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## Appendix- A: List of Publications

### Published Papers

1. Alumina dissolution process to fabricate bimodal pore architecture alumina with superior green and sintered properties. **Vaibhav Pandey**, S. K. Panda, V. K. Singh. *Journal of the American Ceramic Society* (<https://doi.org/10.1111/jace.19280>) (2023).
2. Synthesis, Morphological and Thermomechanical Characterization of Light Weight Silica Foam via Reaction Generated Thermo-Foaming Process. **Vaibhav Pandey**, Mayank Kumar Yadav, Ashutosh Gupta, K. Mohanta, S. K. Panda, V. K. Singh. *Journal of European Ceramic Society*.<https://doi.org/10.1016/j.jeurceramsoc.2022.07.034> (2022).
3. An economic and sustainable approach to transform aluminosilicate-rich solid waste to functionally graded composite foam for high-temperature applications. **Vaibhav Pandey**, Mayank Kumar Yadav, S.K. Panda, V.K. Singh. *Chemosphere* 338 (2023) 139588. <https://doi.org/10.1016/j.chemosphere.2023.139588>

### Submitted Papers

1. Preparation and Characterization of high-strength insulating porous bricks by reusing coal mine overburden waste, red mud and rice husk. **Vaibhav Pandey**, S. K. Panda, V. K. Singh. *Journal of Cleaner Production*.(Under Revision).

### Papers in Writing

1. An economic and sustainable methodology to transform alumina rich solid wastes into high strength light weight porous composites. **Vaibhav Pandey**, S. K. Panda, V. K. Singh. *Resources, Conservation and Recycling*.
2. Fabrication of Mullite bonded functionally graded SiC foam with gradient microstructure via one step thermo-foaming process. **Vaibhav Pandey**, Asteek Raj, Ashutosh Gupta, Gaurav Shoor, S. K. Panda, V. K. Singh, Manas Ranjan Majhi. *Ceramics International*.

## Appendix: B: Details of Characterization Techniques

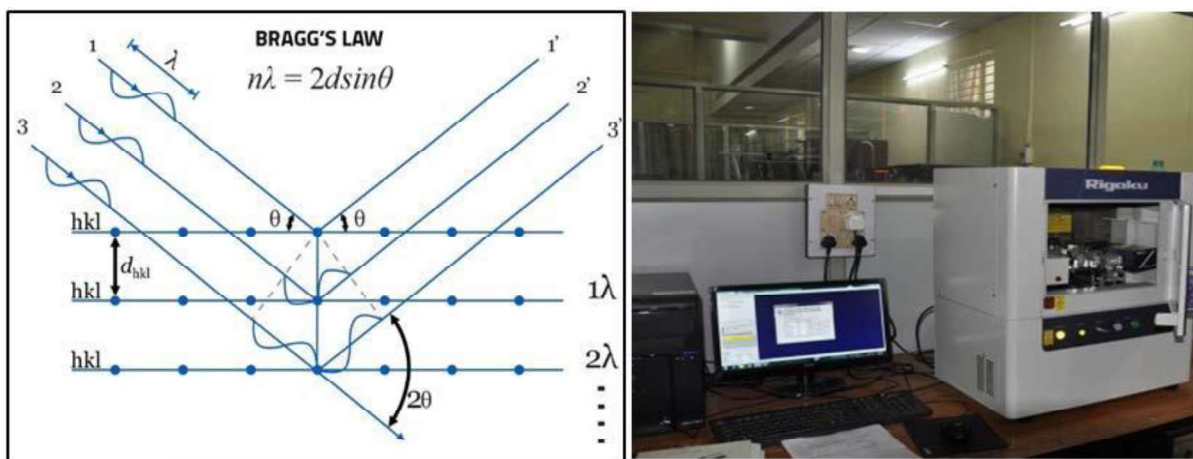
### 1-X-ray Diffraction (XRD)

X-ray diffraction (XRD) is a sophisticated method of examining the crystal structure and interplanar spacing of materials. It relies on the constructive interference of monochromatic X-rays and a crystalline sample. This is achieved by generating a beam of X-rays from a cathode ray tube, filtering it to obtain monochromatic radiation, collimating it to concentrate the beam, and directing it towards the sample. When the conditions align with Bragg's Law (equation 1), the incident rays interacting with the sample result in constructive interference and the production of a diffracted ray:

$$n\lambda = 2d \sin \theta \dots\dots\dots(1)$$

Where,  $n$  is an integer value,  $\lambda$  represents the wavelength of X-rays,  $d$  is interplanar spacing generating the diffraction, and  $\theta$  is the diffraction angle.

Bragg's law establishes a connection between the wavelength of electromagnetic radiation, the diffraction angle, and the lattice spacing in a crystal sample. The resulting diffracted X-rays are subsequently detected, processed, and quantified. To ensure coverage of all potential lattice diffraction directions resulting from the random orientation of the powdered material, the sample is scanned across a range of  $2\theta$  angles.



**Fig. Appendix(B).0.1 Schematic representation of principle of X-Ray diffraction spectroscopy (XRD) and Photograph of XRD machine**

X-ray diffractometers are comprised of three essential components: an X-ray tube, a sample holder, and an X-ray detector. In a cathode ray tube, X-rays are generated through the heating of a filament to produce electrons, accelerating these electrons towards a target by applying

voltage, and bombarding the target material with electrons. When the energy of the electrons is sufficient to displace inner shell electrons in the target material, characteristic X-ray spectra are emitted. These spectra consist of various components, with the most common ones being  $K\alpha$  and  $K\beta$ .  $K\alpha$  further includes  $K\alpha_1$  and  $K\alpha_2$ , where  $K\alpha_1$  has a slightly shorter wavelength and twice the intensity of  $K\alpha_2$ . The specific wavelengths depend on the target material, such as Cu, Fe, Mo, or Cr. To achieve monochromatic X-rays required for diffraction, filtration is necessary using foils or crystal monochrometers. Since  $K\alpha_1$  and  $K\alpha_2$  have closely related wavelengths, a weighted average of the two is typically used. Copper (Cu) is commonly employed as the target material for single-crystal diffraction, with  $CuK\alpha$  radiation having a wavelength of  $1.5418\text{\AA}$ . These X-rays are collimated and directed towards the sample. By rotating the sample and detector, the intensity of the reflected X-rays is recorded. When the incident X-ray geometry aligns with the Bragg Equation, constructive interference takes place, resulting in a peak in intensity. A detector captures and processes the X-ray signal, converting it into a count rate, which can be outputted to devices such as printers or computer monitors. Fig. Appendix(B).0.2 provides a schematic representation and photograph of an XRD setup.

## **2-Scanning Electron Microscopy (SEM)**

The scanning electron microscope (SEM) is utilized to examine the surface morphology of a sample at varying magnifications, resolutions, and depths of focus, offering higher resolution compared to an optical microscope. In this technique, a well-focused monoenergetic electron beam is directed onto the solid surface under investigation. The interaction between the beam and the surface leads to diverse scattering processes. Secondary electrons (SE) and backscattered electrons (BSE) are the primary signals used in SEM to capture surface morphology. These SE or BSE signals are collected and converted into current signals, which are then amplified to control the brightness of the cathode ray tube (CRT) or monitor screen. SEM operates under vacuum conditions, necessitating special sample preparation to ensure no moisture is present in the sample or chamber, as it would impede the vacuum. Conductive materials such as metals generally do not require additional preparation. However, ceramic samples are typically coated with a thin layer of gold to enhance their conductivity. In the current study, SEM analysis was conducted to examine the surface morphology of all sintered samples. The SEM instrument used was the INSPECT 50 FEI model, specifically the EVO-scanning electron microscope MA15/18 by CARL ZEISS MICROSCOPY LTD. This particular SEM is equipped with energy dispersive spectroscopy (EDX), allowing for elemental analysis

of the prepared samples through EDX and mapping techniques at specific magnifications (Fig. Appendix(B).2).



**Fig. Appendix(B).2 Scanning Electron machine setup**

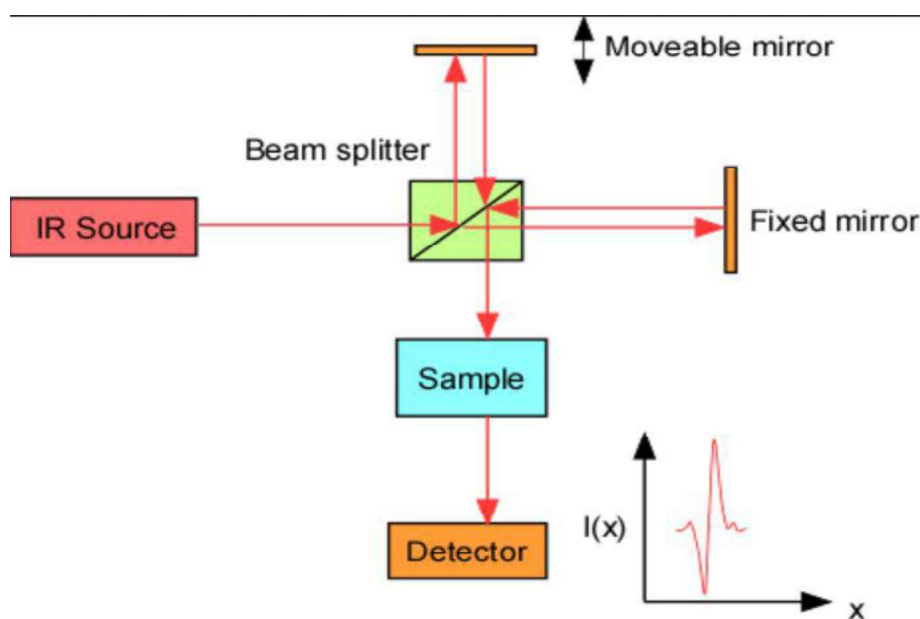
### **3-Fourier Transform Infrared Spectroscopy (FTIR)**

FTIR spectroscopy is a valuable technique used to identify the presence of various functional groups in both organic and inorganic molecules or compounds. It serves as a powerful tool for determining the types of chemical bonds within a molecule, generating an infrared absorption spectrum that acts as a unique "fingerprint" for the molecule. The term Fourier Transform Infrared Spectroscopy (FTIR) describes the advancement in data collection and conversion from an interference pattern to a spectrum. The method is based on the principles of the Michelson interferometer, which consists of a beam splitter, a fixed mirror, and a movable mirror that precisely moves back and forth.

During FTIR spectroscopy, radiation from a source strikes the beam splitter and separates into two beams. One beam transmits through the beam splitter and reaches the static mirror, while the other beam reflects off the beam splitter towards the moving mirror. The fixed and moving mirrors reflect the radiation back to the beam splitter. Half of the reflected radiation is transmitted, and the other half is reflected at the beam splitter. This results in one beam passing

through the sample, detected by a detector, and displayed as spectra on a computer, while the second beam returns to the radiation source.

In our study, the FTIR spectra of the samples were obtained using the "PerkinElmer Spectrum 100" instrument. The radiation sources passed through a KBr window, and the transmitted data were collected by a LiTaO<sub>3</sub> detector. The spectra were displayed in transmitted mode. Sample pellets were prepared by mixing the sample with KBr in a ratio of 1:100. The sample pellets were then scanned in the range of 400-4000 cm<sup>-1</sup> with a spectral resolution of 4.0 cm<sup>-1</sup> and a scan speed of 0.2 cm/sec. Fig Appendix(B).3 provides a depiction of the working principle and a photograph of the FTIR spectrophotometer.



**Fig. Appendix(B).0.3 Schematic representation of working of Fourier transforms infrared (FTIR) spectrophotometer and its photograph**

#### **4-Transmission Electron Microscopy (TEM)**

The Transmission Electron Microscope (TEM) is a highly powerful microscope that utilizes an electron beam to focus on a specimen, producing a greatly magnified and detailed image. The magnification capability of TEM exceeds that of a light microscope by over 2 million times, enabling easy characterization of the specimen's morphological features, compositions, and crystallization information. In this technique, electrons are generated by a heated tungsten filament in the electron gun and focused onto the specimen through condenser lenses (as shown in Fig. Appendix(B).0.4). When the electrons reach the specimen, they are scattered and focused by magnetic lenses, resulting in a large, clear image. If the electrons pass through a fluorescent screen, a polychromatic image is formed. The density of the specimen determines

the extent of electron scattering, with denser specimens producing darker images due to fewer electrons reaching the screen for visualization. Conversely, thinner and more transparent specimens appear brighter.

TEM operates on similar principles as a light microscope but employs electrons instead of light. Since electrons possess both wave and particle properties, their de Broglie wavelength is significantly smaller than that of light, granting them higher resolution capabilities. This allows users of the instrument to examine fine details, down to the scale of a single column of atoms, which is tens of thousands of times smaller than the smallest resolvable object in a light microscope. TEM serves as a crucial analytical method in various scientific fields, including physical, chemical, and biological sciences. At lower magnifications, TEM images display contrast due to electron absorption in the material, as well as the thickness and composition of the specimen. At higher magnifications, complex wave interactions modulate the image intensity, necessitating expert analysis of the observed images. Alternate modes of TEM usage enable the observation of modulations in chemical identity, crystal orientation, electronic structure, sample-induced electron phase shift, in addition to regular absorption-based imaging.

In the present study, TEM analysis of the sample was performed using the TECNAI 20 G2 Electron Microscope operated at an accelerating voltage of 200 kV (as depicted in Figure 4.9). The samples were prepared by simply mounting a dilute solution of the sample on a carbon-coated TEM grid and allowing it to dry under a table lamp for 5 hours, followed by overnight vacuum drying.



**Fig. Appendix(B).0.4 Schematic representation of principle of Transmission electron microscopy (TEM) and Photograph of TEM**

## **5-Micro computed tomography (micro-CT)**

Micro computed tomography (micro-CT) is a non-destructive imaging technique that allows for the three-dimensional visualization and analysis of the internal structure of objects at a microscopic scale. It operates based on the principles of X-ray imaging and computed tomography. In micro-CT, a sample is placed on a rotating stage and exposed to a series of X-ray beams from different angles. As the X-rays pass through the sample, they interact with the material and are attenuated to varying degrees depending on the density and composition of the internal structures. The attenuated X-rays are detected by a high-resolution detector, generating a series of two-dimensional projections. These projections are then processed using mathematical algorithms to reconstruct a three-dimensional image of the internal structure of the sample. By capturing a large number of projections from different angles, micro-CT is able to provide detailed information about the spatial distribution of various materials and structures within the sample, enabling quantitative analysis, virtual slicing, and non-destructive evaluation of the sample's internal features. Micro-CT finds applications in a wide range of fields, including materials science, biomedical research, geology, archaeology, and more.

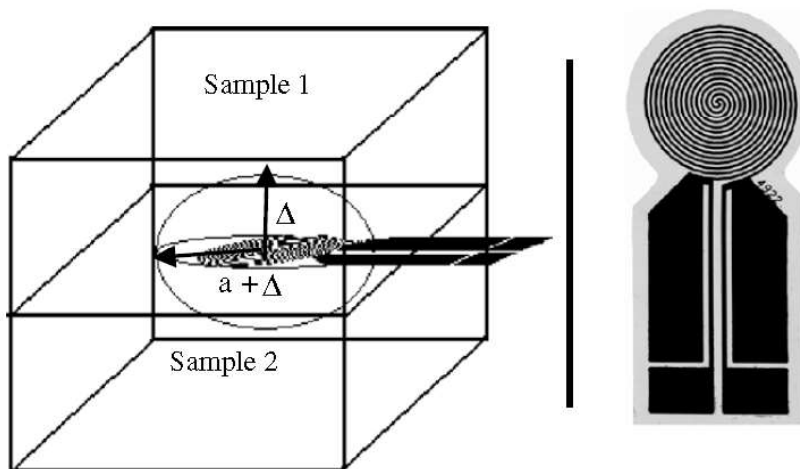
In our case, micro-computed tomography ( $\mu$ CT) analysis of the polished samples was conducted using the SkyScan1076 CT scanner from Aartselaar, Belgium. The specimen used had a cuboidal shape with dimensions of 15 mm in length, 10 mm in breadth, and 10 mm in height. The specimen was vertically mounted on the stage of the CT scanner. An X-ray source with a voltage of 86 kV and a current of 110 mA was employed for the analysis. During the scanning process, images were acquired at a rotation step of  $0.1^\circ$ . Subsequently, the obtained images were reconstructed using SkyScan's volumetric NRecon reconstruction software.

## **6-Thermal Conductivity**

The basic principle of TPS method involves using a plane element that functions as both a temperature sensor and a heat source. This element is constructed using a thin nickel foil (10  $\mu\text{m}$ ) in a spiral shape, embedded in an insulating layer typically made of Kapton (70  $\mu\text{m}$  thick). The temperature-sensitive (TPS) element is positioned between two samples, with the sensor faces in contact with the surfaces of the samples, as shown in Figure 1. It is important to note that two samples with similar characteristics are required for this technique.

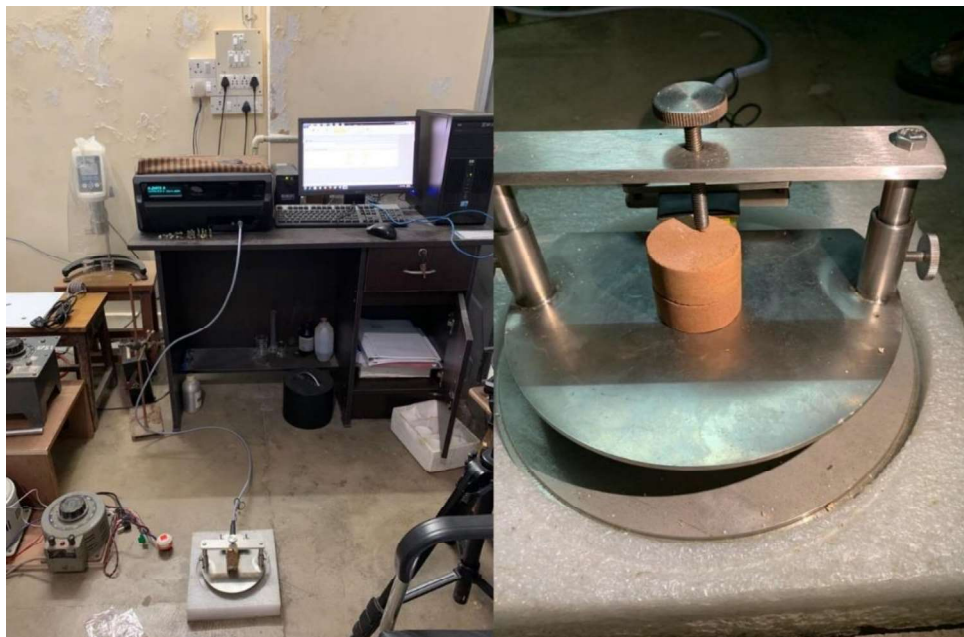
This method offers several advantages compared to standard methods. It enables fast and straightforward experiments, providing access to a wide range of thermal conductivities ranging from 0.02 to 400 W/m K. Sample preparation requires minimal effort, and the method

offers flexibility in sample size. Additionally, by changing the diameter of the sensor, it is possible to perform localized or bulk measurements without significant adjustments.



**Fig. Appendix(B)0.5 Setup for TPS Measurements and Sensor Configuration**

In our work, TPS-500, Goteborg, Sweden, made Hot Disk instrument has been used to measure the thermal conductivity of the porous silica. Measurements are taken by sandwiching the hot disk sensors between two identical silica foam pieces. The sensor system has an electrically conducting pattern in the shape of a double spiral laminated between two thin sheets of insulating material (Kapton).



**Fig. Appendix(B)0.6 Thermal Conductivity testing setup**