

---

---

## List of Publications

---

---

### International Journals:

1. Bandgap and electrochemical engineering for disordered  $\text{LaFeO}_3$ , **Uma Sharma**, U.K. Kailash Veerappan, Pardeep K. Jha, Priyanka A. Jha, and Prabhakar Singh, Journal of Applied Physics, 131 (2), 024901.
2. Charge particle dynamics and electrochemical behavior of  $\text{SrTiO}_{3-\delta}$  as anode material for IT-SOFC applications, **Uma Sharma**, Vani Pawar, Prabhakar Singh, International Journal of Hydrogen Energy, Pages 1278-1289.
3. Highly efficient and robust bifunctional environmental friendly  $\text{LaSrFeTiO}_3$  catalyst for  $\text{H}_2/\text{O}_2$  fuel cells, **Uma Sharma**, Priyanka A. Jha, Pardeep K. Jha, Prabhakar Singh, (Under Review).
4. Advancements in Bifunctional Catalysts for Unitized Regenerative Fuel Cells: Exploring Polaronic Conduction and Heterostructure Designs, **Uma Sharma**, Priyanka A. Jha, Pardeep K. Jha, Prabhakar Singh, (**Accepted** in Journal of Physics D: Applied Physics).
5. To examine the effect of synthesis techniques on the physical and electrochemical properties of Lanthanum Strontium Ferrite (LSF), **Uma Sharma**, Jay N. Mishra, Vani Pawar, Prabhakar Singh (Under Preparation).

### Papers Published as Conference Proceedings:

- Properties of  $\text{La}_x\text{Sr}_{1-x}\text{TiO}_3$  as SOFC Electrode Material, **Uma Sharma**, Priyanka A. Jha, Pardeep K. Jha and Prabhakar Singh Electrical Materials Today: Proceedings, July 22.

### National/International Conference Presentations:

1. 13th National conference on Solid State Ionics, Department of Physics, IIT Roorkee (Poster Presentation).

- 
2. National webinar series on experimental and computational tools for material research, (ICTMR 2020), Central University of Rajasthan.
  3. International e-Conference on Advanced Functional Materials and Optoelectronic Devices, Centre for Renewable Energy, RBIPS, VBS Purvanchal University, Jaunpur (Poster Presentation).
  4. E- international symposium on synthesis and characterization of smart materials and their potential applications, Guru Govind Singh Indraprastha University, New Delhi.
  5. Online Workshop on Rietveld Refinement Method, organized by UGC-DAE Consortium for Scientific Research, Mumbai Centre in association with Indore Centre.
  6. International Conference on Energy Materials and Devices (ICEMD-2022), Department of Physics (MMV) Banaras Hindu University, Varanasi (Oral Presentation, Paper Published in Materials Today: Proceedings, July 22).
  7. Two-day workshop on “Electrochemistry Techniques” organised by Sophisticated Analytic and Technical Help Institute (SATHI), Banaras Hindu University, Varanasi.
  8. International Union of Materials Research Society, Materials Research Society of India
  9. International Conference in Asia-2022, (IUMRS-ICA 2022), IIT Jodhpur (Poster-Presentation).
  10. 9-day Online hands-on Training on Rietveld Refinement conducted by SIAS Research Forum in collaboration with Department of Physics, PSMO College, Tirurangadi.

---

---

## References

---

---

- [1] U. S. E. I. Administration, "Analysis of the Impacts of the Clean Power Plan," no. May, 2015.
- [2] 'Global Energy & CO2 Status Report 2018', International Energy Agency, Paris," IEA, 2019. France,[https://iea.blob.core.windows.net/assets/23f9eb39-7493-4722-aced61433cbffe10/Global\\_Energy\\_and\\_CO2\\_Status\\_Report\\_2018.pdf](https://iea.blob.core.windows.net/assets/23f9eb39-7493-4722-aced61433cbffe10/Global_Energy_and_CO2_Status_Report_2018.pdf).
- [3] "Statistical Review of World Energy 2023", Energy Institute, London, UK, <https://www.energyinst.org/statistical-review>.
- [4] "Global Energy Trends: Insights From The 2023 Statistical Review Of World Energy," Glob. Energy Trends Insights From 2023 Stat. Rev. World Energy.
- [5] A. W. Ando and K. Baylis, *Spatial environmental and natural resource economics*. 2014.
- [6] "Energy and resources," B. Sorensen. *Energy Resour. Sci.* 1975, 189, 255–260, vol. 189, pp. 255–260, 1975.
- [7] "No Title."
- [8] Dusonchet, L.; Favuzza, S.; Massaro, F.; Telaretti, E.; Zizzo, G., *Technol. Legis. status point Station. energy storages EU. Renew. Sustain. Energy Rev.* 2019, 101, 158-167., vol. 101, pp. 158–167, 2019.
- [9] "No Title."
- [10] "No Title."
- [11] M. Biswas and K. C. Sadanala, "Electrolyte Materials for Solid Oxide Fuel Cell," *J. Powder Metall. Min.*, vol. 02, no. 04, pp. 26–55, 2013.
- [12] S. C. Singhal and K. Kendall, "High Temperature Solid Oxide Fuel Cells : Fundamentals , Design and Applications Edited by : Subhash C Singhal and Kevin Kendall," pp. 197–225, 2003.
- [13] T. Saitoh, "Degradation phenomena in the cathode of a solid oxide fuel cell with an alloy separator," vol. 55, pp. 73–79, 1995.
- [14] C. Sun, R. Hui, and J. Roller, "Cathode materials for solid oxide fuel cells: A review," *J. Solid State Electrochem.*, vol. 14, no. 7, pp. 1125–1144, 2010.
- [15] Wang, Z.-L.; Xu, D.; Xu, J.-J.; Zhang, X.-B., *Oxyg. Electrocat. Met. Batter. from aqueous to nonaqueous electrolytes. Chem. Soc. Rev.* 2014, 43 (22), 7746-7786., vol. 43, no. 22, pp. 7746–7786, 2014.
- [16] Nam, G.; Cho, J.; Liu, M. *Recent Adv. Low-Cost, Highly Effic. BiFunctional Oxyg. Electrocat. High-Performance Zinc-Air Batter. ECS Meet. Abstr. IOP Publ.* 2019; p 50, p. 50, 2014.
- [17] Sun, Y.; Liu, X.; Jiang, Y.; Li, J.; Ding, J.; Hu, W.; Zhong, C., *Recent Adv. challenges divalent multivalent Met. electrodes Met. Batter. J. Mater. Chem. A* 2019, 7 (31), 18183-18208., vol. 7, no. 31, pp. 18183–18208, 2019.
- [18] Ganesan, P.; Ramakrishnan, P.; Prabu, M.; Shanmugam, S., *Nitrogen sulfur Co-doped graphene Support. cobalt sulfide nanoparticles as an Effic. air cathode zinc-air Batter. Electrochim. Acta* 2015, 183, 63-69., vol. 183, pp. 63-69., 2015.

- 
- [19] Yan, Y.; Xia, B. Y.; Zhao, B.; Wang, X., *A Rev. noble-metal-free bifunctional Heterog. Catal. overall Electrochem. water Split. J. Mater. Chem. A* 2016, 4 (45), 17587-17603., vol. 4, no. 45, pp. 17587–17603, 2016.
- [20] Wang, T.; Chutia, A.; Brett, D. J.; Shear. P. R.; He, G.; Chai, G.; Park. I. P., *Palladium Alloy. used as Electrocatal. Oxyg. Reduct. React. Energy Environ. Sci.* 2021, 14 (5), 2639-2669.
- [21] Lee, Y.; Suntivich, J.; May, K. J.; Perry, E. E.; Shao-Horn, Y., *Synth. Act. rutile IrO<sub>2</sub> RuO<sub>2</sub> nanoparticles Oxyg. Evol. acid alkaline Solut. J. Phys. Chem. Lett.* 2012, 3 (3), 399-404.
- [22] P. Majumdar, M. K. Bera, D. Pant, and S. Patra, *Enzymatic Electrocatalysis of CO<sub>2</sub> Reduction*. Elsevier, 2018.
- [23] A. Groß, *Computational modeling of electrocatalytic reactions*. Elsevier, 2018.
- [24] M. Açıkyıldız and A. Gürses, “*Electrocatalysis and the Production of Nanoparticles*,” 2014.
- [25] K. Cheung, W. Wong, D. Ma, T. Lai, and K. Wong, “*Transition metal complexes as electrocatalysts — Development and applications in electro-oxidation reactions*,” vol. 251, pp. 2367–2385, 2007.
- [26] A. Manuscript, “*Materials Chemistry A*,” 2016.
- [27] P. Lacorre, F. Goutenoire, O. Bohnke, R. Retoux, and Y. Lallgant, “*Designing fast oxide-ion conductors based on La<sub>2</sub>Mo<sub>2</sub>O<sub>9</sub>*,” *Nature*, vol. 404, no. 6780, pp. 856–858, 2000.
- [28] C. Zhou et al., “*ScienceDirect Application of electrochemical methods in heterogeneous catalysis*,” *Curr. Opin. Chem. Eng.*, vol. 26, pp. 88–95, 2019.
- [29] W. Zhang, W. Lai, and R. Cao, “*Energy-Related Small Molecule Activation Reactions: Oxygen Reduction and Hydrogen and Oxygen Evolution Reactions Catalyzed by Porphyrin- and Corrole-Based Systems*,” *Chem. Rev.*, vol. 117, no. 4, pp. 3717–3797, 2017.
- [30] Y. Xu and H. Ming, “*Chem Soc Rev*,” vol. 46, no. 2, 2017.
- [31] Y. Yan, Y. Xia, and X. Wang, “*A review on noble-metal-free bifunctional heterogeneous catalysts for overall electrochemical water splitting*,” pp. 17587–17603, 2016.
- [32] J. Ludwig, “*Bifunctional oxygen / air electrodes*,” vol. 155, pp. 23–32, 2006.
- [33] Y. Sun and P. Strasser, “*Chem Soc Rev and photochemical approaches for catalytic*,” pp. 6605–6631, 2020.
- [34] *Self-organized single-atom tungsten supported on*. 2020.
- [35] Y. Zhao et al., “*Recent progress on solid oxide fuel cell: Lowering temperature and utilizing non-hydrogen fuels*,” *Int. J. Hydrogen Energy*, vol. 38, no. 36, pp. 16498–16517, 2013.
- [36] Cherry et al. 1995), “*No Title*,” 1995.
- [37] Coey et al. 1999), “*No Title*,” 1999.
- [38] A. T. Bell, “*The impact of nanoscience on heterogeneous catalysis*,” A. T. Bell, *impact Nanosci. Heterog. Catal. Sci.* 299 1688-91.
- [39] *Chemical structures and performance of perovskite oxides*,” M. A. Peña, J. L. G. Fierro, *Chem. Struct.*
-

---

*Perform. perovskite oxides, Chem. Rev. 101 1981-17.*

- [40] *Perovskite oxides materials science in catalysis,*” R. J. H. Voorhoeve, D.W. Johnson Jr, J. P. Remeika, P. K. Gall. *Perovskite oxides Mater. Sci. Catal. Sci.* 195 827-833.
- [41] T. Nakamura, G. Petzow, L. J. Gauckler, *Stab. perovskite phase LaBO<sub>3</sub> (B=V, Cr, Mn, Fe, Co ,Ni) reducing Atmos. I. Exp. results, Mater. Res. Bull.* 14 649-59.
- [42] *Advanced Materials in Catalysis,*” R. J. H. Voorhoeve *Advanced Mater. Catal. J.J. Burt. R.L. Garten eds., Acad. Press. New York,* 129-80.
- [43] G. Amow, I. J. Davidson, and S. J. Skinner, “*A comparative study of the Ruddlesden-Popper series, Lan+1NinO3n+1 (n = 1, 2 and 3), for solid-oxide fuel-cell cathode applications,*” *Solid State Ionics,* vol. 177, no. 13–14, pp. 1205–1210, 2006.
- [44] G. A. S. J. Skinner, “*ORIGINAL PAPER Recent developments in Ruddlesden – Popper nickelate systems for solid oxide fuel cell cathodes,*” pp. 538–546, 2006.
- [45] Z. Gao, L. V. Mogni, E. C. Miller, J. G. Railsback, and S. A. Barnett, “*A perspective on low-temperature solid oxide fuel cells,*” *Energy Environ. Sci.,* vol. 9, no. 5, pp. 1602–1644, 2016.
- [46] “No Title.”
- [47] J. H. Kuo, H. U. Anderson, D. M. Sparlin, “*Oxidation-reduction Behav. undoped Sr-doped LaMnO<sub>3</sub> Nonstoichiom. defect Struct. J. Solid State Chem.,* vol. 83, no. 1, pp. 52–60, 1989.
- [48] O. Yamamoto, Y. Tak. R. Kanno, M. Noda, “*Perovskite-type oxides as Oxyg. electrodes high Temp. oxide fuel cells,*” *Solid State Ionics,* vol. 22, no. 2–3, pp. 241– 246, 1987.
- [49] T. Kenjo M. Nishiya, “*LaMnO<sub>3</sub> air cathodes Contain. ZrO<sub>2</sub> electrolyte high Temp. solid oxide fuel cells,*” *Solid State Ionics,* vol. 57, no. 3–4, pp. 295–302, 1992.
- [50] S. Korea, “*Rational SOFC material design : Solid oxide fuel cells ( SOFCs ) offer great prospects for the most efficient,*” *Mater. Today,* vol. 14, no. 11, pp. 534–546, 2011.
- [51] A. J. Jacobson, “*Materials for solid oxide fuel cells,*” *Chem. Mater.,* vol. 22, no. 3, pp. 660–674, 2010.
- [52] S. P. Jiang, *Development of lanthanum strontium manganite perovskite cathode materials of solid oxide fuel cells: A review,* vol. 43, no. 21. 2008.
- [53] “*Journal of the American Ceramic Society - 2004 - Kleveland - Reactions between Strontium-Substituted Lanthanum Manganite.pdf.*” .
- [54] S. C. Paulson and V. I. Birss, “*Chromium Poisoning of LSM-YSZ SOFC Cathodes I . Detailed Study of the Distribution of Chromium Species at a Porous , Single-Phase Cathode,*” 2004.
- [55] K. Yasumoto, Y. Ina. M. Shiono, M. Dokiya, “*An (La, Sr)(Co, Cu) O<sub>3</sub>– δ cathode Reduc. Temp. SOFCs,*” *Solid State Ionics,* vol. 148, no. 3–4, pp. 545–549, 2002.
- [56] *The chemical oxygen surface exchange and bulk diffusion coefficient determined by impedance spectroscopy of porous La<sub>0.58</sub> Sr<sub>0.4</sub> Co<sub>0.2</sub> Fe<sub>0.8</sub> O<sub>3</sub>- ( LSCF ) cathodes,*” *Solid State Ionics,* vol. 269, no. January 2015, 2021.
- [57] Y. Fu, A. Subardi, M. Hsieh, and W. Chang, “*(M = Mn, Fe, Ni, Cu) Perovskite Cathodes for IT-SOFCs,*” vol. 1352, no. 37447, pp. 1345–1352, 2016.

- [58] F. Tietz, V. A. C. Haanappel, A. Mai, J. Mertens, and D. Stöver, "Performance of LSCF cathodes in cell tests," *J. Power Sources*, vol. 156, no. 1 SPEC. ISS., pp. 20–22, 2006.
- [59] "Advanced Materials - 2009 - Vohs - High-Performance SOFC Cathodes Prepared by Infiltration.pdf."
- [60] A. Mai et al., "Time-dependent performance of mixed-conducting SOFC cathodes," *Solid State Ionics*, vol. 177, no. 19-25 SPEC. ISS., pp. 1965–1968, 2006.
- [61] A. Agudero, L. Fawcett, and S. Taub, "Materials development for intermediate-temperature solid oxide electrochemical devices Materials development for intermediate-temperature solid oxide electrochemical devices," no. May, 2012.
- [62] D.S. Mathew, R.-S. Juang, An Overv. Struct. Magn. spinel ferrite nanoparticles their Synth. microemulsions, *Chem. Eng. J.* 129 51–65.
- [63] V.A.M. Brabers, "Chapter 3 Progress in spinel ferrite research," V.A.M. Brabers, Chapter 3 Prog. spinel ferrite Res. Handb. Magn. Mater. J. Ind. Eng. Chem. 60 53-84 30 8 189–324.
- [64] (Salah et al. 2012, "No Title."
- [65] "Advances in ferrites," Snelling Adv. ferrites, E.C.; (1), Ed. by Srivastava C.M.; Patni, M.J.; New Delhi, India, Oxford IBH 1989 579.
- [66] "No Title," Lebourgeois, R.; Ganne J.P.; Lioret, B. ; *Phys. J. ; IV Fr. 7 suppl.* 1997, 105.
- [67] "Crystal chemistry of iron-containing perovskites," Phase Transitions," C. McCammon, "Crystal Chem. iron-containing perovskites," *Phase Transitions*, vol. 58, no. 1-3992, pp. 1–26, 1996.
- [68] "bond lengths in LaFeO<sub>3</sub>," M. Marezio P. D. Dernier, "Bond lengths LaFeO<sub>3</sub>," *Mater. Res. Bull.* vol. 6, no. 1, pp. 23–29, 1971.
- [69] "Theoretical and experimental studies of charge ordering in CaFeO<sub>3</sub> and SrFeO<sub>3</sub> crystals," E. A. Kotomin, A. Kuzmin, J. Purans al., "Theoretical Exp. Stud. Charg. ordering CaFeO<sub>3</sub> SrFeO<sub>3</sub> crystals," *Phys. Status Solidi (B)*, vol. 259, no. 1, Artic. ID 2100238, 2022.
- [70] R. Maity, A. Dutta, S. Halder, S. Shannigrahi, K. Mandal, T. P. Sinha, "Enhanced photocatalytic Act. Transp. Prop. Electron. Struct. Mn doped GdFeO<sub>3</sub> Synth. using sol–gel Process. Phys. Chem. Chem. Physics, vol.
- [71] S. Sumithra N. V. Jaya, "Structural, Opt. magnetization Stud. Fe-doped CaSnO<sub>3</sub> nanoparticles via hydrothermal route," *J. Mater. Sci. Mater. Electron.* vol. 29, no. 5, pp. 4048–4057, 2018.
- [72] S. Sumithra N. V. Jaya, "Structural, Opt. magnetization Stud. Fe-doped CaSnO<sub>3</sub> nanoparticles via hydrothermal route," *J. Mater. Sci. Mater. Electron.* vol. 29, no. 5, pp. 4048–4057, 2018. [46] J. Wang, J. B. Neat.
- [73] P. V. Anikina, A. A. Markov, M. V. Patrakeev, I. A. Leonidov, V. L. Kozhevnikov, "High-temperature Transp. Stab. SrFe<sub>1-x</sub>Nb<sub>x</sub>O<sub>3-δ</sub>," *Solid State Sci.* vol. 11, no. 6, pp. 1156–1162, 2009.
- [74] R. Lan, P. I. Cowin, S. Sengodan, S. Tao, "A perovskite oxide with high Conduct. both air reducing Atmos. use as electrode solid oxide fuel cells," *Sci. Reports*, vol. 6, no. 1, Artic. ID 31839, 2016.
- [75] N. T. Hung, L. H. Bac, N. N. Trung, N. T. Hoang, P. Van Vinh, D. D. Dung, "Room-temperature Ferromagn. Fe-based perovskite solid Solut. lead-free Ferroelectr. Bi<sub>0.5</sub>Na<sub>0.5</sub>TiO<sub>3</sub> Mater. J. Magn.

- [76] V. Guigoz, L. Balan, A. Aboulaich, R. Schneider, T. Gries, "Heterostructured thin LaFeO<sub>3</sub>/g-C<sub>3</sub>N<sub>4</sub> Film. Effic. Photoelectrochem. Hydrog. Evol. Int. J. Hydrog. Energy, vol. 45, no. 35, pp. 17468–17479, 2020.
- [77] I. M. Nassar, S. Wu, L. Liang, X. Li, "Facile Prep. 2018 n -type LaFeO<sub>3</sub> perovskite Film Effic. Photoelectrochem. water Split. Chem. Sel. vol. 3, pp. 968–972, 2018.
- [78] B.-J. Kim, E. Fabbri, D. F. Abbott al., "Functional role Fe-doping Co-based perovskite oxide Catal. Oxyg. Evol. React. J. Am. Chem. Soc. vol. 141, no. 13, pp. 5231–5240, 2019.
- [79] O. V. Nkwachukwu, C. Muzenda, B. O. Ojo al., "Photoelectrochemical Degrad. Org. Pollut. a La<sub>3</sub>+ doped BiFeO<sub>3</sub> perovskite," Catal. vol. 11, no. 9, p. 1069, 2021.
- [80] U. A. Joshi, J. S. Jang, P. H. Borse, J. S. Lee, "Microwave Synth. single-crystalline perovskite BiFeO<sub>3</sub> nanocubes photoelectrode photocatalytic Appl. Appl. Phys. Lett. vol. 92, no. 24, p. 242106, 2008.
- [81] V. Lantto, S. Saukko, N. N. Toan, L. F. Reyes, C. G. Granqvist, "Gas Sens. with perovskite-like oxides having ABO<sub>3</sub> BO<sub>3</sub> Struct. J. Electro-ceramics, vol. 13, no. 1-3, pp. 721–726, 2004.
- [82] F. Zaza, V. Pallozzi, E. Serra, "Optimization Work. Cond. perovskite-based gas Sens. devices by multiregression Anal. J. Nanotechnology, vol. 2019, pp. 1–19, 2019.
- [83] L. Ma, S. Ma, X. Shen al., "PrFeO<sub>3</sub> hollow nanofibers as a highly Effic. gas Sens. acetone Detect. Sensors Actuators B Chem. vol. 255, pp. 2546–2554, 2018.
- [84] K. Yang, J. Ma, X. Qiao, Y. Cui, L. Jia, H. Wang, "Hierarchical porous LaFeO<sub>3</sub> nanostructure Effic. trace Detect. formaldehyde," Sensors Actuators B Chem. vol. 313, p. 128022, 2020.
- [85] "Absorption performance of cetyl trimethyl,' Molecules," A. B. La I. Process. "Absorption Perform. cetyl trimethyl," Mol. vol. 25, pp. 1–15, 2020.
- [86] H. Deng, Z. Mao, H. Xu, L. Zhang, Y. Zhong, X. Sui, "Synthesis fibrous LaFeO<sub>3</sub> perovskite oxide Adsorpt. Rhodamine B," Ecotoxicol. Environ. Safety, vol. 168, no. March 2018, pp. 35–44, 2019.
- [87] "Nonstoichiometry and Defect Structure of the Perovskite-Type Oxides La<sub>1-x</sub>Sr<sub>x</sub>FeO<sub>3-δ</sub>," J. Mizusaki, M. Yoshihiro, S. Yamauchi, K. Fueki. Nonstoichiom. Defect Struct. Perovskite-Type Oxides La<sub>1-x</sub>Sr<sub>x</sub>FeO<sub>3-δ</sub>. Jour\_nal Solid State Chem. 58(2)257–266, 1985.
- [88] "Electronic Properties of Ca Doped LaFeO<sub>3</sub>: A First-Principles Study," R. Pushpa, D. Daniel, D. P. Butt. Electron. Prop. Ca Doped LaFeO<sub>3</sub> A First-Principles Study. Solid State Ionics, 249-250184–190, 2013.
- [89] "The Effect of Ca, Sr, and Ba Doping on the Ionic Conductivity and Cathode Performance of LaFeO<sub>3</sub>," F. Bidrawn, S. Lee, J. M. Vohs, R. J. Gorte. Eff. Ca, Sr, Ba Doping Ion. Conduct. Cathode Perform. LaFeO<sub>3</sub>. J. Electrochem. Soc. 155(7)B660–B665, 2008.
- [90] N. Ortiz-Vitoriano, I. Ruiz Larramendi, I. Gil Muro, J. I. Ruiz Larramendi, T. Rojo. Nanoparticles La<sub>0.8</sub>Ca<sub>0.2</sub>Fe<sub>0.8</sub>Ni<sub>0.2</sub>O<sub>3-δ</sub> Perovskite Solid Oxide Fuel Cell Appl. Mater. Res. Bull. 45(10)1513–1519, 2010.
- [91] M.-H. Hung, M. V. M. Rao, D.-S. Tsai. Microstruct. Electri\_cal Prop. Calcium Substituted LaFeO<sub>3</sub> as SOFC Cathode. Mater. Chem. Physics, 101(2-3)297–302, 2007.
- [92] J. Mizusaki, M. Yoshihiro, S. Yamauchi, K. Fueki. Nonstoichiom. Defect Struct. Perovskite-Type Oxides

---

$\text{La}_{1-x}\text{Sr}_x\text{FeO}_{3-\delta}$ . *Journal Solid State Chem.* 58(2)257–266, 1985.

- [93] M. Idrees, M. Nadeem, M. Mehmood, M. Atif, K. H. Chae, M. M. Hassan. *Impedance Spectrosc. Investig. Delocalization Eff. Disord. Induc. by Ni Doping LaFeO<sub>3</sub>*. *J. Phys. D Appl. Physics*, 44(10)105401, 2011.
- [94] H.-C. Wang, C.-L. Wang, J.-L. Zhang, M.-L. Zhao, J. Liu, W.-B. Su, N. Yin, L.-M. Mei. *Cu Doping Eff. Electr. Resist. Seebeck Coeff. Perovskite-Type LaFeO<sub>3</sub> Ceram.* *Chinese Phys. Lett.* 26(10)107301, 2009.
- [95] W. Haron, T. Thawechai, W. Wattanathana, A. Laobuthee, H. Manaspiya, C. Veranitisagul, N. Koonsaeng. *Struct. Charact. Dielectric Prop. La<sub>1-x</sub>CoxFeO<sub>3</sub> LaFe<sub>1-x</sub>CoxO<sub>3</sub> Synth. via Met. Org. Complexes.* *Energy Procedia*, 347.
- [96] K. Zhao, F. He, Z. Huang, G. Wei, A. Zheng, H. Li, Z. Zhao. *Perovskite Type Oxides LaFe<sub>1-x</sub>CoxO<sub>3</sub> Chem. Looping Steam Methane Reforming to Syngas Hydrog. Co-Production.* *Appl. Energy*, 168193–203, 2016.
- [97] N. Lakshminarayanan, J. N. Kuhn, S. A. Rykov, J. M. M. Millet, U. S. Ozkan. *Catal. Today*, 157446–450, 2010.
- [98] Y.-P. Fu, A. Subardi, M.-Y. Hsieh, W.-K. Chang. *Electrochem. Properties La<sub>0.5</sub>Sr<sub>0.5</sub>Co<sub>0.8</sub>M<sub>0.2</sub>O<sub>3-δ</sub> (M=Mn, Fe, Ni, Cu) Perovskite Cathodes IT-SOFCs.* *J. Am. Ceram. Soc.* 99(4)1345–1352, 2016.
- [99] G. W. Coffey, J. S. Hardy, L. R. Pederson, P. C. Rieke, E. C. Thomsen. *Oxyg. Reduct. Act. Lanthanum Strontium Nickel Ferrite.* *Electrochemical Solid State Lett.* 6(6)A121–A124, 2003.
- [100] I. N. Sora, T. Caronna, F. Fontana, C. D. J. Fernandez, A. Caneschi, M. Green. *Cryst. Struct. Magn. Prop. Strontium Copp. Doped Lanthanum Ferrites.* *J. Solid State Chem.* 19133–39, 2012.
- [101] M. Qin, Z. Lan, Y. Ding, K. Xiong, Y. Liu, J. Guo, A study *Hydrog. storage Electrochem. Prop. La<sub>0.55</sub>Pr<sub>0.05</sub>Nd<sub>0.15</sub>Mg<sub>0.25</sub>Ni<sub>3.5</sub>(Co<sub>0.5</sub>Al<sub>0.5</sub>)<sub>x</sub> (x=0.0, 0.1, 0.3, 0.5) Alloy.* *J. Rare Earth.* 30 222.
- [102] A. Belotti, J. Liu, A. Curcio, J. Wang, Z. Wang, E. Quattrocchi, M.B. Effat, F. Ciucci, *Introd. Ag Ba<sub>0.9</sub>La<sub>0.1</sub>FeO<sub>3-δ</sub> Comb. cationic Substit. with Met. Part. Decor.* doi 10.1016/j.matre.2021.100018.
- [103] W. Haron, A. Wisitsoraat, U. Sirimahachai, S. Wongnawa, *A simple Synth. Charact. LaMO<sub>3</sub> (M=Al Co, Fe, Gd) perovskites via Chem. co-precipitation method,* *Songklanakarin, J. Sci. Technol.* 40 484–491.
- [104] H. T. H. Mo, H. Nan, X. Lang, S. Liu, L. Qiao, X. Hu, “*Influence of calcium doping on performance of LaMnO<sub>3</sub> supercapacitors,*” H. Mo, H. Nan, X. Lang, S. Liu, L. Qiao, X. Hu, H. Tian, *Influ. calcium doping Perform. LaMnO<sub>3</sub> supercapacitors,* *Cera. Int.* 44 9733–9741.
- [105] G.R. Monama et a, “*No Tit[1] G.R. Monama e.*”
- [106] T.A. Nguyen, V. Pham, T.L. Pham, L.T.Tr. Nguyen, I.Ya Mittova, V.O. Mittova, L. N. Vo, B.T.T. Nguyen, V.X. Bui, E.L. Viryutina, *Simple Synth. NdFeO<sub>3</sub> nanoparticles by co-precipitation method based a study Therm. Behav. Fe N.*
- [107] M. Fang, X. Yao, W. Li, Y. Li, M. Shui, J. Shu, *Investig. lithium doping perovskite oxide LiMnO<sub>3</sub> as possible LIB anode Mater. Ceram. Int.* 44 8223–8231.
- [108] J. A. Gomez-Cuaspud, E. Vera-Lopez, E. Barrachina, J. B. Carda-Castello, *One step hydrothermal Synth. LaFeO<sub>3</sub> perovskite,* *Av. Cent. del Norte* 39-115.
- [109] Z. Zhou, L.i. Guo, H. Yang, Q. Liu, F. Ye, *Hydrothermal Synth. Magn. Prop. multiferroic rare-earth orthoferrites,* *J. Alloy. Comp.* 583 21–31.

- 
- [110] S. M. Selbach, J. R. Tolchard, A. Fossdal, and T. Grande, "Non-linear thermal evolution of the crystal structure and phase transitions of LaFeO<sub>3</sub> investigated by high temperature X-ray diffraction," *J. Solid State Chem.*, vol. 196, pp. 249–254, 2012.
- [111] M. Idrees, M. Nadeem, and M. Shah, "Anomalous octahedral distortions in LaFe<sub>1-x</sub>Ni<sub>x</sub>O<sub>3</sub>," no. May 2014, 2011.
- [112] C. A. L. Dixon, C. M. Kavanagh, K. S. Knight, W. Kockelmann, D. Finlay, and P. Lightfoot, "Author 's Accepted Manuscript," 2015.
- [113] A. Ritzmann, P. Northwest, A. B. Munoz-garcia, M. Pavone, and J. A. Keith, "Ab initio evaluation of oxygen diffusivity in LaFeO<sub>3</sub>," no. July, 2013.
- [114] K. Huang, H. Y. Lee, and J. B. Goodenough, "Sr - and Ni - Doped LaCoO<sub>3</sub> and LaFeO<sub>3</sub> Perovskites : New Cathode Materials for Solid - Oxide Fuel Cells vol. 145, 1998.
- [115] K. R. N. et Al., "No Title."
- [116] "No Title."
- [117] M. J. Kirshenbaum, M. G. Boebinger, M. J. Katz, M. T. McDowell, and M. Dasog, "Solid-State Route for the Synthesis of Scalable, Luminescent Silicon and Germanium Nanocrystals," *ChemNanoMat*, vol. 4, no. 4, pp. 423–429, 2018.
- [118] Y. X. Gan, A. H. Jayatissa, Z. Yu, X. Chen, and M. Li, "Hydrothermal Synthesis of Nanomaterials," *J. Nanomater.*, vol. 2020, 2020.
- [119] B. L. Iverson and P. B. Dervan, "'No Covariance Structure Analysis of Health-Related Indicators for Homebound Elderly People Focusing on Subjective Health Feeling,'" pp. 7823–7830.
- [120] N. A. Shepelin, Z. P. Tehrani, N. Ohannessian, C. W. Schneider, D. Pergolesi, and T. Lippert, "A practical guide to pulsed laser deposition," *Chem. Soc. Rev.*, vol. 52, no. 7, pp. 2294–2321, 2023.
- [121] A. Reza Sadrolhosseini, M. Adzir Mahdi, F. Alizadeh, and S. Abdul Rashid, "Laser Ablation Technique for Synthesis of Metal Nanoparticle in Liquid," *Laser Technol. its Appl.*, 2019.
- [122] J. A. Bevis, "Applications of Thermal Analysis in Electrical Cable Manufacture," *Princ. Appl. Therm. Anal.*, pp. 164–189, 2008.
- [123] D. Scanning, "Differential Scanning Calorimetry ( DSC ) A Beginner ' s Guide."
- [124] A. K. Zak, W. H. A. Majid, M. E. Abrishami, and R. Youse, "X-ray analysis of ZnO nanoparticles by Williamson e Hall and size e strain plot methods," vol. 13, 2011.
- [125] L. Alexander and H. P. Klug, "Determination of crystallite size with the x-ray spectrometer," *J. Appl. Phys.*, vol. 21, no. 2, pp. 137–142, 1950.
- [126] "( IR ) Theory and Interpretation of IR spectra presentation for additional examples of How do we know :  
 • how atoms are connected together ? • What functional groups exist in the • If we have a specific stereoisomer ? The field of organic structure determ," pp. 853–866.
- [127] "X-Ray Photoelectron Spectroscopy ( XPS )," 2018.
- [128] J. V. Sci, F. A. Stevie, and C. L. Donley, "Introduction to x-ray photoelectron spectroscopy Introduction to x-ray photoelectron spectroscopy," vol. 063204, no. September, 2020.
-

- 
- [129] E. Capabilities, "X-Ray Photoelectron Spectroscopy Applications," pp. 716–724, 2017.
- [130] "Principles of SEM Contents."
- [131] N. Asadi, "Scanning Electron Microscopy ( SEM ) Comparison to Optical Microscopy."
- [132] Hp, "Module II Energy Dispersive Analysis of X-Rays ( Edax )," *A Gatew. to All Post Grad. Courses*, pp. 1–11, 2018.
- [133] F. Applicata, N. Carrara, and V. Madonna, "UV-Vis spectroscopy Abstract :", pp. 1–14, 2018.
- [134] M. R. Sharpe, "Instrumentation: Stray Light in UV-VIS Spectrophotometers," *Anal. Chem.*, vol. 56, no. 2, pp. 339A-356A, 1984.
- [135] J. Franck, M. Wisztorski, J. Stauber, M. Elayed, and M. Salzet, "Application Note," pp. 42–46, 2007.
- [136] H. Zhong, F. Pan, S. Yue, C. Qin, V. Hadjiev, and F. Tian, "Idealizing Tauc Plot for Accurate Bandgap Determination of Semiconductor with UV-Vis : A Case Study for Cubic Boron Arsenide," pp. 1–15.
- [137] V. Pawar, M. Kumar, P. K. Dubey, M. K. Singh, A. S. K. Sinha, and P. Singh, "Influence of synthesis route on structural, optical, and electrical properties of TiO<sub>2</sub>," *Appl. Phys. A Mater. Sci. Process.*, vol. 125, no. 9, pp. 1–14, 2019.
- [138] X. Zhang et al., "Module Iv Transmission," *Plant Dis.*, vol. 77, no. 11, pp. 2240–2250, 2009.
- [139] T. E. Microscope, "No Title."
- [140] X. Yang, "Electron Microscopy Centre," pp. 0–4, 2006.
- [141] S. Nebaba, D. Zavyalov, and A. Pak, "Patterns Detection in SAED Images of Transmission," pp. 1–11.
- [142] O. Marti, B. Drake, P. K. Hansma, O. Marti, B. Drake, and P. K. Hansma, "Atomic force microscopy of liquidcovered surfaces : Atomic resolution images Atomic force microscopy of Uquid ", covered surfaces : Atomic resolution images," vol. 484, no. 1987, pp. 51–54, 2012.
- [143] A. F. Microscope, "9 physical review letters," vol. 56, no. 9, 1986.
- [144] J. C. Dyre, "The random free-energy barrier model for ac conduction in disordered solids," *J. Appl. Phys.*, vol. 64, no. 5, pp. 2456–2468, 1988.
- [145] S. R. Elliott, "A.c. conduction in amorphous chalcogenide and pnictide semiconductors," *Adv. Phys.*, vol. 36, no. 2, pp. 135–217, 1987.
- [146] R. Waser, "Bulk Conductivity and Defect Chemistry of Acceptor-Doped Strontium Titanate in the Quenched State," *J. Am. Ceram. Soc.*, vol. 74, no. 8, pp. 1934–1940, 1991.
- [147] Z. Cheng et al., "Ionic Conduction in Perovskite-type Oxide Solid Solution and its Application to the Solid Electrolyte Fuel Cell k ..\_\_," vol. 11, pp. 105–111, 1971.
- [148] S. Murugavel and B. Roling, "ac Conductivity Spectra of Alkali Tellurite Glasses : Composition-Dependent Deviations from the Summerfield Scaling," no. 2, pp. 8–11, 2002.
- [149] B. Roling, A. Happe, K. Funke, and M. D. Ingram, "Carrier Concentrations and Relaxation Spectroscopy : New Information from Scaling Properties of Conductivity Spectra in Ionically Conducting Glasses," 1997.

- 
- [150] D. L. Sidebottom, "Dimensionality Dependence of the Conductivity Dispersion in Ionic Materials," *Phys. Rev. Lett.*, vol. 99, pp. 983–986, 1999.
- [151] I. M. Hodge, M. D. Ingram, and A. R. West, "Impedance and modulus spectroscopy of polycrystalline solid electrolytes," *J. Electroanal. Chem.*, vol. 74, no. 2, pp. 125–143, 1976.
- [152] D. D. Macdonald, "Reflections on the history of electrochemical impedance spectroscopy," vol. 51, pp. 1376–1388, 2006.
- [153] S. E. E. Profile, "Impedance Spectroscopy for Emerging Photovoltaics Impedance Spectroscopy for Emerging Photovoltaics," no. March, 2019.
- [154] L. Zhang, Y. Pu, and M. Chen, "Complex impedance spectroscopy for capacitive energy-storage ceramics : a review and prospects," vol. 28, 2023.
- [155] B. J. T. S. Irvine, D. C. Sinclair, and A. R. West, "Electroceramics : Characterization by Impedance Spectroscopy," vol. 2, no. 3, pp. 132–138, 1990.
- [156] F. D. Morrison, D. J. Jung, and J. F. Scott, "Constant-phase-element (CPE) modeling of ferroelectric random-access memory lead zirconate-titanate (PZT) capacitors," *J. Appl. Phys.*, vol. 101, no. 9, 2007.
- [157] M. E. Orazem, N. Pébère, and B. Tribollet, "Enhanced Graphical Representation of Electrochemical Impedance Data," *J. Electrochem. Soc.*, vol. 153, no. 4, p. B129, 2006.
- [158] N. Elgrishi, K. J. Rountree, B. D. McCarthy, E. S. Rountree, T. T. Eisenhart, and J. L. Dempsey, "A Practical Beginner's Guide to Cyclic Voltammetry," *J. Chem. Educ.*, vol. 95, no. 2, pp. 197–206, 2018.
- [159] M. Corva, N. Blanc, C. J. Bondue, and K. Tschulik, "Differential Tafel Analysis: A Quick and Robust Tool to Inspect and Benchmark Charge Transfer in Electrocatalysis," *ACS Catal.*, vol. 12, no. 21, pp. 13805–13812, 2022.
- [160] P. Khadke, T. Tichter, T. Boettcher, F. Muench, W. Ensinger, and C. Roth, "A simple and effective method for the accurate extraction of kinetic parameters using differential Tafel plots," *Sci. Rep.*, vol. 11, no. 1, p. 8974, 2021.
- [161] R. A. Buchanan and E. E. Stansbury, "Electrochemical Corrosion," *Handb. Environ. Degrad. Mater. Second Ed.*, pp. 87–125, 2012.
- [162] J.-I. Jung, H. Y. Jeong, J.-S. Lee, M. G. Kim, and J. Cho, "A Bifunctional Perovskite Catalyst for Oxygen Reduction and Evolution," *Angew. Chemie*, vol. 126, no. 18, pp. 4670–4674, 2014.
- [163] W. T. Hernández and U. Valle, "Electromechanical Characterization of Mass Transport at Microelectrode Arrays," no. Figure 1, pp. 101–110, 2014.
- [164] L. B. Mccusker, R. B. Von Dreele, D. E. Cox, D. Louër, and P. Scardi, "Rietveld refinement guidelines," *J. Appl. Crystallogr.*, vol. 32, no. 1, pp. 36–50, 1999.
- [165] J. M. Chem, "Materials Chemistry A," pp. 15367–15379, 2017.
- [166] K. Tsuda and M. Tanaka, "Refinement of crystal structure parameters using convergent-beam electron diffraction: the low-temperature phase of SrTiO<sub>3</sub>," *Acta Crystallogr. Sect. A*, vol. 51, no. 1, pp. 7–19, 1995.
- [167] S. Singh, P. Singh, M. Viviani, and S. Presto, "Dy doped SrTiO<sub>3</sub>: A promising anodic material in solid oxide fuel cells," *Int. J. Hydrogen Energy*, vol. 43, no. 41, pp. 19242–19249, 2018.

- 
- [168] B. Yahmadi, N. Kamoun, C. Guasch, and R. Bennaceur, "Synthesis and characterization of nanocrystallized In<sub>2</sub>S<sub>3</sub> thin films via CBD technique," *Mater. Chem. Phys.*, vol. 127, no. 1–2, pp. 239–247, 2011.
- [169] P. Singh, P. K. Jha, A. S. K. Sinha, P. A. Jha, and P. Singh, "Ion dynamics of non-stoichiometric Na<sub>0.5+x</sub>Bi<sub>0.5-x</sub>TiO<sub>3-δ</sub>: A degradation study," *Solid State Ionics*, vol. 345, no. July 2019, 2020.
- [170] T. Shi, Y. Chen, and X. Guo, "Defect chemistry of alkaline earth metal (Sr/Ba) titanates," *Prog. Mater. Sci.*, vol. 80, pp. 77–132, 2016.
- [171] "© 1977 Nature Publishing Group," 1977.
- [172] P. Singh, P. K. Jha, P. A. Jha, and P. Singh, "Influence of sintering temperature on ion dynamics of Na<sub>0.5</sub>Bi<sub>0.5</sub>TiO<sub>3-δ</sub>: Suitability as an electrolyte material for SOFC," *Int. J. Hydrogen Energy*, vol. 45, no. 34, pp. 17006–17016, 2020.
- [173] M. W. Barsoum, "Fundamentals of ceramics," *Fundam. Ceram.*, pp. 1–612, 2002.
- [174] P. Singh, Raghvendra, O. Parkash, and D. Kumar, "Scaling of low-temperature conductivity spectra of BaSn<sub>1-x</sub>Nbx: Temperature and compositional-independent conductivity," *Phys. Rev. B - Condens. Matter Mater. Phys.*, vol. 84, no. 17, pp. 1–6, 2011.
- [175] S. Singh, P. A. Jha, S. Varma, and P. Singh, "Large polaron hopping phenomenon in lanthanum doped strontium titanate," *J. Alloys Compd.*, vol. 704, pp. 707–716, 2017.
- [176] B. Behera, P. Nayak, and R. N. P. Choudhary, "Impedance spectroscopy study of NaBa<sub>2</sub>V<sub>5</sub>O<sub>15</sub> ceramic," *J. Alloys Compd.*, vol. 436, no. 1–2, pp. 226–232, 2007.
- [177] I. Ahmad, M. J. Akhtar, M. Younas, M. Siddique, and M. M. Hasan, "Small polaronic hole hopping mechanism and Maxwell-Wagner relaxation in NdFeO<sub>3</sub>," *J. Appl. Phys.*, vol. 112, no. 7, 2012.
- [178] O. N. Verma, N. K. Singh, Raghvendra, and P. Singh, "Study of ion dynamics in lanthanum aluminate probed by conductivity spectroscopy," *RSC Adv.*, vol. 5, no. 28, pp. 21614–21619, 2015.
- [179] A. Ghosh and A. Pan, "Scaling of the conductivity spectra in ionic glasses: Dependence on the structure," *Phys. Rev. Lett.*, vol. 84, no. 10, pp. 2188–2190, 2000.
- [180] H. Mahamoud, B. Louati, F. Hlel, and K. Guidara, "0 · 6 ) 2," vol. 34, no. 5, pp. 1069–1075, 2011.
- [181] S. K. Barik, S. Ahmed, and S. Hajra, "Studies of dielectric relaxation and impedance analysis of new electronic material: (Sb<sup>1/2</sup> Na<sup>1/2</sup>)(Fe<sup>2/3</sup> Mo<sup>1/3</sup>)O<sub>3</sub>," *Appl. Phys. A Mater. Sci. Process.*, vol. 125, no. 3, pp. 1–8, 2019.
- [182] T. Mondal et al., "A comparative study on electrical conduction properties of Sr-substituted Ba<sub>1-x</sub>Sr<sub>x</sub>Zr<sub>0.1</sub>Ti<sub>0.9</sub>O<sub>3</sub> (x = 0.00–0.15) ceramics," *Ionics (Kiel)*, vol. 23, no. 9, pp. 2405–2416, 2017.
- [183] J. T. Mefford, W. G. Hardin, S. Dai, K. P. Johnston, and K. J. Stevenson, "Anion charge storage through oxygen intercalation in LaMnO<sub>3</sub> perovskite pseudocapacitor electrodes," *Nat. Mater.*, vol. 13, no. 7, pp. 726–732, 2014.
- [184] D. Guo et al., "Facile synthesis and excellent electrochemical properties of CoMoO<sub>4</sub> nanoplate arrays as supercapacitors," *J. Mater. Chem. A*, vol. 1, no. 24, pp. 7247–7254, 2013.
- [185] E. M. Espinoza, J. A. Clark, J. Soliman, J. B. Derr, M. Morales, and V. I. Vullev, "Practical Aspects of Cyclic Voltammetry: How to Estimate Reduction Potentials When Irreversibility Prevails," *J.*

---

*Electrochem. Soc.*, vol. 166, no. 5, pp. H3175–H3187, 2019.

- [186] Y. Ge, X. Xie, J. Roscher, R. Holze, and Q. Qu, “How to measure and report the capacity of electrochemical double layers, supercapacitors, and their electrode materials,” *J. Solid State Electrochem.*, vol. 24, no. 11–12, pp. 3215–3230, 2020.
- [187] J. Wang, J. Polleux, J. Lim, and B. Dunn, “Pseudocapacitive contributions to electrochemical energy storage in TiO<sub>2</sub> (anatase) nanoparticles,” *J. Phys. Chem. C*, vol. 111, no. 40, pp. 14925–14931, 2007.
- [188] A. S. Bangwal et al., “Compositional effect on oxygen reduction reaction in Pr excess double perovskite Pr<sub>1+x</sub>Ba<sub>1-x</sub>Co<sub>2</sub>O<sub>6-δ</sub> cathode materials,” *Int. J. Hydrogen Energy*, vol. 45, no. 43, pp. 23378–23390, 2020.
- [189] O. J. Guy and K. A. D. Walker, *Graphene Functionalization for Biosensor Applications*, Second Edi. Elsevier Inc., 2016.
- [190] J. T. Last, “Infrared-absorption studies on barium titanate and related materials,” *Phys. Rev.*, vol. 105, no. 6, pp. 1740–1750, 1957.
- [191] A. Periasamy, S. Muruganand, and M. Palaniswamy, “Vibrational studies of Na<sub>2</sub>SO<sub>4</sub>, K<sub>2</sub>SO<sub>4</sub>, NaHSO<sub>4</sub> and KHSO<sub>4</sub> crystals,” *Rasayan J. Chem.*, vol. 2, no. 4, pp. 981–989, 2009.
- [192] R. G. Snyder, J. Kumamoto, and J. A. Ibers, “Vibrational spectrum of crystalline potassium hydroxide,” *J. Chem. Phys.*, vol. 33, no. 4, pp. 1171–1177, 1960.
- [193] M. Chauhan, P. K. Jha, P. A. Jha, and P. Singh, “Influence of the crystalline phase on the electrocatalytic behaviour of Sm<sub>2-x</sub>Sr<sub>x</sub>NiO<sub>4-δ</sub> (x = 0.4 to 1.0) Ruddlesden-Popper-based systems: A comparative study of bulk and thin electrocatalysts,” *Phys. Chem. Chem. Phys.*, vol. 24, no. 9, pp. 5330–5342, 2022.
- [194] S. Jain, J. Shah, N. S. Negi, C. Sharma, and R. K. Kotnala, “Significance of interface barrier at electrode of hematite hydroelectric cell for generating ecopower by water splitting,” *Int. J. Energy Res.*, vol. 43, no. 9, pp. 4743–4755, 2019.
- [195] J. C. Dupin, D. Gonbeau, P. Vinatier, and A. Levasseur, “Systematic XPS studies of metal oxides, hydroxides and peroxides,” *Phys. Chem. Chem. Phys.*, vol. 2, no. 6, pp. 1319–1324, 2000.
- [196] L. Yu, N. Xu, T. Zhu, Z. Xu, M. Sun, and D. Geng, “La<sub>0.4</sub>Sr<sub>0.6</sub>Co<sub>0.7</sub>Fe<sub>0.2</sub>Nb<sub>0.1</sub>O<sub>3-δ</sub> perovskite prepared by the sol-gel method with superior performance as a bifunctional oxygen electrocatalyst,” *Int. J. Hydrogen Energy*, vol. 45, no. 55, pp. 30583–30591, 2020.
- [197] M. Chauhan, A. S. Bangwal, and P. Singh, “Electrochemical performance of A-site substituted SmSrNiO<sub>4-δ</sub> for energy storage applications,” *Int. J. Hydrogen Energy*, vol. 48, no. 14, pp. 5518–5528, 2023.
- [198] V. V. Atuchin, V. G. Kesler, N. V. Pervukhina, and Z. Zhang, “Ti 2p and O 1s core levels and chemical bonding in titanium-bearing oxides,” *J. Electron Spectros. Relat. Phenomena*, vol. 152, no. 1–2, pp. 18–24, 2006.
- [199] R. D. Leapman, L. A. Grunes, and P. L. Fejes, “Study of the L<sub>23</sub> edges in the 3d transition metals and their oxides by electron-energy-loss spectroscopy with comparisons to theory,” *Phys. Rev. B*, vol. 26, no. 2, pp. 614–635, 1982.
- [200] L. Leonat, G. Sb, I. V. Br, and I. Cyclic, “CYCLIC VOLTAMMETRY FOR ENERGY LEVELS ESTIMATION OF ORGANIC MATERIALS,” vol. 75, 2013.

- 
- [201] K. Veerappan, "Bandgap and electrochemical engineering for disordered LaFeO<sub>3</sub>."
- [202] J. Hwang, R. R. Rao, L. Giordano, Y. Katayama, Y. Yu, and Y. Shao-horn, "Perovskites in catalysis and electrocatalysis," vol. 756, no. November, pp. 751–756, 2017.
- [203] Q. Xu, G. Li, Y. Zhang, Q. Yang, Y. Sun, and C. Felser, "Descriptor for Hydrogen Evolution Catalysts Based on the Bulk Band Structure Effect," *ACS Catal.*, vol. 10, no. 9, pp. 5042–5048, 2020.
- [204] T. W. Napporn et al., *Electrochemical measurement methods and characterization on the cell level*. 2018.
- [205] P. Homo and C. E. Poli, "Determination of HOMO and LUMO of [ 6 , 6 ] -Phenyl C61-butyric Acid through Voltametry Characterization," vol. 40, no. 2, pp. 173–176, 2011.
- [206] N. Loew et al., "Electrochimica Acta Cyclic voltammetry and electrochemical impedance simulations of the mediator-type enzyme electrode reaction using finite element method," *Electrochim. Acta*, vol. 367, p. 137483, 2021.
- [207] "164377a0.pdf."
- [208] D. Antipin, M. Risch, N. Gestaltung, and H. Zentrum, "No Title."
- [209] I. Riess, "Mixed ionic – electronic conductors — material properties and applications," vol. 157, pp. 1–17, 2003.
- [210] M. Guo, M. Ji, and W. Cui, "Theoretical investigation of HER/OER/ORR catalytic activity of single atom-decorated graphyne by DFT and comparative DOS analyses," *Appl. Surf. Sci.*, vol. 592, no. January, 2022.
- [211] P. Singh, R. Pandey, T. Miruszewski, K. Dzierzowski, A. Mielewczyk-Gryn, and P. Singh, "Signature of Oxide-Ion Conduction in Alkaline-Earth-Metal-Doped Y<sub>3</sub>GaO<sub>6</sub>," *ACS Omega*, vol. 5, no. 47, pp. 30395–30404, 2020.
- [212] M. Basak, M. L. Rahman, M. F. Ahmed, B. Biswas, and N. Sharmin, "The use of X-ray diffraction peak profile analysis to determine the structural parameters of cobalt ferrite nanoparticles using Debye-Scherrer, Williamson-Hall, Halder-Wagner and Size-strain plot: Different precipitating agent approach," *J. Alloys Compd.*, vol. 895, p. 162694, 2022.
- [213] A. Khorsand Zak, W. H. Abd. Majid, M. E. Abrishami, and R. Yousefi, "X-ray analysis of ZnO nanoparticles by Williamson-Hall and size-strain plot methods," *Solid State Sci.*, vol. 13, no. 1, pp. 251–256, 2011.
- [214] Y. Zhao and J. Zhang, "Microstrain and grain-size analysis from diffraction peak width and graphical derivation of high-pressure thermomechanics," *J. Appl. Crystallogr.*, vol. 41, no. 6, pp. 1095–1108, 2008.
- [215] N. Sharma, H. S. Kushwaha, S. K. Sharma, and K. Sachdev, "Fabrication of LaFeO<sub>3</sub> and rGO-LaFeO<sub>3</sub> microspheres based gas sensors for detection of NO<sub>2</sub> and CO," *RSC Adv.*, vol. 10, no. 3, pp. 1297–1308, 2020.
- [216] Y. Liu, Z. Lockman, A. Aziz, and J. MacManus-Driscoll, "Size dependent ferromagnetism in cerium oxide (CeO<sub>2</sub>) nanostructures independent of oxygen vacancies," *J. Phys. Condens. Matter*, vol. 20, no. 16, 2008.
- [217] J. Jin et al., "Stable GQD@PANi nanocomposites based on benzenoid structure for enhanced specific capacitance," *Int. J. Hydrogen Energy*, vol. 43, no. 17, pp. 8426–8439, 2018.