

1.4 References

1. Chu, S. and Majumdar, A., 2012. Opportunities and challenges for a sustainable energy future. *nature*, 488(7411), pp.294-303.
2. Lee, R., 2011. The outlook for population growth. *Science*, 333(6042), pp.569-573.
3. Cook, T.R., Dogutan, D.K., Reece, S.Y., Surendranath, Y., Teets, T.S. and Nocera, D.G., 2010. Solar energy supply and storage for the legacy and nonlegacy worlds. *Chemical reviews*, 110(11), pp.6474-6502.
4. Global energy trends: Insights from the 2023 statistical review of world energy.
5. Ritchie, H., Roser, M. and Rosado, P., 2020. CO₂ and greenhouse gas emissions. *Our world in data*.
6. Agbossou, K., Chahine, R., Hamelin, J., Laurencelle, F., Anouar, A., St-Arnaud, J.M. and Bose, T.K., 2001. Renewable energy systems based on hydrogen for remote applications. *Journal of power sources*, 96(1), pp.168-172.
7. Vosen, S.R. and Keller, J.O., 1999. Hybrid energy storage systems for stand-alone electric power systems: optimization of system performance and cost through control strategies. *International journal of hydrogen energy*, 24(12), pp.1139-1156.
8. Menzl, F., Wenske, M. and Lehmann, J., 1998. Hydrogen-Production by a Windmill Powered Electrolyser. *HYDROGEN ENERGY PROGRESS*, 1, pp.757-766.
9. Bose, T.K., Agbossou, K., Benard, P. and St-Arnaud, J.M., 1999. New perspectives on renewable energy systems based on hydrogen.
10. Barthels, H., Brocke, W.A., Bonhoff, K., Groehn, H.G., Heuts, G., Lennartz, M., Mai, H., Mergel, J., Schmid, L. and Ritzenhoff, P., 1998. Phoebus-Jülich: an autonomous energy supply system comprising photovoltaics, electrolytic hydrogen, fuel cell. *International Journal of Hydrogen Energy*, 23(4), pp.295-301.
11. Turner, J.A., 2004. Sustainable hydrogen production. *Science*, 305(5686), pp.972-974.
12. Megía, P.J., Vizcaíno, A.J., Calles, J.A. and Carrero, A., 2021. Hydrogen production technologies: From fossil fuels toward renewable sources. A mini review. *Energy & Fuels*, 35(20), pp.16403-16415.
13. Balat, M., 2008. Potential importance of hydrogen as a future solution to environmental and transportation problems. *International journal of hydrogen energy*, 33(15), pp.4013-4029.

14. Pilavachi, P.A., Chatzipanagi, A.I. and Spyropoulou, A.I., 2009. Evaluation of hydrogen production methods using the analytic hierarchy process. *International Journal of hydrogen energy*, 34(13), pp.5294-5303.
15. Turner, J.A., 1999. A realizable renewable energy future. *Science*, 285(5428), pp.687-689.
16. Turner, J.A., 2004. Sustainable hydrogen production. *Science (New York, NY)*, 305, pp.972-974.
17. Whiteley, K.S., Heggs, T.G., Koch, H., Mawer, R.L., and Immel, W., 2000. Polyolefins. in Ullmann's encyclopedia of industrial chemistry (Wiley). pp. 1-103.
18. Bailera, M., Kezibri, N., Romeo, L.M., Espatolero, S., Lisbona, P. and Bouallou, C., 2017. Future applications of hydrogen production and CO₂ utilization for energy storage: Hybrid Power to Gas-Oxycombustion power plants. *International Journal of Hydrogen Energy*, 42(19), pp.13625-13632.
19. Sapountzi, F.M., Gracia, J.M., Fredriksson, H.O. and Niemantsverdriet, J.H., 2017. Electrocatalysts for the generation of hydrogen, oxygen and synthesis gas. *Progress in Energy and Combustion Science*, 58, pp.1-35.
20. Rand, D.A., 2011. A journey on the electrochemical road to sustainability. *Journal of Solid State Electrochemistry*, 15(7-8), pp.1579-1622.
21. Brauns, J. and Turek, T., 2020. Alkaline water electrolysis powered by renewable energy: A review. *Processes*, 8(2), p.248.
22. Wang, Z.L., Xu, D., Xu, J.J. and Zhang, X.B., 2014. Oxygen electrocatalysts in metal–air batteries: from aqueous to nonaqueous electrolytes. *Chemical Society Reviews*, 43(22), pp.7746-7786.
23. Mahmood, N., Tahir, M., Mahmood, A., Zhu, J., Cao, C. and Hou, Y., 2015. Chlorine-doped carbonated cobalt hydroxide for supercapacitors with enormously high pseudocapacitive performance and energy density. *Nano Energy*, 11, pp.267-276.
24. Tahir, M., Mahmood, N., Zhu, J., Mahmood, A., Butt, F.K., Rizwan, S., Aslam, I., Tanveer, M., Idrees, F., Shakir, I. and Cao, C., 2015. One dimensional graphitic carbon nitrides as effective metal-free oxygen reduction catalysts. *Scientific reports*, 5(1), p.12389.
25. Gür, T.M., 2018. Review of electrical energy storage technologies, materials and systems: challenges and prospects for large-scale grid storage. *Energy & Environmental Science*, 11(10), pp.2696-2767.
26. Seh, Z.W., Kibsgaard, J., Dickens, C.F., Chorkendorff, I.B., Nørskov, J.K. and Jaramillo, T.F., 2017. Combining theory and experiment in electrocatalysis: Insights into materials design. *Science*, 355(6321), p.eaad4998.

27. Huang, Z.F., Song, J., Li, K., Tahir, M., Wang, Y.T., Pan, L., Wang, L., Zhang, X. and Zou, J.J., 2016. Hollow cobalt-based bimetallic sulfide polyhedra for efficient all-pH-value electrochemical and photocatalytic hydrogen evolution. *Journal of the American Chemical Society*, 138(4), pp.1359-1365.
28. Meng, C., Ling, T., Ma, T.Y., Wang, H., Hu, Z., Zhou, Y., Mao, J., Du, X.W., Jaroniec, M. and Qiao, S.Z., 2017. Atomically and electronically coupled Pt and CoO hybrid nanocatalysts for enhanced electrocatalytic performance. *Advanced Materials*, 29(9), p.1604607.
29. Lv, Z., Mahmood, N., Tahir, M., Pan, L., Zhang, X. and Zou, J.J., 2016. Fabrication of zero to three dimensional nanostructured molybdenum sulfides and their electrochemical and photocatalytic applications. *Nanoscale*, 8(43), pp.18250-18269.
30. Walter, M.G., Warren, E.L., McKone, J.R., Boettcher, S.W., Mi, Q., Santori, E.A. and Lewis, N.S., 2010. Solar water splitting cells. *Chemical reviews*, 110(11), pp.6446-6473.
31. Suen, N.T., Hung, S.F., Quan, Q., Zhang, N., Xu, Y.J. and Chen, H.M., 2017. Electrocatalysis for the oxygen evolution reaction: recent development and future perspectives. *Chemical Society Reviews*, 46(2), pp.337-365.
32. Ferreira, K.N., Iverson, T.M., Maghlaoui, K., Barber, J. and Iwata, S., 2004. Architecture of the photosynthetic oxygen-evolving center. *Science*, 303(5665), pp.1831-1838.
33. Kanan, M.W. and Nocera, D.G., 2008. In situ formation of an oxygen-evolving catalyst in neutral water containing phosphate and Co^{2+} . *Science*, 321(5892), pp.1072-1075.
34. Mallouk, T.E., 2013. Water electrolysis: Divide and conquer, *Nat. Chem.* 5, pp.362–363.
35. Shih, A.J., Monteiro, M.C., Dattila, F., Pavesi, D., Philips, M., da Silva, A.H., Vos, R.E., Ojha, K., Park, S., van der Heijden, O. and Marcandalli, G., 2022. Water electrolysis. *Nature Reviews Methods Primers*, 2(1), p.84.
36. Yu, M., Budiyanoto, E. and Tüysüz, H., 2022. Principles of water electrolysis and recent progress in cobalt-, nickel-, and iron-based oxides for the oxygen evolution reaction. *Angewandte Chemie International Edition*, 61(1), p.e202103824.
37. Khan, M.A., Zhao, H., Zou, W., Chen, Z., Cao, W., Fang, J., Xu, J., Zhang, L. and Zhang, J., 2018. Recent progresses in electrocatalysts for water electrolysis. *Electrochemical Energy Reviews*, 1, pp.483-530.
38. Nuttall, L.J., Fickett, A.P. and Titterton, W.A., 1975. Hydrogen generation by solid polymer electrolyte water electrolysis. *Hydrogen Energy: Part A*, pp.441-455.

39. Grewe, T., Deng, X., Weidenthaler, C., Schüth, F. and Tüysüz, H., 2013. Design of ordered mesoporous composite materials and their electrocatalytic activities for water oxidation. *Chemistry of Materials*, 25(24), pp.4926-4935.
40. Zha, Y., Disabb-Miller, M.L., Johnson, Z.D., Hickner, M.A. and Tew, G.N., 2012. Metal-ation-based anion exchange membranes. *Journal of the American Chemical Society*, 134(10), pp.4493-4496.
41. Beainy, A., Karami, N. and Moubayed, N., 2014, November. Simulink model for a PEM electrolyzer based on an equivalent electrical circuit. In *International Conference on Renewable Energies for Developing Countries 2014* (pp. 145-149). Ieee.
42. Wei, C., Rao, R.R., Peng, J., Huang, B., Stephens, I.E., Risch, M., Xu, Z.J. and Shao-Horn, Y., 2019. Recommended practices and benchmark activity for hydrogen and oxygen electrocatalysis in water splitting and fuel cells. *Advanced Materials*, 31(31), p.1806296.
43. Wang, J., Cui, W., Liu, Q., Xing, Z., Asiri, A.M. and Sun, X., 2016. Recent progress in cobalt-based heterogeneous catalysts for electrochemical water splitting. *Advanced materials*, 28(2), pp.215-230.
44. Lee, Y., Suntivich, J., May, K.J., Perry, E.E. and Shao-Horn, Y., 2012. Synthesis and activities of rutile IrO₂ and RuO₂ nanoparticles for oxygen evolution in acid and alkaline solutions. *The journal of physical chemistry letters*, 3(3), pp.399-404.
45. Montoya, J.H., Seitz, L.C., Chakthranont, P., Vojvodic, A., Jaramillo, T.F. and Nørskov, J.K., 2017. Materials for solar fuels and chemicals. *Nature materials*, 16(1), pp.70-81.
46. Hunter, B.M., Gray, H.B. and Muller, A.M., 2016. Earth-abundant heterogeneous water oxidation catalysts. *Chemical reviews*, 116(22), pp.14120-14136.
47. Liu, Q., Chang, Z., Li, Z. and Zhang, X., 2018. Flexible metal–air batteries: Progress, challenges, and perspectives. *Small Methods*, 2(2), p.1700231.
48. Durmus, Y.E., Zhang, H., Baakes, F., Desmaizieres, G., Hayun, H., Yang, L., Kolek, M., Küpers, V., Janek, J., Mandler, D. and Passerini, S., 2020. Side by side battery technologies with lithium-ion based batteries. *Advanced energy materials*, 10(24), p.2000089.
49. Wang, C., Yu, Y., Niu, J., Liu, Y., Bridges, D., Liu, X., Pooran, J., Zhang, Y. and Hu, A., 2019. Recent progress of metal–air batteries—a mini review. *Applied Sciences*, 9(14), p.2787.
50. Wang, Y.J., Fang, B., Zhang, D., Li, A., Wilkinson, D.P., Ignaszak, A., Zhang, L. and Zhang, J., 2018. A review of carbon-composited materials as air-electrode bifunctional electrocatalysts for metal–air batteries. *Electrochemical Energy Reviews*, 1, pp.1-34.

51. Zhang, X.B. ed., 2018. *Metal-Air Batteries: Fundamentals and Applications*. John Wiley & Sons.
52. Wang, H.F. and Xu, Q., 2019. Materials design for rechargeable metal-air batteries. *Matter*, 1(3), pp.565-595.
53. Li, Y. and Lu, J., 2017. Metal–air batteries: will they be the future electrochemical energy storage device of choice?. *ACS Energy Letters*, 2(6), pp.1370-1377.
54. Kraytsberg, A. and Ein-Eli, Y., 2011. Review on Li–air batteries—Opportunities, limitations and perspective. *Journal of Power Sources*, 196(3), pp.886-893.
55. Ma, C., Xu, N., Qiao, J., Jian, S. and Zhang, J., 2016. Facile synthesis of NiCo₂O₄ nanosphere-carbon nanotubes hybrid as an efficient bifunctional electrocatalyst for rechargeable Zn–air batteries. *International Journal of Hydrogen Energy*, 41(21), pp.9211-9218.
56. Li, Y. and Dai, H., 2014. Recent advances in zinc–air batteries. *Chemical Society Reviews*, 43(15), pp.5257-5275.
57. Grande, L., Paillard, E., Hassoun, J., Park, J.B., Lee, Y.J., Sun, Y.K., Passerini, S. and Scrosati, B., 2015. The lithium/air battery: still an emerging system or a practical reality?. *Advanced materials*, 27(5), pp.784-800.
58. Liu, X., Yuan, Y., Liu, J., Liu, B., Chen, X., Ding, J., Han, X., Deng, Y., Zhong, C. and Hu, W., 2019. Utilizing solar energy to improve the oxygen evolution reaction kinetics in zinc–air battery. *Nature communications*, 10(1), p.4767.
59. Lu, J., Li, L., Park, J.B., Sun, Y.K., Wu, F. and Amine, K., 2014. Aprotic and aqueous Li–O₂ batteries. *Chemical reviews*, 114(11), pp.5611-5640.
60. Shao, Y., Ding, F., Xiao, J., Zhang, J., Xu, W., Park, S., Zhang, J.G., Wang, Y. and Liu, J., 2013. Making Li-air batteries rechargeable: Material challenges. *Advanced Functional Materials*, 23(8), pp.987-1004.
61. Zhang, T., Imanishi, N., Shimonishi, Y., Hirano, A., Takeda, Y., Yamamoto, O. and Sammes, N., 2010. A novel high energy density rechargeable lithium/air battery. *Chemical Communications*, 46(10), pp.1661-1663.
62. Cheng, F. and Chen, J., 2012. Metal–air batteries: from oxygen reduction electrochemistry to cathode catalysts. *Chemical Society Reviews*, 41(6), pp.2172-2192.
63. Fan, L., Tu, Z. and Chan, S.H., 2021. Recent development of hydrogen and fuel cell technologies: A review. *Energy Reports*, 7, pp.8421-8446.
64. Sharaf, O.Z. and Orhan, M.F., 2014. An overview of fuel cell technology: Fundamentals and applications. *Renewable and sustainable energy reviews*, 32, pp.810-853.

65. Chen, T.W., Anushya, G., Chen, S.M., Kalimuthu, P., Mariyappan, V., Gajendran, P. and Ramachandran, R., 2022. Recent advances in nanoscale based electrocatalysts for metal-air battery, fuel cell and water-splitting applications: An overview. *Materials*, 15(2), p.458.
66. Ananthachar, V. and Duffy, J.J., 2005. Efficiencies of hydrogen storage systems onboard fuel cell vehicles. *Solar Energy*, 78(5), pp.687-694.
67. Yoshida, T. and Kojima, K., 2015. Toyota MIRAI fuel cell vehicle and progress toward a future hydrogen society. *The Electrochemical Society Interface*, 24(2), p.45.
68. Barbir, F., Molter, T. and Dalton, L., 2005. Efficiency and weight trade-off analysis of regenerative fuel cells as energy storage for aerospace applications. *International Journal of Hydrogen Energy*, 30(4), pp.351-357.
69. Maclay, J.D., Brouwer, J. and Samuelsen, G.S., 2006. Dynamic analyses of regenerative fuel cell power for potential use in renewable residential applications. *International Journal of Hydrogen Energy*, 31(8), pp.994-1009.
70. Ralph, T.R., Hards, G.A., Keating, J.E., Campbell, S.A., Wilkinson, D.P., Davis, M., St-Pierre, J. and Johnson, M.C., 1997. Low cost electrodes for proton exchange membrane fuel cells: Performance in single cells and Ballard stacks. *Journal of the Electrochemical Society*, 144(11), p.3845.
71. Pu, Z., Zhang, G., Hassanpour, A., Zheng, D., Wang, S., Liao, S., Chen, Z. and Sun, S., 2021. Regenerative fuel cells: Recent progress, challenges, perspectives, and their applications for space energy system. *Applied Energy*, 283, p.116376.
72. Qiang, M.Z., 2005. Hydrogen: Green Energy in the 21st Century.
73. Yu, E.H., Wang, X., Krewer, U., Li, L. and Scott, K., 2012. Direct oxidation alkaline fuel cells: from materials to systems. *Energy & Environmental Science*, 5(2), pp.5668-5680.
74. Sadhasivam, T., Dhanabalan, K., Roh, S.H., Kim, T.H., Park, K.W., Jung, S., Kurkuri, M.D. and Jung, H.Y., 2017. A comprehensive review on unitized regenerative fuel cells: Crucial challenges and developments. *international journal of hydrogen energy*, 42(7), pp.4415-4433.
75. Shih, Z.Y., Periasamy, A.P., Hsu, P.C. and Chang, H.T., 2013. Synthesis and catalysis of copper sulfide/carbon nanodots for oxygen reduction in direct methanol fuel cells. *Applied Catalysis B: Environmental*, 132, pp.363-369.
76. Xia, W., Mahmood, A., Liang, Z., Zou, R. and Guo, S., 2016. Earth-abundant nanomaterials for oxygen reduction. *Angewandte Chemie International Edition*, 55(8), pp.2650-2676.
77. Regmi, Y.N., Peng, X., Fornaciari, J.C., Wei, M., Myers, D.J., Weber, A.Z. and Danilovic, N., 2020. A low temperature unitized regenerative fuel cell realizing 60% round trip

- efficiency and 10000 cycles of durability for energy storage applications. *Energy & Environmental Science*, 13(7), pp.2096-2105.
78. Li, X., Popov, B.N., Kawahara, T., Yanagi, H. 2011. Recent advances in non-precious metal catalyst for oxygen reduction reaction in polymer electrolyte fuel cells. *Energy Environ. Sci.*, 4, pp.114–130.
79. Li, S., Hao, X., Abudula, A. and Guan, G., 2019. Nanostructured Co-based bifunctional electrocatalysts for energy conversion and storage: current status and perspectives. *Journal of Materials Chemistry A*, 7(32), pp.18674-18707.
80. Zhao, D., Zhuang, Z., Cao, X., Zhang, C., Peng, Q., Chen, C. and Li, Y., 2020. Atomic site electrocatalysts for water splitting, oxygen reduction and selective oxidation. *Chemical Society Reviews*, 49(7), pp.2215-2264.
81. Ibrahim, H., Ilinca, A. and Perron, J., 2008. Energy storage systems—Characteristics and comparisons. *Renewable and sustainable energy reviews*, 12(5), pp.1221-1250.
82. Nong, H.N., Falling, L.J., Bergmann, A., Klingenhof, M., Tran, H.P., Spöri, C., Mom, R., Timoshenko, J., Zichittella, G., Knop-Gericke, A. and Piccinin, S., 2020. Key role of chemistry versus bias in electrocatalytic oxygen evolution. *Nature*, 587(7834), pp.408-413.
83. Seh, Z.W., Kibsgaard, J., Dickens, C.F., Chorkendorff, I.B., Nørskov, J.K. and Jaramillo, T.F., 2017. Combining theory and experiment in electrocatalysis: Insights into materials design. *Science*, 355(6321), p.eaad4998.
84. Busch, M., Halck, N.B., Kramm, U.I., Siahrostami, S., Krtil, P. and Rossmeisl, J., 2016. Beyond the top of the volcano?—A unified approach to electrocatalytic oxygen reduction and oxygen evolution. *Nano Energy*, 29, pp.126-135.
85. Shen, P.K., Wang, C.Y., Sun, X. and Zhang, J. eds., 2018. *Electrochemical energy: advanced materials and technologies*. CRC press.
86. Han, L., Dong, S. and Wang, E., 2016. Transition-metal (Co, Ni, and Fe)-based electrocatalysts for the water oxidation reaction. *Advanced materials*, 28(42), pp.9266-9291.
87. Gong, M. and Dai, H., 2015. A mini review of NiFe-based materials as highly active oxygen evolution reaction electrocatalysts. *Nano Research*, 8, pp.23-39.
88. Zhang, C., Wang, B., Shen, X., Liu, J., Kong, X., Chuang, S.S., Yang, D., Dong, A. and Peng, Z., 2016. A nitrogen-doped ordered mesoporous carbon/graphene framework as bifunctional electrocatalyst for oxygen reduction and evolution reactions. *Nano Energy*, 30, pp.503-510.

89. Man, I.C., Su, H.Y., Calle-Vallejo, F., Hansen, H.A., Martínez, J.I., Inoglu, N.G., Kitchin, J., Jaramillo, T.F., Nørskov, J.K. and Rossmeisl, J., 2011. Universality in oxygen evolution electrocatalysis on oxide surfaces. *ChemCatChem*, 3(7), pp.1159-1165
90. Matsumoto, Y. and Sato, E., 1986. Electrocatalytic properties of transition metal oxides for oxygen evolution reaction. *Materials chemistry and physics*, 14(5), pp.397-426.
91. Fabbri, E., Haberer, A., Walz, K., Kötter, R. and Schmidt, T.J., 2014. Developments and perspectives of oxide-based catalysts for the oxygen evolution reaction. *Catalysis Science & Technology*, 4(11), pp.3800-3821.
92. Hall, D.E., 1983. Ni(OH)₂-Impregnated Anodes for Alkaline Water Electrolysis. *Journal of The Electrochemical Society*, 130(2), p.317.
93. Bockris, J.O.M. and Otagawa, T., 2002. Mechanism of oxygen evolution on perovskites. *The Journal of Physical Chemistry*, 87(15), pp.2960-2971.
94. Wade, W.H. and Hackerman, N., 1957. Anodic phenomena at an iron electrode. *Transactions of the Faraday Society*, 53, pp.1636-1647.
95. Bockris, J.O.M., 1956. Kinetics of activation controlled consecutive electrochemical reactions: anodic evolution of oxygen. *The Journal of Chemical Physics*, 24(4), pp.817-827.
96. Suen, N.T., Hung, S.F., Quan, Q., Zhang, N., Xu, Y.J. and Chen, H.M., 2017. Electrocatalysis for the oxygen evolution reaction: recent development and future perspectives. *Chemical Society Reviews*, 46(2), pp.337-365.
97. Lyons, M.E. and Burke, L.D., 1987. Mechanism of oxygen reactions at porous oxide electrodes. Part 1.—Oxygen evolution at RuO₂ and Ru_xSn_{1-x}O₂ electrodes in alkaline solution under vigorous electrolysis conditions. *Journal of the Chemical Society, Faraday Transactions 1: Physical Chemistry in Condensed Phases*, 83(2), pp.299-321.
98. Lyons, M.E. and Floquet, S., 2011. Mechanism of oxygen reactions at porous oxide electrodes. Part 2—Oxygen evolution at RuO₂, IrO₂ and Ir_xRu_{1-x}O₂ electrodes in aqueous acid and alkaline solution. *Physical Chemistry Chemical Physics*, 13(12), pp.5314-5335.
99. Subbaraman, R., Tripkovic, D., Chang, K.C., Strmcnik, D., Paulikas, A.P., Hirunsit, P., Chan, M., Greeley, J., Stamenkovic, V. and Markovic, N.M., 2012. Trends in activity for the water electrolyser reactions on 3d M (Ni, Co, Fe, Mn) hydr(oxy)oxide catalysts. *Nature materials*, 11(6), pp.550-557.
100. Subbaraman, R., Tripkovic, D., Strmcnik, D., Chang, K.C., Uchimura, M., Paulikas, A.P., Stamenkovic, V. and Markovic, N.M., 2011. Enhancing hydrogen evolution activity

- in water splitting by tailoring Li⁺-Ni(OH)₂-Pt interfaces. *Science*, 334(6060), pp.1256-1260.
101. Anantharaj, S., Karthick, K. and Kundu, S., 2017. Evolution of layered double hydroxides (LDH) as high-performance water oxidation electrocatalysts: A review with insights on structure, activity and mechanism. *Materials Today Energy*, 6, pp.1-26.
 102. Jiao, Y., Zheng, Y., Jaroniec, M. and Qiao, S.Z., 2015. Design of electrocatalysts for oxygen-and hydrogen-involving energy conversion reactions. *Chemical Society Reviews*, 44(8), pp.2060-2086.
 103. Masa, J., Weide, P., Peeters, D., Sinev, I., Xia, W., Sun, Z., Somsen, C., Muhler, M. and Schuhmann, W., 2016. Amorphous cobalt boride (Co₂B) as a highly efficient nonprecious catalyst for electrochemical water splitting: oxygen and hydrogen evolution. *Advanced Energy Materials*, 6(6), p.1502313.
 104. Anantharaj, S., Ede, S.R., Sakthikumar, K., Karthick, K., Mishra, S. and Kundu, S., 2016. Recent trends and perspectives in electrochemical water splitting with an emphasis on sulfide, selenide, and phosphide catalysts of Fe, Co, and Ni: a review. *Acs Catalysis*, 6(12), pp.8069-8097.
 105. Bard, A.J., Faulkner, L.R. and White, H.S., 2022. *Electrochemical methods: fundamentals and applications*. John Wiley & Sons.
 106. Zhu, Y.P., Xu, X., Su, H., Liu, Y.P., Chen, T. and Yuan, Z.Y., 2015. Ultrafine metal phosphide nanocrystals in situ decorated on highly porous heteroatom-doped carbons for active electrocatalytic hydrogen evolution. *ACS Applied Materials & Interfaces*, 7(51), pp.28369-28376.
 107. Gorlin, Y. and Jaramillo, T.F., 2010. A bifunctional nonprecious metal catalyst for oxygen reduction and water oxidation. *Journal of the American Chemical Society*, 132(39), pp.13612-13614.
 108. Gong, M., Li, Y., Wang, H., Liang, Y., Wu, J.Z., Zhou, J., Wang, J., Regier, T., Wei, F. and Dai, H., 2013. An advanced Ni-Fe layered double hydroxide electrocatalyst for water oxidation. *Journal of the American Chemical Society*, 135(23), pp.8452-8455.
 109. Tahir, M., Pan, L., Idrees, F., Zhang, X., Wang, L., Zou, J.J. and Wang, Z.L., 2017. Electrocatalytic oxygen evolution reaction for energy conversion and storage: a comprehensive review. *Nano Energy*, 37, pp.136-157.
 110. Shinagawa, T., Garcia-Esparza, A.T. and Takanabe, K., 2015. Insight on Tafel slopes from a microkinetic analysis of aqueous electrocatalysis for energy conversion. *Scientific reports*, 5(1), p.13801.

111. Doyle, R.L. and Lyons, M.E., 2016. The oxygen evolution reaction: mechanistic concepts and catalyst design. *Photoelectrochemical Solar Fuel Production: From Basic Principles to Advanced Devices*, pp.41-104.
112. Han, L., Dong, S. and Wang, E., 2016. Transition-metal (Co, Ni, and Fe)-based electrocatalysts for the water oxidation reaction. *Advanced materials*, 28(42), pp.9266-9291.
113. Anantharaj, S., Kundu, S. and Noda, S., 2021. "The Fe Effect": A review unveiling the critical roles of Fe in enhancing OER activity of Ni and Co based catalysts. *Nano Energy*, 80, p.105514.
114. Vrubel, H., Moehl, T., Grätzel, M. and Hu, X., 2013. Revealing and accelerating slow electron transport in amorphous molybdenum sulphide particles for hydrogen evolution reaction. *Chemical Communications*, 49(79), pp.8985-8987.
115. Anantharaj, S., Karthik, P.E. and Kundu, S., 2017. Petal-like hierarchical array of ultrathin Ni (OH)₂ nanosheets decorated with Ni (OH)₂ nanoburles: a highly efficient OER electrocatalyst. *Catalysis Science & Technology*, 7(4), pp.882-893.
116. Kumar, T.N., Sivabalan, S., Chandrasekaran, N. and Phani, K.L., 2015. Synergism between polyurethane and polydopamine in the synthesis of Ni-Fe alloy monoliths. *Chemical Communications*, 51(10), pp.1922-1925.
117. Guo, S.X., Liu, Y., Bond, A.M., Zhang, J., Karthik, P.E., Maheshwaran, I., Kumar, S.S. and Phani, K.L.N., 2014. Facile electrochemical co-deposition of a graphene-cobalt nanocomposite for highly efficient water oxidation in alkaline media: direct detection of underlying electron transfer reactions under catalytic turnover conditions. *Physical Chemistry Chemical Physics*, 16(35), pp.19035-19045.
118. Guo, S.X., Liu, Y., Bond, A.M., Zhang, J., Karthik, P.E., Maheshwaran, I., Kumar, S.S. and Phani, K.L.N., 2014. Facile electrochemical co-deposition of a graphene-cobalt nanocomposite for highly efficient water oxidation in alkaline media: direct detection of underlying electron transfer reactions under catalytic turnover conditions. *Physical Chemistry Chemical Physics*, 16(35), pp.19035-19045.
119. Karthik, P.E., Raja, K.A., Kumar, S.S., Phani, K.L.N., Liu, Y., Guo, S.X., Zhang, J. and Bond, A.M., 2015. Electroless deposition of iridium oxide nanoparticles promoted by condensation of [Ir(OH)₆]²⁻ on an anodized Au surface: application to electrocatalysis of the oxygen evolution reaction. *RSC Advances*, 5(5), pp.3196-3199.

120. McCrory, C.C., Jung, S., Peters, J.C. and Jaramillo, T.F., 2013. Benchmarking heterogeneous electrocatalysts for the oxygen evolution reaction. *Journal of the American Chemical Society*, 135(45), pp.16977-16987.
121. Wei, C., Sun, S., Mandler, D., Wang, X., Qiao, S.Z. and Xu, Z.J., 2019. Approaches for measuring the surface areas of metal oxide electrocatalysts for determining their intrinsic electrocatalytic activity. *Chemical Society Reviews*, 48(9), pp.2518-2534.
122. Gao, M., Sheng, W., Zhuang, Z., Fang, Q., Gu, S., Jiang, J. and Yan, Y., 2014. Efficient water oxidation using nanostructured α -nickel-hydroxide as an electrocatalyst. *Journal of the American Chemical Society*, 136(19), pp.7077-7084.
123. Anantharaj, S., Jayachandran, M. and Kundu, S., 2016. Unprotected and interconnected Ru 0 nano-chain networks: advantages of unprotected surfaces in catalysis and electrocatalysis. *Chemical Science*, 7(5), pp.3188-3205.
124. Zou, X. and Zhang, Y., 2015. Noble metal-free hydrogen evolution catalysts for water splitting. *Chemical Society Reviews*, 44(15), pp.5148-5180.
125. Liu, K., Zhong, H., Meng, F., Zhang, X., Yan, J. and Jiang, Q., 2017. Recent advances in metal–nitrogen–carbon catalysts for electrochemical water splitting. *Materials Chemistry Frontiers*, 1(11), pp.2155-2173.
126. Hu, J., Zhang, C., Meng, X., Lin, H., Hu, C., Long, X. and Yang, S., 2017. Hydrogen evolution electrocatalysis with binary-nonmetal transition metal compounds. *Journal of Materials Chemistry A*, 5(13), pp.5995-6012.
127. Guo, Y., Tang, J., Wang, Z., Sugahara, Y. and Yamauchi, Y., 2018. Hollow porous heterometallic phosphide nanocubes for enhanced electrochemical water splitting. *Small*, 14(44), p.1802442.
128. Xu, K., Chen, P., Li, X., Tong, Y., Ding, H., Wu, X., Chu, W., Peng, Z., Wu, C. and Xie, Y., 2015. Metallic nickel nitride nanosheets realizing enhanced electrochemical water oxidation. *Journal of the American Chemical Society*, 137(12), pp.4119-4125.
129. Iwakura, C., Hirao, K. and Tamura, H., 1977. Anodic evolution of oxygen on ruthenium in acidic solutions. *Electrochimica acta*, 22(4), pp.329-334.
130. Beni, G., Schiavone, L.M., Shay, J.L., Dautremont-Smith, W.C. and Schneider, B.S., 1979. Electrocatalytic oxygen evolution on reactively sputtered electrochromic iridium oxide films. *Nature*, 282(5736), pp.281-283.
131. Oliva, P., Leonardi, J., Laurent, J.F., Delmas, C., Braconnier, J.J., Figlarz, M., Fievet, F. and De Guibert, A., 1982. Review of the structure and the electrochemistry of nickel hydroxides and oxy-hydroxides. *Journal of Power sources*, 8(2), pp.229-255.

132. Burke, L.D., Lyons, M.E. and Murphy, O.J., 1982. Formation of hydrous oxide films on cobalt under potential cycling conditions. *Journal of electroanalytical chemistry and interfacial electrochemistry*, 132, pp.247-261.
133. Reier, T., Oezaslan, M. and Strasser, P., 2012. Electrocatalytic oxygen evolution reaction (OER) on Ru, Ir, and Pt catalysts: a comparative study of nanoparticles and bulk materials. *Acs Catalysis*, 2(8), pp.1765-1772.
134. Frydendal, R., Paoli, E.A., Knudsen, B.P., Wickman, B., Malacrida, P., Stephens, I.E. and Chorkendorff, I., 2014. Benchmarking the stability of oxygen evolution reaction catalysts: the importance of monitoring mass losses. *ChemElectroChem*, 1(12), pp.2075-2081.
135. Jiao, Y., Zheng, Y., Jaroniec, M. and Qiao, S.Z., 2015. Design of electrocatalysts for oxygen- and hydrogen-involving energy conversion reactions. *Chemical Society Reviews*, 44(8), pp.2060-2086.
136. Danilovic, N., Subbaraman, R., Chang, K.C., Chang, S.H., Kang, Y.J., Snyder, J., Paulikas, A.P., Strmcnik, D., Kim, Y.T., Myers, D. and Stamenkovic, V.R., 2014. Activity–stability trends for the oxygen evolution reaction on monometallic oxides in acidic environments. *The journal of physical chemistry letters*, 5(14), pp.2474-2478.
137. Lee, Y., Suntivich, J., May, K.J., Perry, E.E. and Shao-Horn, Y., 2012. Synthesis and activities of rutile IrO₂ and RuO₂ nanoparticles for oxygen evolution in acid and alkaline solutions. *The journal of physical chemistry letters*, 3(3), pp.399-404.
138. Pi, Y., Zhang, N., Guo, S., Guo, J. and Huang, X., 2016. Ultrathin laminar Ir superstructure as highly efficient oxygen evolution electrocatalyst in broad pH range. *Nano letters*, 16(7), pp.4424-4430.
139. Makarova, M.V., Jirkovský, J., Klementová, M., Jirka, I., Macounová, K. and Krtíl, P., 2008. The electrocatalytic behavior of Ru_{0.8}Co_{0.2}O_{2-x} —the effect of particle shape and surface composition. *Electrochimica Acta*, 53(5), pp.2656-2664.
140. Li, H., Tang, Q., He, B. and Yang, P., 2016. Robust electrocatalysts from an alloyed Pt–Ru–M (M= Cr, Fe, Co, Ni, Mo)-decorated Ti mesh for hydrogen evolution by seawater splitting. *Journal of Materials Chemistry A*, 4(17), pp.6513-6520.
141. Cherevko, S., Geiger, S., Kasian, O., Kulyk, N., Grote, J.P., Savan, A., Shrestha, B.R., Merzlikin, S., Breitbach, B., Ludwig, A. and Mayrhofer, K.J., 2016. Oxygen and hydrogen evolution reactions on Ru, RuO₂, Ir, and IrO₂ thin film electrodes in acidic and alkaline electrolytes: A comparative study on activity and stability. *Catalysis Today*, 262, pp.170-180.

142. Kötz, R.+, Lewerenz, H.J. and Stucki, S., 1983. XPS studies of oxygen evolution on Ru and RuO₂ anodes. *Journal of The Electrochemical Society*, 130(4), p.825.
143. Kötz, R.+, Neff, H. and Stucki, S., 1984. Anodic Iridium Oxide Films: XPS-Studies of Oxidation State Changes and O₂ evolution. *Journal of the Electrochemical Society*, 131(1), p.72.
144. Yao, W., Yang, J., Wang, J. and Nuli, Y., 2007. Chemical deposition of platinum nanoparticles on iridium oxide for oxygen electrode of unitized regenerative fuel cell. *Electrochemistry Communications*, 9(5), pp.1029-1034.
145. Ardizzone, S., Bianchi, C.L., Cappelletti, G., Ionita, M., Minguzzi, A., Rondinini, S. and Vertova, A., 2006. Composite ternary SnO₂-IrO₂-Ta₂O₅ oxide electrocatalysts. *Journal of Electroanalytical Chemistry*, 589(1), pp.160-166.
146. Da Silva, L.A., Alves, V.A., Da Silva, M.A.P., Trasatti, S. and Boodts, J.F.C., 1997. Oxygen evolution in acid solution on IrO₂+TiO₂ ceramic films. A study by impedance, voltammetry and SEM. *Electrochimica Acta*, 42(2), pp.271-281.
147. Balko, E.N. and Nguyen, P.H., 1991. Iridium-tin mixed oxide anode coatings. *Journal of applied electrochemistry*, 21(8), pp.678-682.
148. Benedetti, A., Riello, P., Battaglin, G., De Battisti, A. and Barbieri, A., 1994. Physicochemical properties of thermally prepared Ti-supported IrO₂+ZrO₂ electrocatalysts. *Journal of Electroanalytical Chemistry*, 376(1-2), pp.195-202.
149. Chen, G., Chen, X. and Yue, P.L., 2002. Electrochemical Behavior of Novel Ti/IrO_x-Sb₂O₅-SnO₂ Anodes. *The Journal of Physical Chemistry B*, 106(17), pp.4364-4369.
150. Audichon, T., Mamaca, N., Morais, C., Servat, K., Napporn, T.W., Mayousse, E., Guillet, N. and Kokoh, K.B., 2013. Synthesis of Ru_xIr_{1-x}O₂ anode electrocatalysts for proton exchange membrane water electrolysis. *ECS Transactions*, 45(21), p.47.
151. Bode, H., Dehmelt, K. and Witte, J.J.E.A., 1966. Zur kenntnis der nickelhydroxidelektrode—I. Über das nickel (II)-hydroxidhydrat. *Electrochimica Acta*, 11(8), pp.1079-1087.
152. Oliva, P., Leonardi, J., Laurent, J.F., Delmas, C., Braconnier, J.J., Figlarz, M., Fievet, F. and De Guibert, A., 1982. Review of the structure and the electrochemistry of nickel hydroxides and oxy-hydroxides. *Journal of Power sources*, 8(2), pp.229-255.
153. Młynarek, G., Paszkiewicz, M. and Radniecka, A., 1984. The effect of ferric ions on the behaviour of a nickelous hydroxide electrode. *Journal of applied electrochemistry*, 14(2), pp.145-149.

154. Arciga-Duran, E., Meas, Y., Pérez-Bueno, J.J., Ballesteros, J.C. and Trejo, G., 2018. Effect of oxygen vacancies in electrodeposited NiO towards the oxygen evolution reaction: Role of Ni-Glycine complexes. *Electrochimica Acta*, 268, pp.49-58.
155. Trotochaud, L., Young, S.L., Ranney, J.K. and Boettcher, S.W., 2014. Nickel-iron oxyhydroxide oxygen-evolution electrocatalysts: the role of intentional and incidental iron incorporation. *Journal of the American Chemical Society*, 136(18), pp.6744-6753.
156. El Wakkad, S.E.S. and Hickling, A., 1950. The anodic behaviour of metals. Part VI.—Cobalt. *Transactions of the Faraday Society*, 46, pp.820-824.
157. Dangwal Pandey, A., Jia, C., Schmidt, W., Leoni, M., Schwickardi, M., Schüth, F. and Weidenthaler, C., 2012. Size-controlled synthesis and microstructure investigation of Co₃O₄ nanoparticles for low-temperature CO oxidation. *The Journal of Physical Chemistry C*, 116(36), pp.19405-19412.
158. Lyons, M.E. and Brandon, M.P., 2008. The oxygen evolution reaction on passive oxide covered transition metal electrodes in aqueous alkaline solution. Part 1- Nickel. *International Journal of Electrochemical Science*, 3(12), pp.1386-1424.
159. Lyons, M.E. and Brandon, M.P., 2008. The oxygen evolution reaction on passive oxide covered transition metal electrodes in aqueous alkaline solution. Part 1- Nickel. *International Journal of Electrochemical Science*, 3(12), pp.1386-1424.
160. Rasiyah, P. and Tseung, A.C.C., 1983. A Mechanistic Study of Oxygen Evolution on NiCo₂O₄: II. Electrochemical Kinetics. *Journal of the Electrochemical Society*, 130(12), p.2384.
161. Singh, R.N., Koenig, J.F., Poillerat, G. and Chartier, P., 1990. Electrochemical Studies on Protective Thin Co₃O₄ and NiCo₂O₄ Films Prepared on Titanium by Spray Pyrolysis for Oxygen Evolution. *Journal of The Electrochemical Society*, 137(5), p.1408.
162. Koza, J.A., He, Z., Miller, A.S. and Switzer, J.A., 2012. Electrodeposition of Crystalline Co₃O₄: A Catalyst for the Oxygen Evolution Reaction. *Chemistry of Materials*, 24(18), pp.3567-3573.
163. Zheng, Z., Geng, W., Wang, Y., Huang, Y. and Qi, T., 2017. NiCo₂O₄ nanoflakes supported on titanium suboxide as a highly efficient electrocatalyst towards oxygen evolution reaction. *International Journal of Hydrogen Energy*, 42(1), pp.119-124.
164. Shao, Y., Zheng, M., Cai, M., He, L. and Xu, C., 2017. Improved electrocatalytic performance of core-shell NiCo/NiCoO_x with amorphous FeOOH for oxygen-evolution reaction. *Electrochimica Acta*, 257, pp.1-8.

165. Matsumoto, Y., Kurimoto, J. and Sato, E., 1979. Oxygen evolution on SrFeO₃ electrode. *Journal of Electroanalytical Chemistry and Interfacial Electrochemistry*, 102(1), pp.77-83.
166. Matsumoto, Y., Yamada, S., Nishida, T. and Sato, E., 1980. Oxygen evolution on La_{1-x}Sr_xFe_{1-y}Co_yO₃ series oxides. *Journal of The Electrochemical Society*, 127(11), p.2360.
167. Bockris, J.O.M. and Otagawa, T., 1984. The electrocatalysis of oxygen evolution on perovskites. *Journal of The Electrochemical Society*, 131(2), p.290.
168. Suntivich, J., May, K.J., Gasteiger, H.A., Goodenough, J.B. and Shao-Horn, Y., 2011. A perovskite oxide optimized for oxygen evolution catalysis from molecular orbital principles. *Science*, 334(6061), pp.1383-1385.
169. Mizushima, K.J.P.C., Jones, P.C., Wiseman, P.J. and Goodenough, J.B., 1980. Li_xCoO₂ (0 < x < 1): A new cathode material for batteries of high energy density. *Materials Research Bulletin*, 15(6), pp.783-789.
170. Yoshino, A., Sanechika, K. and Nakajima, T., 1987. "Secondary Battery," *US Patent 4,668,595* (Vol. 26). 1987-05.
171. Mendiboure, A., Delmas, C. and Hagemuller, P., 1984. New layered structure obtained by electrochemical deintercalation of the metastable LiCoO₂ (O₂) variety. *Materials research bulletin*, 19(10), pp.1383-1392.
172. Dahn, J.R., Fuller, E.W., Obrovac, M. and Von Sacken, U., 1994. Thermal stability of Li_xCoO₂, Li_xNiO₂ and λ-MnO₂ and consequences for the safety of Li-ion cells. *Solid State Ionics*, 69(3-4), pp.265-270.
173. Jiang, J.R.D.J. and Dahn, J.R., 2004. ARC studies of the thermal stability of three different cathode materials: LiCoO₂; Li[Ni_{0.1}Co_{0.8}Mn_{0.1}]O₂; and LiFePO₄, in LiPF₆ and LiBoB EC/DEC electrolytes. *Electrochemistry Communications*, 6(1), pp.39-43.
174. Chebiam, R.V., Prado, F. and Manthiram, A., 2001. Soft Chemistry Synthesis and Characterization of Layered Li_{1-x}Ni_{1-y}Co_yO_{2-δ} (0 ≤ x ≤ 1 and 0 ≤ y ≤ 1). *Chemistry of materials*, 13(9), pp.2951-2957.
175. Chebiam, R.V., Kannan, A.M., Prado, F. and Manthiram, A., 2001. Comparison of the chemical stability of the high energy density cathodes of lithium-ion batteries. *Electrochemistry communications*, 3(11), pp.624-627.
176. Jiang, S., Suo, H., Zheng, X., Zhang, T., Lei, Y., Wang, Y.X., Lai, W.H. and Wang, G., 2022. Lightest Metal Leads to Big Change: Lithium-Mediated Metal Oxides for Oxygen Evolution Reaction. *Advanced Energy Materials*, 12(33), p.2201934.

177. Yang, J., Tang, D., Liu, Y., Li, W. and Li, J., 2023. Lithium Electrochemical Tuning Engineering in an Aqueous System of LiCoO₂ for Enhanced Oxygen Evolution Activity. *Journal of The Electrochemical Society*, 170(4), p.046502.
178. Nitta, N., Wu, F., Lee, J.T. and Yushin, G., 2015. Li-ion battery materials: present and future. *Materials today*, 18(5), pp.252-264.
179. Lu, Z., Jiang, K., Chen, G., Wang, H. and Cui, Y., 2018. Lithium electrochemical tuning for electrocatalysis. *Advanced Materials*, 30(48), p.1800978.
180. Gardner, G.P., Go, Y.B., Robinson, D.M., Smith, P.F., Hadermann, J., Abakumov, A., Greenblatt, M. and Dismukes, G.C., 2012. Structural requirements in lithium cobalt oxides for the catalytic oxidation of water. *Angewandte Chemie International Edition*, 51(7), pp.1616-1619.
181. Maiyalagan, T., Jarvis, K.A., Therese, S., Ferreira, P.J. and Manthiram, A., 2014. Spinel-type lithium cobalt oxide as a bifunctional electrocatalyst for the oxygen evolution and oxygen reduction reactions. *Nature communications*, 5(1), p.3949.
182. Lu, Z., Wang, H., Kong, D., Yan, K., Hsu, P.C., Zheng, G., Yao, H., Liang, Z., Sun, X. and Cui, Y., 2014. Electrochemical tuning of layered lithium transition metal oxides for improvement of oxygen evolution reaction. *Nature communications*, 5(1), p.4345.
183. Chen, Z., Wang, J., Chao, D., Baikie, T., Bai, L., Chen, S., Zhao, Y., Sum, T.C., Lin, J. and Shen, Z., 2016. Hierarchical porous LiNi_{1/3}Co_{1/3}Mn_{1/3}O₂ nano-/micro spherical cathode material: minimized cation mixing and improved Li⁺ mobility for enhanced electrochemical performance. *Scientific reports*, 6(1), p.25771.
184. Colligan, N., Augustyn, V. and Manthiram, A., 2015. Evidence of Localized Lithium Removal in Layered and Lithiated Spinel Li_{1-x}CoO₂ (0 ≤ x ≤ 0.9) under Oxygen Evolution Reaction Conditions. *The Journal of Physical Chemistry C*, 119(5), pp.2335-2340.
185. Wang, J., Kim, S.J., Liu, J., Gao, Y., Choi, S., Han, J., Shin, H., Jo, S., Kim, J., Ciucci, F. and Kim, H., 2021. Redirecting dynamic surface restructuring of a layered transition metal oxide catalyst for superior water oxidation. *Nature Catalysis*, 4(3), pp.212-222.
186. Augustyn, V. and Manthiram, A., 2015. Characterization of Layered LiMO₂ Oxides for the Oxygen Evolution Reaction of Metal–Air Batteries (M = Mn, Co, Ni). *ChemPlusChem*, 80(2), pp.422-427.
187. Gupta, A., Chemelewski, W.D., Buddie Mullins, C. and Goodenough, J.B., 2015. High-rate oxygen evolution reaction on Al-doped LiNiO₂. *Advanced Materials*, 27(39), pp.6063-6067.

188. Huang, D., Yu, J., Zhang, Z., Engtrakul, C., Burrell, A., Zhou, M., Luo, H. and Tenent, R.C., 2020. Enhancing the electrocatalysis of $\text{LiNi}_{0.5}\text{Co}_{0.2}\text{Mn}_{0.3}\text{O}_2$ by introducing lithium deficiency for oxygen evolution reaction. *ACS Applied Materials & Interfaces*, 12(9), pp.10496-10502.
189. Zheng, X., Chen, Y., Zheng, X., Zhao, G., Rui, K., Li, P., Xu, X., Cheng, Z., Dou, S.X. and Sun, W., 2019. Electronic structure engineering of LiCoO_2 toward enhanced oxygen electrocatalysis. *Advanced Energy Materials*, 9(16), p.1803482.
190. Song, J., Wei, C., Huang, Z.F., Liu, C., Zeng, L., Wang, X. and Xu, Z.J., 2020. A review on fundamentals for designing oxygen evolution electrocatalysts. *Chemical Society Reviews*, 49(7), pp.2196-2214.
191. Kuznetsov, D.A., Han, B., Yu, Y., Rao, R.R., Hwang, J., Román-Leshkov, Y. and Shao-Horn, Y., 2018. Tuning redox transitions via inductive effect in metal oxides and complexes, and implications in oxygen electrocatalysis. *Joule*, 2(2), pp.225-244.

2.5 References

1. Jansen, M., 2002. A concept for synthesis planning in solid-state chemistry. *Angewandte Chemie International Edition*, 41(20), pp.3746-3766.
2. Smida, Y.B., Marzouki, R., Kaya, S., Erkan, S., Zid, M.F. and Hamzaoui, A.H., 2020. Synthesis Methods in Solid-State Chemistry. *Synthesis Methods and Crystallization, Intech Open*. doi: 10.5772/intechopen.93337.
3. Evecan, D., Kaplan, Ş.S., Sönmez, M.Ş., Yıldırım, S., Okutan, M., Deligöz, H. and Zayim, E., 2019. Smart glass electrochromic device fabrication of uniform tungsten oxide films from its powder synthesized by solution combustion method. *Microelectronic Engineering*, 215, p.110989.
4. González-Cortés, S.L. and Imbert, F.E., 2013. Fundamentals, properties and applications of solid catalysts prepared by solution combustion synthesis (SCS). *Applied Catalysis A: General*, 452, pp.117-131.
5. Banger, K.K., Yamashita, Y., Mori, K., Peterson, R.L., Leedham, T., Rickard, J. and Sirringhaus, H.J.N.M., 2011. Low-temperature, high-performance solution-processed metal oxide thin-film transistors formed by a 'sol-gel on chip' process. *Nature Materials*, 10(1), pp.45-50.
6. Shao, Z., Zhou, W. and Zhu, Z., 2012. Advanced synthesis of materials for intermediate-temperature solid oxide fuel cells. *Progress in Materials Science*, 57(4), pp.804-874.
7. Danks, A.E., Hall, S.R. and Schnepf, Z.J.M.H., 2016. The evolution of 'sol-gel' chemistry as a technique for materials synthesis. *Materials Horizons*, 3(2), pp.91-112.
8. Birol, H., Rambo, C.R., Guiotoku, M. and Hotza, D., 2013. Preparation of ceramic nanoparticles via cellulose-assisted glycine nitrate process: a review. *Rsc Advances*, 3(9), pp.2873-2884.
9. Patil, K.C., 2008. *Chemistry of nanocrystalline oxide materials: combustion synthesis, properties and applications*. World Scientific.
10. Drits, V., Środoń, J. and Eberl, D.D., 1997. XRD measurement of mean crystallite thickness of illite and illite/smectite: Reappraisal of the Kubler index and the Scherrer equation. *Clays and Clay Minerals*, 45, pp.461-475.
11. Thodeti, S., Reddy, R.M. and Kumar, J.S., 2016. Synthesis and characterization of pure and indium doped SnO₂ nanoparticles by sol-gel methods. *Int. J. Sci. Eng. Res*, 7, pp.310-317.
12. Guo, J., Meng, Z., Qiao, Y. and Li, B., 2023, February. Numerical Analysis of Structural Color for Photonic Crystal Hydrogel. In *Photonics* (Vol. 10, No. 2, p. 186). MDPI.

13. Ho, K.M., Chan, C.T. and Soukoulis, C.M., 1990. Existence of a photonic gap in periodic dielectric structures. *Physical Review Letters*, 65(25), p.3152.
14. Rietveld, H.M., 1969. A profile refinement method for nuclear and magnetic structures. *Journal of applied Crystallography*, 2(2), pp.65-71.
15. Chaudhari, S., 1999, December. Electron microscopy: an essential tool for the synthesis of thin film for practical applications. In *Proceeding of national conference on electron microscopy at DMSRDE* (pp. 1-3).
16. Pennycook, S.J. and Nellist, P.D. eds., 2011. *Scanning transmission electron microscopy: imaging and analysis*. Springer Science & Business Media.
17. Inkson, B.J., 2016. Scanning electron microscopy (SEM) and transmission electron microscopy (TEM) for materials characterization. In *Materials characterization using nondestructive evaluation (NDE) methods* (pp. 17-43). Woodhead publishing.
18. Scimeca, M., Bischetti, S., Lamsira, H.K., Bonfiglio, R. and Bonanno, E., 2018. Energy Dispersive X-ray (EDX) microanalysis: A powerful tool in biomedical research and diagnosis. *European journal of histochemistry: EJH*, 62(1).
19. Carter, C.B. and Williams, D.B. eds., 2016. *Transmission electron microscopy: Diffraction, imaging, and spectrometry*. Springer.
20. Kohl, H. and Reimer, L., 2008. Transmission Electron Microscopy. *Springer Series in Optical Sciences*, 36.
21. Van der Heide, P., 2011. *X-ray photoelectron spectroscopy: an introduction to principles and practices*. John Wiley & Sons.
22. Mane, A.T. and Patil, V.B., 2016. X-ray photoelectron spectroscopy of nanofillers and their polymer nanocomposites. In *Spectroscopy of Polymer Nanocomposites* (pp. 452-467). William Andrew Publishing.
23. Nageswaran, G., Choudhary, Y.S. and Jagannathan, S., 2017. Inductively coupled plasma mass spectrometry. In *Spectroscopic methods for nanomaterials characterization* (pp. 163-194). Elsevier.
24. Košler, J. and Sylvester, P.J., 2003. Present trends and the future of zircon in geochronology: laser ablation ICPMS. *Reviews in Mineralogy and Geochemistry* 53 (1), 243–275.
25. De Haseth, J.A., 1982. Fourier transform infrared spectrometry. In *Fourier, Hadamard, and Hilbert Transforms in Chemistry* (pp. 387-420). Boston, MA: Springer US.

26. Ojeda, J.J. and Dittrich, M., 2012. Fourier transform infrared spectroscopy for molecular analysis of microbial cells. *Microbial Systems Biology: Methods and Protocols*, pp.187-211.
27. Naderi, M., 2015. Surface area: brunauer–emmett–teller (BET). In *Progress in filtration and separation* (pp. 585-608). Academic Press.
28. Bardestani, R., Patience, G.S. and Kaliaguine, S., 2019. Experimental methods in chemical engineering: specific surface area and pore size distribution measurements—BET, BJH, and DFT. *The Canadian Journal of Chemical Engineering*, 97(11), pp.2781-2791.
29. Elgrishi, N., Rountree, K.J., McCarthy, B.D., Rountree, E.S., Eisenhart, T.T. and Dempsey, J.L., 2018. A practical beginner's guide to cyclic voltammetry. *Journal of chemical education*, 95(2), pp.197-206.
30. Bard, A.J., Faulkner, L.R. and White, H.S., 2022. *Electrochemical methods: fundamentals and applications*. John Wiley & Sons.
31. Guy, O.J. and Walker, K.A.D., 2016. Graphene functionalization for biosensor applications. *Silicon Carbide Biotechnology*, pp.85-141.

3.5 References

1. Ferreira, K.N., Iverson, T.M., Maghlaoui, K., Barber, J. and Iwata, S., 2004. Architecture of the photosynthetic oxygen-evolving center. *Science*, 303(5665), pp.1831-1838.
2. Kanan, M.W. and Nocera, D.G., 2008. In situ formation of an oxygen-evolving catalyst in neutral water containing phosphate and Co^{2+} . *Science*, 321(5892), pp.1072-1075.
3. Dau, H., Limberg, C., Reier, T., Risch, M., Roggan, S. and Strasser, P., 2010. The mechanism of water oxidation: from electrolysis via homogeneous to biological catalysis. *ChemCatChem*, 2(7), pp.724-761.
4. Trasatti, S., 1980. Electrocatalysis by oxides—attempt at a unifying approach. *Journal of Electroanalytical Chemistry and Interfacial Electrochemistry*, 111(1), pp.125-131.
5. Manchanda, R., Brudvig, G.W. and Crabtree, R.H., 1995. High-valent oxomanganese clusters: structural and mechanistic work relevant to the oxygen-evolving center in photosystem II. *Coordination Chemistry Reviews*, 144, pp.1-38.
6. Dismukes, G.C., Brimblecombe, R., Felton, G.A., Pryadun, R.S., Sheats, J.E., Spiccia, L. and Swiegers, G.F., 2009. Development of bioinspired Mn_4O_4^- cubane water oxidation catalysts: lessons from photosynthesis. *Accounts of Chemical Research*, 42(12), pp.1935-1943.
7. Brimblecombe, R., Bond, A.M., Dismukes, G.C., Swiegers, G.F. and Spiccia, L., 2009. Electrochemical investigation of Mn_4O_4^- cubane water-oxidizing clusters. *Physical Chemistry Chemical Physics*, 11(30), pp.6441-6449.
8. Lyons, M.E. and Brandon, M.P., 2010. A comparative study of the oxygen evolution reaction on oxidised nickel, cobalt and iron electrodes in base. *Journal of Electroanalytical Chemistry*, 641(1-2), pp.119-130.
9. Lyons, M.E. and Brandon, M.P., 2008. The oxygen evolution reaction on passive oxide covered transition metal electrodes in aqueous alkaline solution. Part 1- Nickel. *International Journal of Electrochemical Science*, 3(12), pp.1386-1424.
10. Gao, M., Sheng, W., Zhuang, Z., Fang, Q., Gu, S., Jiang, J. and Yan, Y., 2014. Efficient water oxidation using nanostructured α -nickel-hydroxide as an electrocatalyst. *Journal of the American Chemical Society*, 136(19), pp.7077-7084.
11. Jin, K., Seo, H., Hayashi, T., Balamurugan, M., Jeong, D., Go, Y.K., Hong, J.S., Cho, K.H., Kakizaki, H., Bonnet-Mercier, N. and Kim, M.G., 2017. Mechanistic investigation of water oxidation catalyzed by uniform, assembled MnO nanoparticles. *Journal of the American Chemical Society*, 139(6), pp.2277-2285.

12. Koza, J.A., He, Z., Miller, A.S. and Switzer, J.A., 2012. Electrodeposition of Crystalline Co_3O_4 A Catalyst for the Oxygen Evolution Reaction. *Chemistry of Materials*, 24(18), pp.3567-3573.
13. Trotochaud, L., Young, S.L., Ranney, J.K. and Boettcher, S.W., 2014. Nickel–iron oxyhydroxide oxygen-evolution electrocatalysts: the role of intentional and incidental iron incorporation. *Journal of the American Chemical Society*, 136(18), pp.6744-6753.
14. Singh, R.N., Koenig, J.F., Poillerat, G. and Chartier, P., 1990. Electrochemical Studies on Protective Thin Co_3O_4 and NiCo_2O_4 Films Prepared on Titanium by Spray Pyrolysis for Oxygen Evolution. *Journal of The Electrochemical Society*, 137(5), p.1408.
15. Suntivich, J., May, K.J., Gasteiger, H.A., Goodenough, J.B. and Shao-Horn, Y., 2011. A perovskite oxide optimized for oxygen evolution catalysis from molecular orbital principles. *Science*, 334(6061), pp.1383-1385.
16. Augustyn, V. and Manthiram, A., 2015. Characterization of Layered LiMO_2 Oxides for the Oxygen Evolution Reaction of Metal–Air Batteries (M= Mn, Co, Ni). *ChemPlusChem*, 80(2), pp.422-427.
17. Lee, S.W., Carlton, C., Risch, M., Surendranath, Y., Chen, S., Furutsuki, S., Yamada, A., Nocera, D.G. and Shao-Horn, Y., 2012. The nature of lithium battery materials under oxygen evolution reaction conditions. *Journal of the American Chemical Society*, 134(41), pp.16959-16962.
18. Maiyalagan, T., Jarvis, K.A., Therese, S., Ferreira, P.J. and Manthiram, A., 2014. Spinel-type lithium cobalt oxide as a bifunctional electrocatalyst for the oxygen evolution and oxygen reduction reactions. *Nature communications*, 5(1), p.3949.
19. Lu, Z., Wang, H., Kong, D., Yan, K., Hsu, P.C., Zheng, G., Yao, H., Liang, Z., Sun, X. and Cui, Y., 2014. Electrochemical tuning of layered lithium transition metal oxides for improvement of oxygen evolution reaction. *Nature communications*, 5(1), p.4345.
20. Augustyn, V., Therese, S., Turner, T.C. and Manthiram, A., 2015. Nickel-rich layered $\text{LiNi}_{1-x}\text{M}_x\text{O}_2$ (M= Mn, Fe, and Co) electrocatalysts with high oxygen evolution reaction activity. *Journal of Materials Chemistry A*, 3(32), pp.16604-16612.
21. Gupta, A., Chemelewski, W.D., Buddie Mullins, C. and Goodenough, J.B., 2015. High-rate oxygen evolution reaction on Al-doped LiNiO_2 . *Advanced Materials*, 27(39), pp.6063-6067.
22. Zhu, K., Wu, T., Zhu, Y., Li, X., Li, M., Lu, R., Wang, J., Zhu, X. and Yang, W., 2017. Layered Fe-substituted LiNiO_2 electrocatalysts for high-efficiency oxygen evolution reaction. *ACS Energy Letters*, 2(7), pp.1654-1660.

23. Ye, W., Fang, X., Chen, X. and Yan, D., 2018. A three-dimensional nickel–chromium layered double hydroxide micro/nanosheet array as an efficient and stable bifunctional electrocatalyst for overall water splitting. *Nanoscale*, 10(41), pp.19484-19491.
24. Kim, S., Ma, X., Ong, S.P. and Ceder, G., 2012. A comparison of destabilization mechanisms of the layered Na_xMO_2 and Li_xMO_2 compounds upon alkali de-intercalation. *Physical Chemistry Chemical Physics*, 14(44), pp.15571-15578.
25. (a) Yang, Y., Dang, L., Shearer, M.J., Sheng, H., Li, W., Chen, J., Xiao, P., Zhang, Y., Hamers, R.J. and Jin, S., 2018. Highly active trimetallic NiFeCr layered double hydroxide electrocatalysts for oxygen evolution reaction. *Advanced Energy Materials*, 8(15), p.1703189.
(b) Sun, Z., Yuan, M., Yang, H., Lin, L., Jiang, H., Ge, S., Li, H., Sun, G., Ma, S. and Yang, X., 2019. 3D porous amorphous γ -CrOOH on Ni foam as bifunctional electrocatalyst for overall water splitting. *Inorganic Chemistry*, 58(6), pp.4014-4018.
26. Gao, M., Ma, N., Yu, C. and Liu, Y., 2021. In situ synthesis of Fe-doped CrOOH nanosheets for efficient electrocatalytic water oxidation. *Nanotechnology*, 32(28), p.28LT01.
27. Ohzuku, T., Ueda, A., Nagayama, M., Iwakoshi, Y. and Komori, H., 1993. Comparative study of LiCoO_2 , $\text{LiNi}_{1/2}\text{Co}_{1/2}\text{O}_2$ and LiNiO_2 for 4 volt secondary lithium cells. *Electrochimica Acta*, 38(9), pp.1159-1167.
28. Feng, G.X., Li, L.F., Liu, J.Y., Liu, N., Li, H., Yang, X.Q., Huang, X.J., Chen, L.Q., Nam, K.W. and Yoon, W.S., 2009. Enhanced electrochemical lithium storage activity of LiCrO_2 by size effect. *Journal of Materials Chemistry*, 19(19), pp.2993-2998.
29. Cherkashinin, G., Nikolowski, K., Ehrenberg, H., Jacke, S., Dimesso, L. and Jaegermann, W., 2012. The stability of the SEI layer, surface composition and the oxidation state of transition metals at the electrolyte–cathode interface impacted by the electrochemical cycling: X-ray photoelectron spectroscopy investigation. *Physical Chemistry Chemical Physics*, 14(35), pp.12321-12331.
30. Powell, C.J. and Larson, P.E., 1978. Quantitative surface analysis by X-ray photoelectron spectroscopy. *Applications of Surface Science*, 1(2), pp.186-201.
31. Scofield, J.H., 1976. Hartree-Slater subshell photoionization cross-sections at 1254 and 1487 eV. *Journal of Electron Spectroscopy and Related Phenomena*, 8(2), pp.129-137.
32. Penn, D.R., 1976. Quantitative chemical analysis by ESCA. *Journal of Electron Spectroscopy and Related Phenomena*, 9(1), pp.29-40.

33. Anantharaj, S., Reddy, P.N. and Kundu, S., 2017. Core-oxidized amorphous cobalt phosphide nanostructures: an advanced and highly efficient oxygen evolution catalyst. *Inorganic Chemistry*, 56(3), pp.1742-1756.
34. Jung, S., McCrory, C.C., Ferrer, I.M., Peters, J.C. and Jaramillo, T.F., 2016. Benchmarking nanoparticulate metal oxide electrocatalysts for the alkaline water oxidation reaction. *Journal of Materials Chemistry A*, 4(8), pp.3068-3076.
35. Cheng, X., Fabbri, E., Yamashita, Y., Castelli, I.E., Kim, B., Uchida, M., Haumont, R., Puente-Orench, I. and Schmidt, T.J., 2018. Oxygen evolution reaction on perovskites: A multieffect descriptor study combining experimental and theoretical methods. *ACS Catalysis*, 8(10), pp.9567-9578.
36. Rao, M.M., Liebenow, C., Jayalakshmi, M., Wulff, H., Guth, U. and Scholz, F., 2001. High-temperature combustion synthesis and electrochemical characterization of LiNiO₂, LiCoO₂ and LiMn₂O₄ for lithium-ion secondary batteries. *Journal of Solid State Electrochemistry*, 5, pp.348-354.
37. Rao, M.M., Jayalakshmi, M., Schaff, O., Guth, U., Wulff, H. and Scholz, F., 1999. Electrochemical behaviour of solid lithium nickelate (LiNiO₂) in an aqueous electrolyte system. *J. Solid State Electrochem*, 4, p.17.
38. Singh, R.N., Singh, J.P., Lal, B., Thomas, M.J.K. and Bera, S., 2006. New NiFe_{2-x}Cr_xO₄ spinel films for O₂ evolution in alkaline solutions. *Electrochimica acta*, 51(25), pp.5515-5523.
39. Kim, S., Ma, X., Ong, S.P. and Ceder, G., 2012. A comparison of destabilization mechanisms of the layered Na_xMO₂ and Li_xMO₂ compounds upon alkali de-intercalation. *Physical Chemistry Chemical Physics*, 14(44), pp.15571-15578.
40. Weckhuysen, B.M., Wachs, I.E. and Schoonheydt, R.A., 1996. Surface chemistry and spectroscopy of chromium in inorganic oxides. *Chemical Reviews*, 96(8), pp.3327-3350.
41. Julien, C.M. and Massot, M., 2004, September. Vibrational spectroscopy of electrode materials for rechargeable lithium batteries: III. Oxide frameworks. In *Proceedings of the International Workshop Advanced Techniques for Energy Sources Investigation and Testing* (pp. 1-17).
42. Trivedi, M.K., Tallapragada, R.M., Branton, A., Trivedi, D., Nayak, G., Latiyal, O. and Jana, S., 2015. Characterization of physical, thermal and structural properties of chromium (VI) oxide powder: Impact of biofield treatment. *Powder Metallurgy & Mining*, 4(1).
43. Monico, L., Janssens, K., Cotte, M., Sorace, L., Vanmeert, F., Brunetti, B.G. and Miliani, C., 2016. Chromium speciation methods and infrared spectroscopy for studying the

- chemical reactivity of lead chromate-based pigments in oil medium. *Microchemical Journal*, 124, pp.272-282.
44. Ratnasamy, P. and Leonard, A.J., 1972. Structural evolution of chromia. *The Journal of Physical Chemistry*, 76(13), pp.1838-1843.
45. Bai, Y.L., Xu, H.B., Zhang, Y. and Li, Z.H., 2006. Reductive conversion of hexavalent chromium in the preparation of ultra-fine chromia powder. *Journal of Physics and Chemistry of Solids*, 67(12), pp.2589-2595.
46. Liang, S.T., Zhang, H.L. and Xu, H.B., 2020. Preparation of hexagonal and amorphous chromium oxyhydroxides by facile hydrolysis of K_xCrO_y . *Transactions of Nonferrous Metals Society of China*, 30(5), pp.1397-1405.

4.5 References

1. Chebiam, R.V., Kannan, A.M., Prado, F. and Manthiram, A., 2001. Comparison of the chemical stability of the high energy density cathodes of lithium-ion batteries. *Electrochemistry Communications*, 3(11), pp.624-627.
2. Chebiam, R.V., Prado, F. and Manthiram, A., 2001. Soft Chemistry Synthesis and Characterization of Layered $\text{Li}_{1-x}\text{Ni}_{1-y}\text{Co}_y\text{O}_{2-\delta}$ ($0 \leq x \leq 1$ and $0 \leq y \leq 1$). *Chemistry of Materials*, 13(9), pp.2951-2957.
3. Lu, Z., Wang, H., Kong, D., Yan, K., Hsu, P.C., Zheng, G., Yao, H., Liang, Z., Sun, X. and Cui, Y., 2014. Electrochemical tuning of layered lithium transition metal oxides for improvement of oxygen evolution reaction. *Nature Communications*, 5(1), p.4345.
4. Maiyalagan, T., Jarvis, K.A., Therese, S., Ferreira, P.J. and Manthiram, A., 2014. Spinel-type lithium cobalt oxide as a bifunctional electrocatalyst for the oxygen evolution and oxygen reduction reactions. *Nature Communications*, 5(1), p.3949.
5. Colligan, N., Augustyn, V. and Manthiram, A., 2015. Evidence of Localized Lithium Removal in Layered and Lithiated Spinel $\text{Li}_{1-x}\text{CoO}_2$ ($0 \leq x \leq 0.9$) under Oxygen Evolution Reaction Conditions. *The Journal of Physical Chemistry C*, 119(5), pp.2335-2340.
6. Augustyn, V. and Manthiram, A., 2015. Effects of chemical versus electrochemical delithiation on the oxygen evolution reaction activity of nickel-rich layered LiMO_2 . *The Journal of Physical Chemistry Letters*, 6(19), pp.3787-3791.
7. Augustyn, V., Therese, S., Turner, T.C. and Manthiram, A., 2015. Nickel-rich layered $\text{LiNi}_{1-x}\text{M}_x\text{O}_2$ (M=Mn, Fe, and Co) electrocatalysts with high oxygen evolution reaction activity. *Journal of Materials Chemistry A*, 3(32), pp.16604-16612.
8. Gupta, A., Chemelewski, W.D., Buddie Mullins, C. and Goodenough, J.B., 2015. High-rate oxygen evolution reaction on Al-doped LiNiO_2 . *Advanced Materials*, 27(39), pp.6063-6067.
9. Balasubramanian, M., McBreen, J., Davidson, I.J., Whitfield, P.S. and Kargina, I., 2002. In situ X-ray absorption study of a layered manganese-chromium oxide-based cathode material. *Journal of the Electrochemical Society*, 149(2), p.A176.
10. Lyu, Y., Ben, L., Sun, Y., Tang, D., Xu, K., Gu, L., Xiao, R., Li, H., Chen, L. and Huang, X., 2015. Atomic insight into electrochemical inactivity of lithium chromate (LiCrO_2): Irreversible migration of chromium into lithium layers in surface regions. *Journal of Power Sources*, 273, pp.1218-1225.

11. Soni, V., Mondal, R., Singh, A.N., Singh, P. and Gupta, A., 2023. Dumbbell Defect Containing Chromium-Rich Lithium-Vacant Layered $\text{Li}_y\text{Cr}_{1-x}\text{Fe}_x\text{O}_2$ ($y \leq 1$, $0 \leq x \leq 0.2$): An Unexplored and Highly Efficient Electrocatalyst for the Oxygen Evolution Reaction. *ACS Applied Energy Materials*, 6(3), pp.1308-1320.
12. Kuznetsov, D.A., Han, B., Yu, Y., Rao, R.R., Hwang, J., Román-Leshkov, Y. and Shao-Horn, Y., 2018. Tuning redox transitions via inductive effect in metal oxides and complexes, and implications in oxygen electrocatalysis. *Joule*, 2(2), pp.225-244.
13. Bera, K., Karmakar, A., Karthick, K., Sankar, S.S., Kumaravel, S., Madhu, R. and Kundu, S., 2021. Enhancement of the OER kinetics of the less-explored $\alpha\text{-MnO}_2$ via nickel doping approaches in alkaline medium. *Inorganic Chemistry*, 60(24), pp.19429-19439.
14. Tang, L., Zhang, W., Lin, D., Ren, Y., Zheng, H., Luo, Q., Wei, L., Liu, H., Chen, J. and Tang, K., 2020. The hexagonal perovskite $\text{Ba}_{0.5}\text{Sr}_{0.5}\text{Co}_{0.8}\text{Fe}_{0.2}\text{O}_{3-\delta}$ as an efficient electrocatalyst for the oxygen evolution reaction. *Inorganic Chemistry Frontiers*, 7(22), pp.4488-4497.
15. Fernandes, J.D., Melo, D.M.D.A., Zinner, L.B., Salustiano, C.D.M., Silva, Z.R., Martinelli, A.E., Cerqueira, M., Junior, C.A., Longo, E. and Bernardi, M.I.B., 2002. Low-temperature synthesis of single-phase crystalline LaNiO_3 perovskite via Pechini method. *Materials Letters*, 53(1-2), pp.122-125.
16. Castro-García, S., Castro-Couceiro, A., Senaris-Rodríguez, M.A., Soulette, F. and Julien, C., 2003. Influence of aluminum doping on the properties of LiCoO_2 and $\text{LiNi}_{0.5}\text{Co}_{0.5}\text{O}_2$ oxides. *Solid State Ionics*, 156(1-2), pp.15-26.
17. Ohzuku, T., Ueda, A., Nagayama, M., Iwakoshi, Y. and Komori, H., 1993. Comparative study of LiCoO_2 , $\text{LiNi}_{1/2}\text{Co}_{1/2}\text{O}_2$ and LiNiO_2 for 4volt secondary lithium cells. *Electrochimica Acta*, 38(9), pp.1159-1167.
18. Guilmard, M., Rougier, A., Grüne, M., Croguennec, L. and Delmas, C., 2003. Effects of aluminum on the structural and electrochemical properties of LiNiO_2 . *Journal of Power Sources*, 115(2), pp.305-314.
19. Vegard, L., Dale, H., 1928. Tests on mixed crystals and alloys. *Z Krystallogr* **1928**, 67(1), pp.148-162.
20. Joint Committee on Powder Diffraction Standards, File no. 074-0919.
21. Jones, C.D.W., Rossen, E. and Dahn, J.R., 1994. Structure and electrochemistry of $\text{Li}_x\text{Cr}_y\text{Co}_{1-y}\text{O}_2$. *Solid State Ionics*, 68(1-2), pp.65-69.
22. Shannon, R. D., 1976. *Acta Crystallogr. Sect. A*, 32, p.751.

23. Li, K. and Xue, D., 2006. Estimation of electronegativity values of elements in different valence states. *The Journal of Physical Chemistry A*, 110(39), pp.11332-11337.
24. Feng, G.X., Li, L.F., Liu, J.Y., Liu, N., Li, H., Yang, X.Q., Huang, X.J., Chen, L.Q., Nam, K.W. and Yoon, W.S., 2009. Enhanced electrochemical lithium storage activity of LiCrO₂ by size effect. *Journal of Materials Chemistry*, 19(19), pp.2993-2998.
25. Kalyani, P. and Kalaiselvi, N., 2005. Various aspects of LiNiO₂ chemistry: A review. *Science and Technology of Advanced Materials*, 6(6), p.689.
26. Weckhuysen, B.M., Wachs, I.E. and Schoonheydt, R.A., 1996. Surface chemistry and spectroscopy of chromium in inorganic oxides. *Chemical Reviews*, 96(8), pp.3327-3350.
27. Taha, T.A. and El-Molla, M.M., 2020. Green simple preparation of LiNiO₂ nanopowder for lithium ion battery. *Journal of Materials Research and Technology*, 9(4), pp.7955-7960.
28. Anantharaj, S., Reddy, P.N. and Kundu, S., 2017. Core-oxidized amorphous cobalt phosphide nanostructures: an advanced and highly efficient oxygen evolution catalyst. *Inorganic Chemistry*, 56(3), pp.1742-1756.
29. Lu, Z. and Dahn, J.R., 2002. Structure and electrochemistry of layered Li[Cr_xLi_(1/3-x/3)Mn_(2/3-2x/3)]O₂. *Journal of the Electrochemical Society*, 149(11), p.A1454.
30. McCrory, C.C., Jung, S., Ferrer, I.M., Chatman, S.M., Peters, J.C. and Jaramillo, T.F., 2015. Benchmarking hydrogen evolving reaction and oxygen evolving reaction electrocatalysts for solar water splitting devices. *Journal of the American Chemical Society*, 137(13), pp.4347-4357.

5.5 References

1. Li, J., 2022. Oxygen evolution reaction in energy conversion and storage: design strategies under and beyond the energy scaling relationship. *Nano-Micro Letters*, 14(1), p.112.
2. Holladay, J.D., Hu, J., King, D.L. and Wang, Y., 2009. An overview of hydrogen production technologies. *Catalysis today*, 139(4), pp.244-260.
3. Feng, L.L., Yu, G., Wu, Y., Li, G.D., Li, H., Sun, Y., Asefa, T., Chen, W. and Zou, X., 2015. High-index faceted Ni₃S₂ nanosheet arrays as highly active and ultrastable electrocatalysts for water splitting. *Journal of the American Chemical Society*, 137(44), pp.14023-14026.
4. Jaiswal, S., Mondal, R., Kushwaha, V., Gupta, A. and Singh, P., 2023. Tuning of Redox Energy of Transition-Metal Ions through the Utilization of Interlayer Potentials in Layered Perovskites: Development of a Titanium-Based Superior HER Catalyst in an Acidic Medium. *ACS Applied Energy Materials*, 6(14), pp.7323-7334.
5. Fabbri, E., Haberer, A., Walz, K., Kötter, R. and Schmidt, T.J., 2014. Developments and perspectives of oxide-based catalysts for the oxygen evolution reaction. *Catalysis Science & Technology*, 4(11), pp.3800-3821.
6. Ouyang, T., Ye, Y.Q., Wu, C.Y., Xiao, K. and Liu, Z.Q., 2019. Heterostructures composed of N-doped carbon nanotubes encapsulating cobalt and β -Mo₂C nanoparticles as bifunctional electrodes for water splitting. *Angewandte Chemie International Edition*, 58(15), pp.4923-4928.
7. Koper, M.T., 2011. Thermodynamic theory of multi-electron transfer reactions: Implications for electrocatalysis. *Journal of Electroanalytical Chemistry*, 660(2), pp.254-260.
8. Lyons, M.E. and Brandon, M.P., 2010. A comparative study of the oxygen evolution reaction on oxidised nickel, cobalt and iron electrodes in base. *Journal of Electroanalytical Chemistry*, 641(1-2), pp.119-130.
9. McCrory, C.C., Jung, S., Peters, J.C. and Jaramillo, T.F., 2013. Benchmarking heterogeneous electrocatalysts for the oxygen evolution reaction. *Journal of the American Chemical Society*, 135(45), pp.16977-16987.
10. Hong, W.T., Risch, M., Stoerzinger, K.A., Grimaud, A., Suntivich, J. and Shao-Horn, Y., 2015. Toward the rational design of non-precious transition metal oxides for oxygen electrocatalysis. *Energy & Environmental Science*, 8(5), pp.1404-1427.

11. Kuznetsov, D.A., Han, B., Yu, Y., Rao, R.R., Hwang, J., Román-Leshkov, Y. and Shao-Horn, Y., 2018. Tuning redox transitions via inductive effect in metal oxides and complexes, and implications in oxygen electrocatalysis. *Joule*, 2(2), pp.225-244.
12. Lee, S.W., Carlton, C., Risch, M., Surendranath, Y., Chen, S., Furutsuki, S., Yamada, A., Nocera, D.G. and Shao-Horn, Y., 2012. The nature of lithium battery materials under oxygen evolution reaction conditions. *Journal of the American Chemical Society*, 134(41), pp.16959-16962.
13. Augustyn, V. and Manthiram, A., 2015. Characterization of Layered LiMO₂ Oxides for the Oxygen Evolution Reaction of Metal–Air Batteries (M= Mn, Co, Ni). *ChemPlusChem*, 80(2), pp.422-427.
14. Lu, Z., Wang, H., Kong, D., Yan, K., Hsu, P.C., Zheng, G., Yao, H., Liang, Z., Sun, X. and Cui, Y., 2014. Electrochemical tuning of layered lithium transition metal oxides for improvement of oxygen evolution reaction. *Nature communications*, 5(1), p.4345.
15. Maiyalagan, T., Jarvis, K.A., Therese, S., Ferreira, P.J. and Manthiram, A., 2014. Spinel-type lithium cobalt oxide as a bifunctional electrocatalyst for the oxygen evolution and oxygen reduction reactions. *Nature communications*, 5(1), p.3949.
16. Augustyn, V., Therese, S., Turner, T.C. and Manthiram, A., 2015. Nickel-rich layered LiNi_{1-x}M_xO₂ (M= Mn, Fe, and Co) electrocatalysts with high oxygen evolution reaction activity. *Journal of Materials Chemistry A*, 3(32), pp.16604-16612.
17. Gupta, A., Chemelewski, W.D., Buddie Mullins, C. and Goodenough, J.B., 2015. High-rate oxygen evolution reaction on Al-doped LiNiO₂. *Advanced Materials*, 27(39), pp.6063-6067.
18. Zhu, K., Wu, T., Zhu, Y., Li, X., Li, M., Lu, R., Wang, J., Zhu, X. and Yang, W., 2017. Layered Fe-substituted LiNiO₂ electrocatalysts for high-efficiency oxygen evolution reaction. *ACS Energy Letters*, 2(7), pp.1654-1660.
19. Balasubramanian, M., McBreen, J., Davidson, I.J., Whitfield, P.S. and Kargina, I., 2002. In situ X-ray absorption study of a layered manganese-chromium oxide-based cathode material. *Journal of the electrochemical Society*, 149(2), p.A176.
20. Lu, Z. and Dahn, J.R., 2002. Structure and electrochemistry of layered Li [Cr_xLi_(1/3-x/3)Mn_(2/3-2x/3)]O₂. *Journal of the Electrochemical Society*, 149(11), p.A1454.
21. Soni, V., Mondal, R., Singh, A.N., Singh, P. and Gupta, A., 2023. Dumbbell Defect Containing Chromium-Rich Lithium-Vacant Layered Li_yCr_{1-x}Fe_xO₂ (y≤ 1, 0≤ x≤ 0.2): An Unexplored and Highly Efficient Electrocatalyst for the Oxygen Evolution Reaction. *ACS Applied Energy Materials*, 6(3), pp.1308-1320.

22. Dixit, M., Markovsky, B., Aurbach, D. and Major, D.T., 2017. Unraveling the effects of Al doping on the electrochemical properties of $\text{LiNi}_{0.5}\text{Co}_{0.2}\text{Mn}_{0.3}\text{O}_2$ using first principles. *Journal of The Electrochemical Society*, 164(1), p.A6359.
23. Bera, K., Karmakar, A., Karthick, K., Sankar, S.S., Kumaravel, S., Madhu, R. and Kundu, S., 2021. Enhancement of the OER kinetics of the less-explored $\alpha\text{-MnO}_2$ via nickel doping approaches in alkaline medium. *Inorganic Chemistry*, 60(24), pp.19429-19439.
24. Tang, L., Zhang, W., Lin, D., Ren, Y., Zheng, H., Luo, Q., Wei, L., Liu, H., Chen, J. and Tang, K., 2020. The hexagonal perovskite $\text{Ba}_{0.5}\text{Sr}_{0.5}\text{Co}_{0.8}\text{Fe}_{0.2}\text{O}_{3-\delta}$ as an efficient electrocatalyst for the oxygen evolution reaction. *Inorganic Chemistry Frontiers*, 7(22), pp.4488-4497.
25. Fernandes, J.D., Melo, D.M.D.A., Zinner, L.B., Salustiano, C.D.M., Silva, Z.R., Martinelli, A.E., Cerqueira, M., Junior, C.A., Longo, E. and Bernardi, M.I.B., 2002. Low-temperature synthesis of single-phase crystalline LaNiO_3 perovskite via Pechini method. *Materials Letters*, 53(1-2), pp.122-125.
26. García, S.C., Couceiro, A.C., Rodriguez, M.A.S., Soulette, F. and Julien, C., 2003. Influence of aluminum doping on the properties of LiCoO_2 and $\text{LiNi}_{0.5}\text{Co}_{0.5}\text{O}_2$ oxides. *Solid State Ionics*, 156, p.15.
27. Ohzuku, T., Ueda, A., Nagayama, M., Iwakoshi, Y. and Komori, H., 1993. Comparative study of LiCoO_2 , $\text{LiNi}_{12}\text{Co}_{12}\text{O}_2$ and LiNiO_2 for 4 volt secondary lithium cells. *Electrochimica Acta*, 38(9), pp.1159-1167.
28. Guilmard, M., Rougier, A., Grüne, M., Croguennec, L. and Delmas, C., 2003. Effects of aluminum on the structural and electrochemical properties of LiNiO_2 . *Journal of Power Sources*, 115(2), pp.305-314.
29. File, P.D., 1967. Joint committee on powder diffraction standards. *ASTM, Philadelphia, Pa*, pp.9-185.
30. Vegard, L.; Dale, H.; *Krystallogr. Z.* 1928, 67, 148.
31. West, A. R. 1984, *Solid State Chemistry and its Applications*, Wiley, New York p. 367.
32. Shannon, R. D. 1976, *Acta Crystallogr. Sect. A*, 32, 751.
33. Lyu, Y., Ben, L., Sun, Y., Tang, D., Xu, K., Gu, L., Xiao, R., Li, H., Chen, L. and Huang, X., 2015. Atomic insight into electrochemical inactivity of lithium chromate (LiCrO_2): Irreversible migration of chromium into lithium layers in surface regions. *Journal of Power Sources*, 273, pp.1218-1225.

34. Feng, G.X., Li, L.F., Liu, J.Y., Liu, N., Li, H., Yang, X.Q., Huang, X.J., Chen, L.Q., Nam, K.W. and Yoon, W.S., 2009. Enhanced electrochemical lithium storage activity of LiCrO₂ by size effect. *Journal of Materials Chemistry*, 19(19), pp.2993-2998.
35. Singh, P., Hegde, M.S. and Gopalakrishnan, J., 2008. Ce_{2/3}Cr_{1/3}O_{2+y}: a new oxygen storage material based on the fluorite structure. *Chemistry of Materials*, 20(23), pp.7268-7273.
36. Xiao, B., Tang, Q., Dai, X., Wu, F., Chen, H., Li, J., Mai, Y. and Gu, Y., 2022. Enhanced Interfacial Kinetics and High Rate Performance of LiCoO₂ Thin-Film Electrodes by Al Doping and In Situ Al₂O₃ Coating. *ACS omega*, 7(35), pp.31597-31606.
37. Andreeva, R.A., Stoyanova, E.A., Tsanev, A.S. and Stoychev, D.S., 2016. Corrosion behavior of anodically formed oxide films on aluminum, sealed in cerium-ions containing solutions. *Bulg. Chem. Commun*, 48, pp.96-102.
38. Limcharoen, A., Pakpum, C. and Limsuwan, P., 2012. An X-ray photoelectron spectroscopy investigation of redeposition from fluorine-based plasma etch on magnetic recording slider head substrate. *Procedia Engineering*, 32, pp.1043-1049.
39. Cherkashinin, G., Nikolowski, K., Ehrenberg, H., Jacke, S., Dimesso, L. and Jaegermann, W., 2012. The stability of the SEI layer, surface composition and the oxidation state of transition metals at the electrolyte–cathode interface impacted by the electrochemical cycling: X-ray photoelectron spectroscopy investigation. *Physical Chemistry Chemical Physics*, 14(35), pp.12321-12331.
40. Gupta, A., Kumar, A., Waghmare, U.V. and Hegde, M.S., 2009. Origin of activation of lattice oxygen and synergistic interaction in bimetal-ionic Ce_{0.89}Fe_{0.1}Pd_{0.01}O_{2-δ} catalyst. *Chemistry of materials*, 21(20), pp.4880-4891.
41. Scofield, J.H., 1976. Hartree-Slater subshell photoionization cross-sections at 1254 and 1487 eV. *Journal of Electron Spectroscopy and Related Phenomena*, 8(2), pp.129-137.
42. Penn, D.R., 1976. Quantitative chemical analysis by ESCA. *Journal of Electron Spectroscopy and Related Phenomena*, 9(1), pp.29-40.
43. Petrie, J.R., Cooper, V.R., Freeland, J.W., Meyer, T.L., Zhang, Z., Lutterman, D.A. and Lee, H.N., 2016. Enhanced bifunctional oxygen catalysis in strained LaNiO₃ perovskites. *Journal of the American Chemical Society*, 138(8), pp.2488-2491.
44. Chen, J., Wu, J., Liu, Y., Hu, X. and Geng, D., 2018. Assemblage of Perovskite LaNiO₃ Connected with In Situ Grown Nitrogen-Doped Carbon Nanotubes as High-Performance Electrocatalyst for Oxygen Evolution Reaction. *physica status solidi (a)*, 215(21), p.1800380.

45. Anantharaj, S., Reddy, P.N. and Kundu, S., 2017. Core-oxidized amorphous cobalt phosphide nanostructures: an advanced and highly efficient oxygen evolution catalyst. *Inorganic Chemistry*, 56(3), pp.1742-1756.

6.3 References

1. Bera, K., Karmakar, A., Karthick, K., Sankar, S.S., Kumaravel, S., Madhu, R. and Kundu, S., 2021. Enhancement of the OER kinetics of the less-explored α - MnO_2 via nickel doping approaches in alkaline medium. *Inorganic Chemistry*, 60(24), pp.19429-19439.
2. Tang, L., Zhang, W., Lin, D., Ren, Y., Zheng, H., Luo, Q., Wei, L., Liu, H., Chen, J. and Tang, K., 2020. The hexagonal perovskite $\text{Ba}_{0.5}\text{Sr}_{0.5}\text{Co}_{0.8}\text{Fe}_{0.2}\text{O}_{3-\delta}$ as an efficient electrocatalyst for the oxygen evolution reaction. *Inorganic Chemistry Frontiers*, 7(22), pp.4488-4497.
3. Petrie, J.R., Cooper, V.R., Freeland, J.W., Meyer, T.L., Zhang, Z., Lutterman, D.A. and Lee, H.N., 2016. Enhanced bifunctional oxygen catalysis in strained LaNiO_3 perovskites. *Journal of the American Chemical Society*, 138(8), pp.2488-2491.
4. Chen, J., Wu, J., Liu, Y., Hu, X. and Geng, D., 2018. Assemblage of Perovskite LaNiO_3 Connected with In Situ Grown Nitrogen-Doped Carbon Nanotubes as High-Performance Electrocatalyst for Oxygen Evolution Reaction. *physica status solidi (a)*, 215(21), p.1800380.
5. Anantharaj, S., Reddy, P.N. and Kundu, S., 2017. Core-oxidized amorphous cobalt phosphide nanostructures: an advanced and highly efficient oxygen evolution catalyst. *Inorganic Chemistry*, 56(3), pp.1742-1756.

List of Publications

1. **Vaishali Soni**, Abhay Narayan Singh, Preetam Singh and Asha Gupta, *Photocatalytic dye-degradation activity of nano-crystalline $Ti_{1-x}M_xO_{2-\delta}$ ($M = Ag, Pd, Fe, Ni$ and $x = 0, 0.01$) for water pollution abatement.* **RSC Advances**. 2022, 12, 18794-18805 (<https://doi.org/10.1039/D2RA02847F>).
2. **Vaishali Soni**, Rakesh Mondal, Abhay Narayan Singh, Preetam Singh, and Asha Gupta, *Dumbbell Defect Containing Chromium-Rich Lithium-Vacant Layered $Li_yCr_{1-x}Fe_xO_2$ ($y \leq 1, 0 \leq x \leq 0.2$): An Unexplored and Highly Efficient Electrocatalyst for the Oxygen Evolution Reaction.* **ACS Applied Energy Materials**. 2023, 6, 1308–1320 (<https://doi.org/10.1021/acsaem.2c03056>).
3. **Vaishali Soni**, Shraddha Jaiswal, Preetam Singh, & Asha Gupta, *Aluminium doped Lithium-vacant Layered $Li_{1-x}Cr_{1-x}Al_xO_2$: A Potentially Active Electrocatalyst for Oxygen Evolution Reaction.* **ACS Applied Energy Materials**. 2024, 7, 8, 3175–3186 (<https://doi.org/10.1021/acsaem.3c03160>).
4. **Vaishali Soni**, Shraddha Jaiswal, Krishna Gopal Nigam, Preetam Singh, & Asha Gupta, *Nickel doped Lithium-vacant Layered $Li_yCr_{1-x}Ni_xO_2$: A Potentially Active Electrocatalyst for Oxygen Evolution Reaction.* **Journal of Material Chemistry A**, 2024 (<https://doi.org/10.1039/d4ta02717e>).
5. Shraddha Jaiswal, **Vaishali Soni**, Preetam Singh, & Asha Gupta, *Role of cation deficiency and inductive effect in Ti-doped NiO for developing superior electrocatalysts for Oxygen Evolution Reaction.* **ACS Applied Energy Materials** (Revision submitted), Manuscript ID: ae-2024-01774z.

International/National Conferences Attended

1. The international conference on “**ChemCatCon 2.0: Mechanistic investigation on heterogeneous processes**” IIT Gandhinagar, India, 14-16 May 2022.
2. The International conference on **Beyond Fossil Fuels: The Future of Alternative Energy Technologies (B: FAT-2020)**, IIT (BHU) Varanasi, India, 23-25, July 2022.