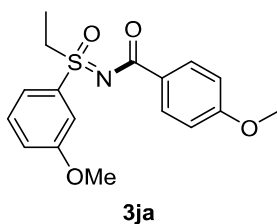


The reaction was carried out using 189 mg of sulfoximine according to general procedure. The title compound was obtained as white solid, 265 mg, 82% yield, m.p. 155–157 °C. The residue was isolated by silica-gel column chromatography using 18% ethyl acetate:hexane as an eluent, R_f 0.52 (40% EA: hex). ¹H NMR (500 MHz, CDCl₃) δ 8.09 (d, *J* = 8.8 Hz, 2H), 7.96 (d, *J* = 8.6 Hz, 2H), 7.56 (d, *J* = 8.6 Hz, 2H), 6.89 (d, *J* = 8.8 Hz, 2H), 3.84 (s, 3H), 3.43 (s, 3H); ¹³C NMR (125 MHz, CDCl₃) δ 173.8, 163.1, 140.6, 137.8,

131.6, 130.0, 128.8, 128.0, 113.3, 55.5, 44.5. HRMS (ESI-TOF) calcd. for C₁₅H₁₅ClNO₃S [M+H]⁺ 324.0461; found 324.0471.

3.7.30

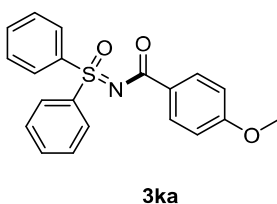
N-(4-methoxybenzoyl)-*S*-ethyl-*S*-(3-methoxyphenyl) sulfoximine (3ja):



The reaction was carried out using 119 mg of sulfoximine according to general procedure. The title compound was obtained as white solid, 290 mg, 87% yield, m.p. 102–103 °C. The residue was isolated by silica-gel column chromatography using 18% ethyl acetate:hexane as an eluent, *R_f* 0.55 (40% EA: hex). ¹H NMR (500 MHz, CDCl₃) δ 8.15–8.12 (m, 2H), 7.55–7.47 (m, 3H), 7.18–7.16 (m, 1H), 6.93–6.85 (m, 2H), 3.87 (s, 3H), 3.85 (s, 3H), 3.58 (qd, *J* = 7.3, 2.6 Hz, 2H), 1.31 (t, *J* = 7.4 Hz, 3H); ¹³C NMR (125 MHz, CDCl₃) δ 173.8, 162.9, 160.4, 138.1, 131.6, 130.7, 128.5, 120.0, 120.0, 113.3, 112.9, 55.8, 55.5, 50.8, 7.4. HRMS (ESI-TOF) calcd. for C₁₉H₂₀NO₄S [M+H]⁺ 334.1113 found 334.1102.

3.7.31

N-(4-methoxybenzoyl)-*S,S*-diphenyl sulfoximine (3ka):

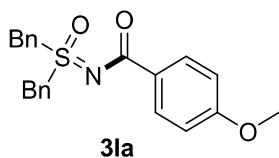


The reaction was carried out using 217 mg of sulfoximine according to general procedure. The title compound was obtained as white solid, 301 mg, 86% yield, m.p. 148–149 °C. The residue was isolated by silica-gel column chromatography using 16% ethyl acetate:hexane as an eluent, *R_f* 0.61 (40% EA: hex). ¹H NMR (500 MHz, CDCl₃) δ 8.23–8.20 (m, 2H), 8.07–8.05 (m, 4H), 7.59–7.51 (m, 6H), 6.94–6.92 (m, 2H), 3.86 (s, 3H); ¹³C NMR (125 MHz,

CDCl₃) δ 173.5, 163.0, 140.2, 133.2, 131.7, 129.6, 128.6, 127.7, 113.3, 55.5. HRMS (ESI-TOF) calcd. for C₂₀H₁₈NO₃S [M+H]⁺ 352.1007; found 352.1000.

3.7.32

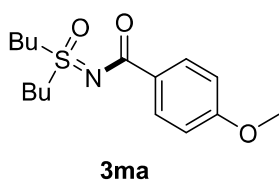
N-(4-methoxybenzoyl)-*S,S*-dibenzyl sulfoximine (3la):



The reaction was carried out using 245 mg of sulfoximine according to general procedure. The title compound was obtained as white solid, 288 mg, 76% yield, m.p. 134–135 °C. The residue was isolated by silica-gel column chromatography using 16% ethyl acetate:hexane as an eluent, *R_f* 0.61 (40% EA: hex). ¹H NMR (500 MHz, CDCl₃) δ 8.05 (d, *J* = 8.8 Hz, 2H), 7.49 – 7.29 (m, 10H), 6.88 (d, *J* = 8.8 Hz, 2H), 4.79 (d, *J* = 13.5 Hz, 2H), 4.63 (d, *J* = 13.5 Hz, 2H), 3.85 (s, 3H). ¹³C NMR (126 MHz, CDCl₃) δ 173.7, 163.0, 131.5, 129.5, 129.1, 128.4, 126.5, 113.3, 56.9, 55.5. HRMS (ESI-TOF) calcd. for C₁₅H₁₆NO₃S [M+H]⁺ 279.1242; found 279.1246.

3.7.33

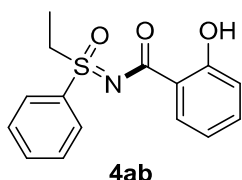
N-(4-methoxybenzoyl)-*S,S*-dibutyl sulfoximine (3ma):



The reaction was carried out using 177 mg of sulfoximine according to general procedure. The title compound was obtained as white solid, 255 mg, 82% yield, m.p. 35-37 °C. The residue was isolated by silica-gel column chromatography using 16% ethyl acetate:hexane as an eluent, *R_f* 0.61 (40% EA: hex). ¹H NMR (500 MHz, CDCl₃) δ 8.02–7.96 (m, 2H), 6.79 (d, *J* = 8.7 Hz, 2H), 3.75 (s, 3H), 3.52–3.44 (m, 2H), 3.31–3.24 (m, 2H), 1.77–1.72 (m, 4H), 1.39 (dd, *J* = 15.0, 7.5 Hz, 4H), 0.86 (t, *J* = 7.4 Hz, 6H). ¹³C NMR (125

MHz, CDCl₃) δ 173.4, 162.9, 131.4, 128.5, 113.3, 55.5, 51.4, 29.8, 23.6, 21.8, 13.7. HRMS (ESI-TOF) calcd. for C₁₅H₁₆NO₃S [M+H]⁺ 311.1555; found 311.1552.

3.7.34



N-(2-Hydroxybenzoyl)-*S,S*-ethylphenylsulfoximine (**4ab**):

The reaction was carried out using 150 mg of *N*-Benzoyl-*S,S*-methylphenylsulfoximine **3ab** according to mentioned procedure. The title compound was obtained as colorless viscous oil (111 mg, 0.38mmol, 70%). The residue was isolated by silica-gel column chromatography using 45% ethyl acetate:hexane as an eluent, R_f 0.3 (80% EA: hex). ¹H NMR (500 MHz, CDCl₃) δ 11.87 (s, 1H), 8.12 (dd, J = 7.9, 1.7 Hz, 1H), 8.02–7.95 (m, 2H), 7.72 (dd, J = 10.6, 4.3 Hz, 1H), 7.63 (dd, J = 10.6, 4.9 Hz, 2H), 7.44–7.37 (m, 1H), 6.92 (d, J = 8.3 Hz, 1H), 6.89–6.84 (m, 1H), 3.59 (ddt, J = 25.7, 14.2, 7.2 Hz, 2H), 1.34 (t, J = 7.4 Hz, 3H). ¹³C NMR (125 MHz, CDCl₃) δ 178.2, 162.2, 136.1, 135.0, 134.2, 131.2, 129.9, 128.0, 118.6, 117.6, 76.9, 51.2, 7.2. HRMS (ESI-TOF) calcd. for C₁₅H₁₆NO₃S [M+H]⁺ 290.0851; found 290.0845.

3.8 Spectral data of product

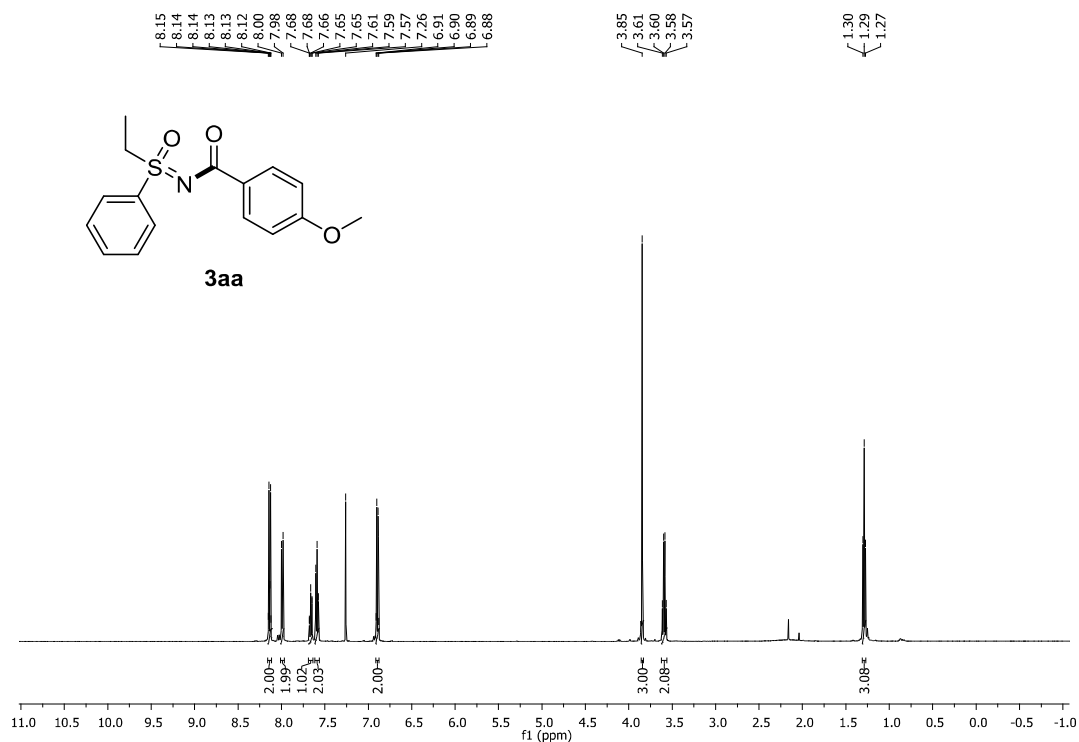


Figure 3.2: ¹H NMR for 3aa

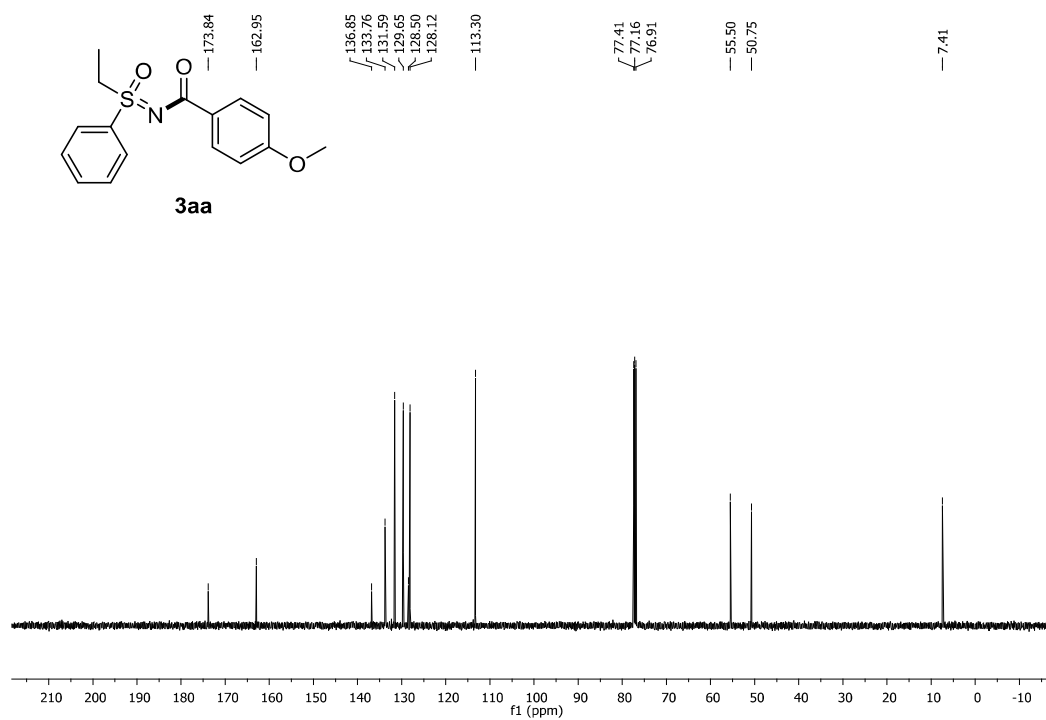


Figure 3.3: ¹³C NMR for 3aa

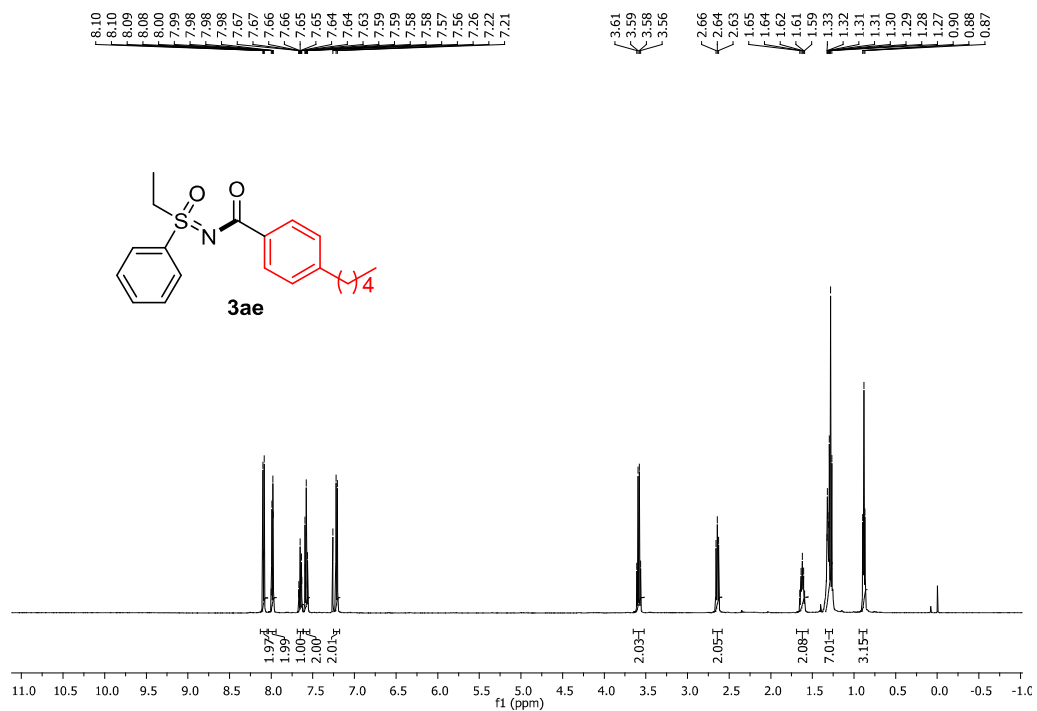


Figure 3.4: ¹H NMR for 3ae

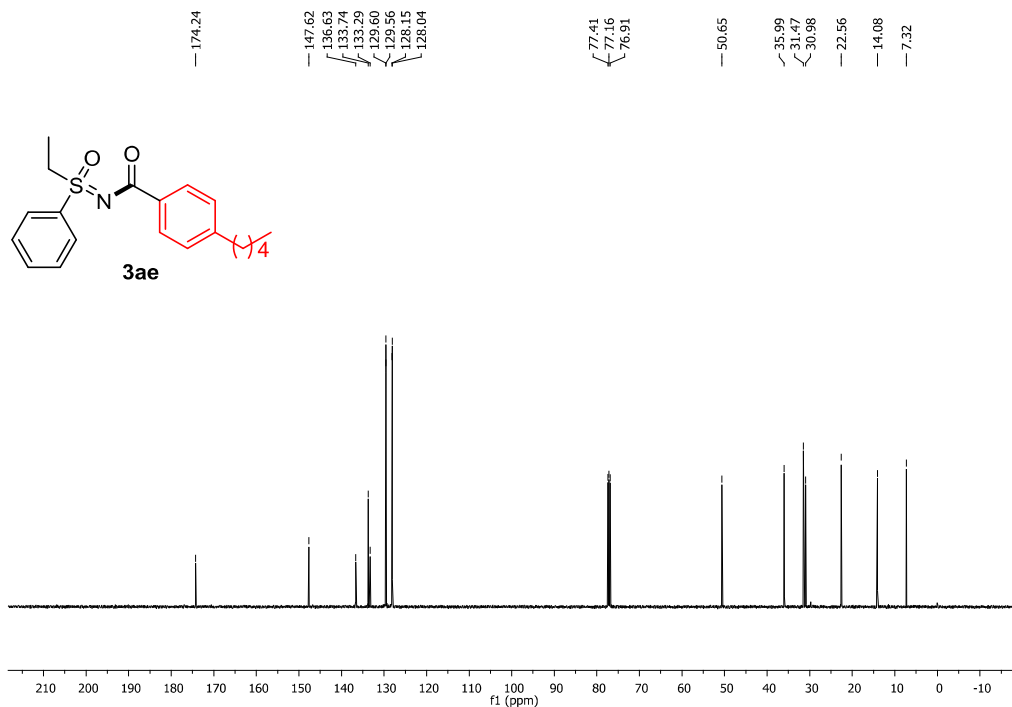


Figure 3.5: ¹³C NMR for 3ae

3.9 References

- [1] U. Lucking, "Sulfoximine: a neglected opportunity in medicinal chemistry," *Angewandte Chemie International Edition*, **52** (2013) 9399-9408.
- [2] J.A. Sirvent, U. Lucking, "Novel pieces for the emerging picture of sulfoximine in drug discovery: synthesis and evaluation of sulfoximine of marketed drugs and advanced clinical candidates," *ChemMedChem*, **12** (2017) 487-501.
- [3] M. Frings, C. Bolm, A. Blum, C. Gnamm, "Sulfoximines from a medicinal chemist's perspective: physicochemical and in vitro parameters relevant for drug discovery," *European Journal of Medicinal Chemistry*, **126** (2017) 225-245.
- [4] U. Lucking, "Neglected sulfur (vi) pharmacophores in drug discovery: exploration of novel chemical space by the interplay of drug design and method development," *Organic Chemistry Frontiers*, **6** (2019) 1319-1324.
- [5] M. Reggelin, C. Zur, "Sulfoximines: structures, properties and synthetic applications," *Synthesis-Stuttgart*, **01** (2000) 1-64.
- [6] V. Bizet, R. Kowalczyk, C. Bolm, "Fluorinated sulfoximines: syntheses, properties and applications," *Chemical Society Reviews*, **43** (2014) 2426-2438.
- [7] M.R. Yadav, R.K. Rit, A.K. Sahoo, "Sulfoximines: a reusable directing group for chemo and regioselective *ortho* C-H oxidation of arenes," *Chemistry—A European Journal*, **18** (2012) 5541-5545.
- [8] A. Hosseinian, L.Z. Fekri, A. Monfared, E. Vessally, M. Nikpass, "Transition-metal-catalyzed C–N cross-coupling reactions of *N*-unsubstituted sulfoximines: a review," *Journal of Sulfur Chemistry*, **39** (2018) 674-698.
- [9] M.T. Reetz, O.G. Bondarev, H.J. Gais, C. Bolm, "BINOL-derived *N*-phosphino sulfoximines as ligands for asymmetric catalysis," *Tetrahedron Letters*, **46** (2005) 5643-5646.
- [10] M. Frings, I. Thome, C. Bolm, "Synthesis of chiral sulfoximine-based thioureas and their application in asymmetric organo-catalysis," *Beilstein Journal of Organic Chemistry*, **8** (2012) 1443-1451;
- [11] Z. Li, H. Yu, C. Bolm, "Dibenzothiophene sulfoximine as an NH₃ surrogate in the synthesis of primary amines by copper-catalyzed C-X and C-H bond amination," *Angewandte Chemie International Edition*, **56** (2017) 9532-9535.
- [12] M. Zenzola, R. Doran, R. Luisi, J.A. Bull, "Synthesis of sulfoximine carbamates by rhodium-catalyzed nitrene transfer of carbamates to sulfoxides," *The Journal of Organic Chemistry*, **80** (2015) 6391-6399.
- [13] J.A. Bull, L. Degennaro, R. Luisi, "Straightforward strategies for the preparation of *NH*-sulfoximines: a serendipitous story," *Synlett*, **28** (2017) 2525-2538.
- [14] C. Moessner, C. Bolm, "Cu(OAc)₂-Catalyzed *N*-arylations of sulfoximines with aryl boronic acids," *Organic Letters*, **7** (2005) 2667-2669.
- [15] J. Sedelmeier, C. Bolm, "Efficient copper-catalyzed *N*-arylation of sulfoximines with aryl iodides and aryl bromides," *The Journal of Organic Chemistry*, **70** (2005) 6904-6906.
- [16] F. Teng, J. Cheng, J.T. Yu, "Copper-catalyzed *N*-methylation/ethylation of sulfoximines," *Organic & Biomolecular Chemistry*, **13** (2015) 9934-9937.
- [17] C.M. Hendriks, R.A. Bohmann, M. Bohlem, C. Bolm, "*N*-Alkylations of *NH*-sulfoximines and *NH*-sulfondiimines with alkyl halides mediated by potassium

-
- hydroxide in dimethyl sulfoxide,” *Advanced Synthesis & Catalysis*, **356** (2014) 1847-1852.
- [18] J.R. Dehli, C. Bolm, “Palladium-catalyzed *N*-vinylation of sulfoximines,” *The Journal of Organic Chemistry*, **69** (2004) 8518-8520.
- [19] L. Wang, H. Huang, D.L. Priebbenow, F.F. Pan, C. Bolm, “Copper-catalyzed oxidative cross-coupling of sulfoximines and alkynes,” *Angewandte Chemie International Edition*, **52** (2013) 3478-3480.
- [20] F. Teng, J.T. Yu, Z. Zhou, H.K. Chu, J. Cheng, “Copper-catalyzed *N*-cyanation of sulfoximines by AIBN,” *The Journal of Organic Chemistry*, **80** (2015) 2822-2826.
- [21] D. Zhang, H. Wang, C. Bolm, “Photocatalytic difunctionalisations of alkenes with *N*-SCN sulfoximines,” *Chemical Communications*, **54** (2018) 5772-5775.
- [22] F. Teng, J. Cheng, C. Bolm, “Silver-mediated *N*-trifluoromethylation of sulfoximines,” *Organic Letters*, **17** (2015) 3166-3169.
- [23] C. Bohnen, C. Bolm, “*N*-Trifluoromethylthiolated sulfoximines,” *Organic Letters*, **17** (2015) 3011-3013.
- [24] G.Y. Cho, C. Bolm, “Palladium-catalyzed α -arylation of sulfoximines,” *Organic Letters*, **7** (2005) 1351-1354.
- [25] L. Wang, D.L. Priebbenow, L.H. Zou, C. Bolm, “The copper-catalyzed oxidative *N*-acylation of sulfoximines,” *Advanced Synthesis & Catalysis*, **355** (2013) 1490-1494.
- [26] W.J. Qin, Y. Li, X.X. Yu, W.P. Deng, “TBAI/TBHP catalyzed direct *N*-acylation of sulfoximines with aldehydes,” *Tetrahedron*, **71** (2015) 1182-1186.
- [27] A. Porey, S. Santra, J. Guin, “Direct *N*-acylation of sulfoximines with aldehydes by *N*-heterocyclic carbene catalysis under oxidative conditions,” *Asian Journal of Organic Chemistry*, **5** (2016) 870-873.
- [28] Y. Zou, J. Xiao, Z.H. Peng, W.R. Dong, D.L. An, “Transition metal-free arylation of *NH*-sulfoximines with methyl arenes,” *Chemical Communications*, **51** (2015) 14889-14892.
- [29] D.L. Priebbenow, C. Bolm, “C–H Activation of methyl arenes in the MnO₂-mediated arylation of *N*-chlorosulfoximines,” *Organic Letters*, **16** (2014) 1650-1652.
- [30] M. Muneeswara, S.S. Kotha, G. Sekar, “Iron-catalyzed one-pot *N*-arylation of *NH*-sulfoximines with methylarenes through benzylic C-H bond oxidation,” *Synthesis*, **48** (2016) 1541-1549.
- [31] C. Pimpasri, L. Sumunnee, S. Yotphan, “Copper-catalyzed oxidative decarboxylative coupling of α -keto acids and sulfoximines,” *Organic & Biomolecular Chemistry*, **15** (2017) 4320-4327.
- [32] B.D. Bala, N. Sharma, G. Sekar, “Sulfoximinocarbonylation of aryl halides using heterogeneous Pd/C catalyst,” *RSC Advances*, **6** (2016) 97152-97159.
- [33] M. Beller, X.-F. Wu, “Transition metal catalyzed carbonylation reactions: carbonylative activation of C–X bonds.” *Springer-Verlag Berlin Heidelberg*, **2013**.
- [34] M. Franck-Neumann, Some uses of metal carbonyl complexes in organic synthesis. In: de Meijere A., tom Dieck H. (eds) *Organometallics in organic synthesis*. *Springer, Berlin, Heidelberg*, **1988**.
- [35] X.Y. Wu, R. Ronn, T. Gossas, M. Larhed, “Easy-to-execute carbonylations: microwave synthesis of acyl sulfonamides using Mo(CO)₆ as a solid carbon monoxide source,” *The Journal of Organic Chemistry*, **70** (2005) 3094-3098.
-

-
- [36] N.F.K. Kaiser, A. Hallberg, M. Larhed, "In situ generation of carbon monoxide from solid molybdenum hexacarbonyl. A convenient and fast route to palladium-catalyzed carbonylation reactions," *Journal of Combinatorial Chemistry*, **4** (2002) 109-111.
- [37] K. Yamazaki, Y. Kondo, "Carbonylation reaction of aryl halides on a polymer support using in situ-generated carbon monoxide without the assistance of microwaves," *Journal of Combinatorial Chemistry*, **6** (2004) 121-125.
- [38] P. Nordeman, L.R. Odell, M. Larhed, "Aminocarbonylations employing Mo(CO)₆ and a bridged two-vial system: allowing the use of nitro group substituted aryl iodides and aryl bromides," *The Journal of Organic Chemistry*, **77** (2012) 11393-11398.
- [39] L.R. Odell, F. Russo, M. Larhed, "Molybdenum hexacarbonyl mediated CO gas-free carbonylative reactions," *Synlett*, **23** (2012) 685-698.
- [40] L. Akerbladh, L.S. Schembri, M. Larhed, L.R. Odell, "Palladium (0)-catalyzed carbonylative one-pot synthesis of *N*-acylguanidines," *The Journal of Organic Chemistry*, **82** (2017) 12520-12529.
- [41] B. Roberts, D. Liptrot, L. Alcaraz, T. Luker, M.J. Stocks, "Molybdenum-mediated carbonylation of aryl halides with nucleophiles using microwave irradiation," *Organic Letters*, **12** (2010) 4280-4283.
- [42] W. Ren, M. Yamane, "Mo(CO)₆-Mediated carbamoylation of aryl halides," *The Journal of Organic Chemistry*, **75** (2010) 8410-8415.
- [43] S. Gupta, P. Chaudhary, N. Muniyappan, S. Sabiah, J. Kandasamy, "Copper promoted *N*-alkylation of sulfoximines with alkylboronic acid under mild conditions," *Organic & Biomolecular Chemistry*, **15** (2017) 8493-8498.
- [44] S. Gupta, S. Baranwal, N. Muniyappan, S. Sabiah, J. Kandasamy, "Copper-catalyzed *N*-arylation of sulfoximines with arylboronic acids under mild conditions," *Synthesis*, **51** (2019) 2171-2182.
- [45] S. Baranwal, J. Kandasamy "Copper catalyzed *N*-arylation of sulfoximines with aryldiazonium salts in the presence of DABCO under mild conditions," *Tetrahedron Letters*, **61** (2020) 152079.
- [46] M.R. Yadav, R.K. Rit, A.K. Sahoo, "Sulfoximine directed intermolecular *ortho*-C–H amidation of arenes with sulfonyl azides," *Organic Letters*, **15** (2013) 1638-1641.
- [47] M.R. Yadav, M. Shankar, E. Ramesh, K. Ghosh, A.K. Sahoo, "Ruthenium-catalyzed *ortho*-C–H mono- and di-imidation of arenes with *N*-tosyloxyphthalimide," *Organic Letters*, **17** (2015) 1886-1889.
- [48] K. Raghuvanshi, D. Zell, L. Ackermann, "Ruthenium (II)-catalyzed C–H oxygenations of reusable sulfoximine benzamides," *Organic Letters*, **19** (2017) 1278-1281.
- [49] M. Shankar, K. Ghosh, K. Mukherjee, R.K. Rit, A.K. Sahoo, "One-pot unsymmetrical {[4+2] and [4+2]} double annulations of *o/o'*-C–H bonds of arenes: access to unusual pyranoisoquinolines," *Organic Letters*, **20** (2018) 5144-5148.
- [50] P. Das, J. Guin, "Direct C(sp²)-H hydroxylation of arenes with palladium (II)/oxygen using sulfoximines as a recyclable directing group," *ChemCatChem*, **10** (2018) 2370-2373.
- [51] M.R. Yadav, R.K. Rit, M. Shankar, A.K. Sahoo, "Sulfoximine-directed ruthenium-catalyzed *ortho*-C–H alkenylation of (hetero) arenes: synthesis of EP3 receptor antagonist analogue," *The Journal of Organic Chemistry*, **79** (2014) 6123-6134.
-